

ATHENS2017



Session XVII

Assessing the 3T method as a replacement to R1 formula for measuring the efficiency of waste-to-energy plants

Stergios Vakalis, Konstantinos Moustakas and Maria Loizidou

**5th International Conference on
Sustainable Solid Waste Management**

21–24 June 2017 www.athens2017.uest.gr

What is “waste-to-energy”

- It is the term that addresses the energy production by means of thermal treatment of waste.
- It primarily refers to combustion of municipal solid waste.
 - Commercial and Industrial waste are also considered
 - Thermal processes like gasification and pyrolysis are becoming more popular.
- The term should not be confused with “energy from waste”, which is a more general term that includes a broader range of technological possibilities.

Waste-to-energy data

- In 2014 more than 88 million tons of waste were thermally treated in waste-to-energy plants (Ella Stengler - C.E.W.E.P., 2016)
- For the production of:
 - 38 billion KWh electricity
 - 88 billion KWh heat
- After thermal treatment there are solid residues of approximately 30 % by weight and 10 % by volume that are primarily disposed to landfills.

The dual nature of waste-to-energy

- Historically, all the “Waste Framework Directives” that have been issued by the European Commission, separate the waste management strategies into Recovery Operations and Disposal Operations.
- Waste-to-energy technologies have the inherent problem that they do not belong entirely on the one category or the other.
 - Directive 2008/98/EU of the European parliament and of the council of 19 November 2008 on waste
 - waste is used principally as a fuel for energy generation and thus they belong to category 1 of the Recovery Operations (ANNEX I), i.e. R 1.
 - the residues of the treatment are landfilled on land and thus they belong to category 10 of the Disposal Operations (ANNEX II), i.e. D 10.

Issues that derive from the “duality”

- The issue of “duality” has been of high importance because each waste-to-energy facility could be considered an energy production or a disposal facility according to the category that is assigned.
- This influences the level of the gates fees but also the overall taxation of the waste-to-energy facilities.

Introduction of the R1 formula

- In order to address this issue European Commission integrated the R1 formula (that was developed by Dieter Reimann) in the second revision of the Waste Framework Directive of 2008.

- $$R1 = \frac{(E_p - (E_f + E_i))}{0.97 * (E_w + E_f)}$$


- $$R1 = \frac{(\text{Energy produced} - \text{Energy from fuels} - \text{Other energy imported})}{0.97 * (\text{Energy of waste input} + \text{Energy from fuels})}$$

Utilization of the R1 formula

- The parameters for each waste-to-energy facility are inserted to the R1 formula and the ones who have values over 0.65 (or 0.6 for older plants) achieve the R1 status.
- It should be denoted that the R1 formula played an important role in assisting the waste-to-energy plants to receive a legal status, especially during a period that the specifics of the waste-to-energy technologies were not fully understood by the lawmakers.
- Therefore, the significance of the R1 formula for the waste-to-energy sector should be stated.
- It must be pointed out that the R1 formula does not claim to be a pure energy efficiency formula but a “utilization efficiency” formula.

In 1 ton of bottom ash:

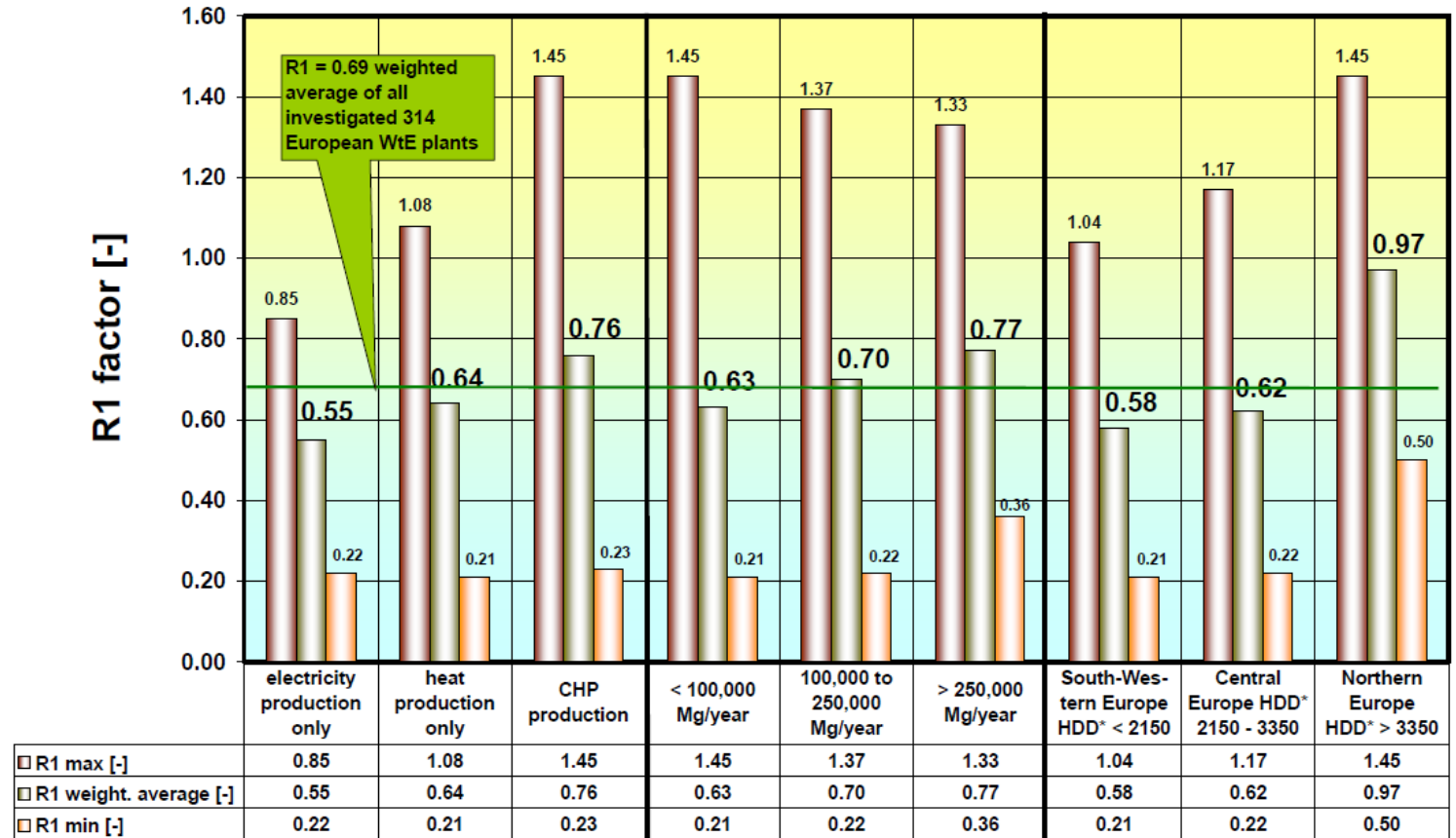
M. Castaldi & N. Themelis (2010). The Case for Increasing the Global Capacity for Waste to Energy (WTE). Waste and Biomass Valor 1:91–105.

- The quality of secondary aluminium is affected by its oxidation level (Astrup & Grosso, 2016)
 - Waste-to-energy plants are not only energy production units but also metal recovery facilities.
- 

Weighted significance of CHP

$$R1 = \frac{(E_p - (E_f + E_i))}{0.97 * (E_w + E_f)}$$

2.6 for electricity
1.1 for heat
1 for other fuels



* HDD = Heating Degree Day

Reimann 2012

Is there a possible alternative?

Which parameters do we need?

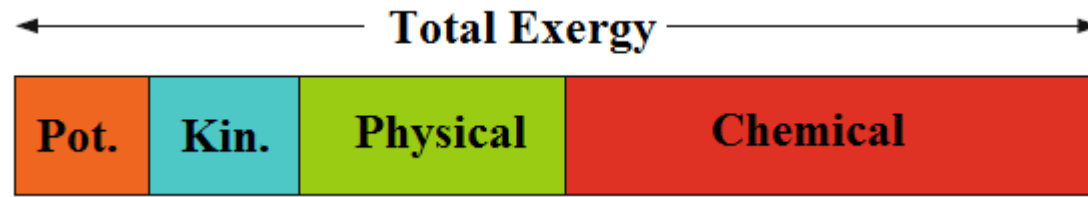
Combined Heat and Power efficiency

- CHP efficiency is the first basic parameter that we should take into consideration
- The case of heat vs electricity
 - Physical exergy instead of R1 factors (2.6 & 1.1)
- Chemical exergy of gaseous fuels, bio-oil etc.
- Chemical exergy of metals

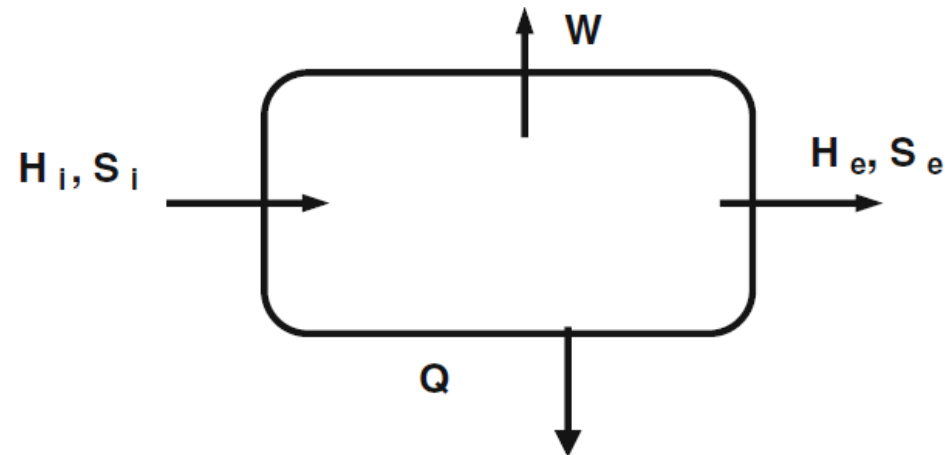
The concept of exergy

Measure of the maximum amount of work that can theoretically be obtained by bringing a resource into equilibrium with its surroundings through a reversible process.

$$[B = h - h_o - T_o (s - s_o)]$$



- A linear combination of the entropy and energy balances
- Reflects the 'quality' of energy



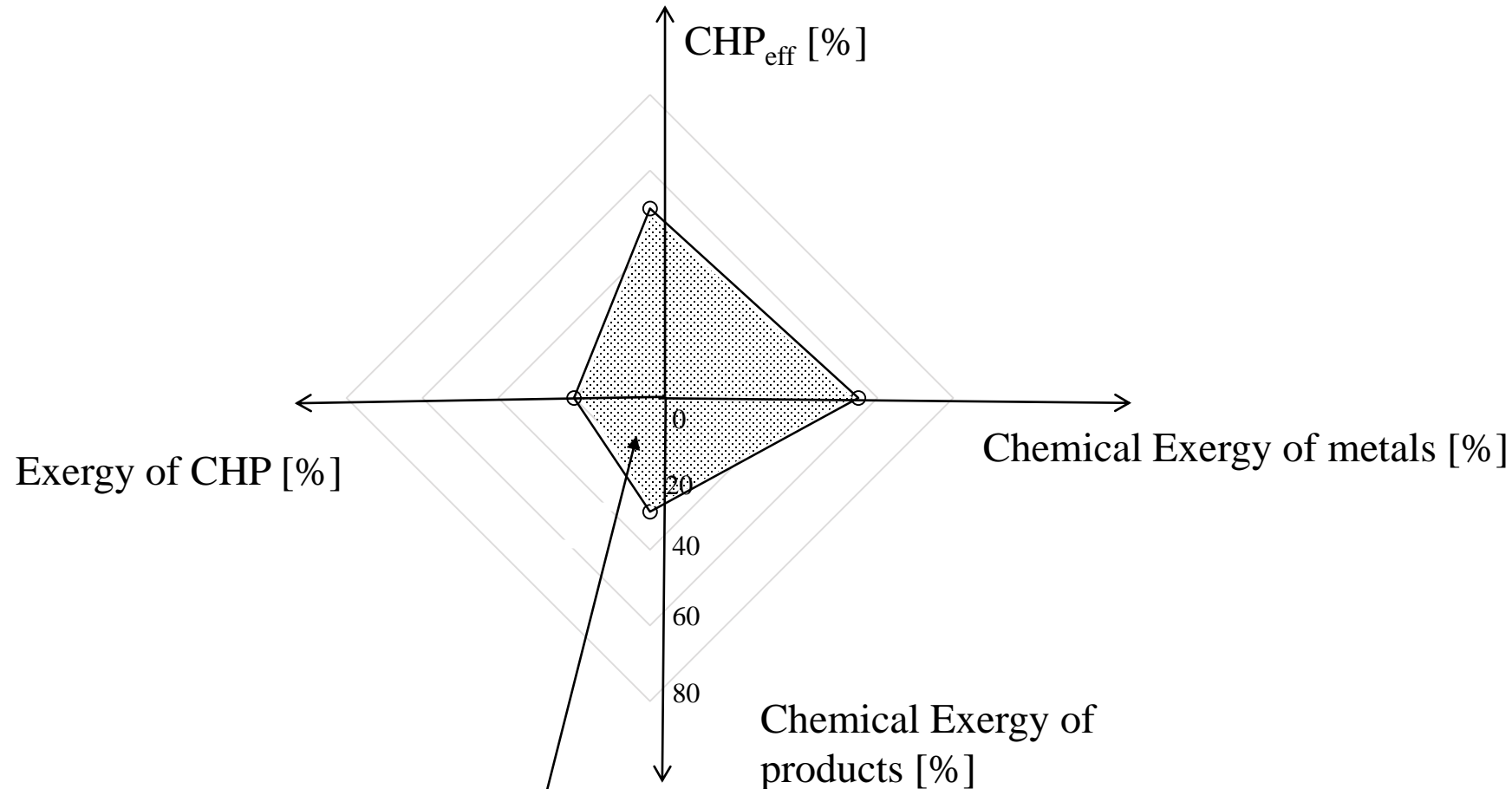
Exergy of different streams

Physical Exergy	Chemical Exergy																							
CHP	Products (e.g. Gaseous fuels)	Residue metals																						
<p>- Conversion of electricity into work on a 1:1 basis</p> <p>Exergy of heat depends on temperature and pressure</p> <p>e.g. Steam with 100 MJ (P: 1 atm, T: 450 K) → 33.3 MJ (P: 1 atm, T: 550 K) → 45.5 MJ (P: 1 atm, T: 650 K) → 63.9 MJ</p>	<table><tr><th><u>Sustance</u></th><th><u>Chemical exergy</u></th></tr><tr><td>Carbon Monoxide</td><td>275 kJ/<u>mol</u></td></tr><tr><td>Hydrogen</td><td>236 kJ/<u>mol</u></td></tr><tr><td>Methane</td><td>831 kJ/<u>mol</u></td></tr><tr><td>Carbon (graphite)</td><td>410 kJ/<u>mol</u></td></tr><tr><td>Carbon Dioxide</td><td>20 kJ/<u>mol</u></td></tr></table>	<u>Sustance</u>	<u>Chemical exergy</u>	Carbon Monoxide	275 kJ/ <u>mol</u>	Hydrogen	236 kJ/ <u>mol</u>	Methane	831 kJ/ <u>mol</u>	Carbon (graphite)	410 kJ/ <u>mol</u>	Carbon Dioxide	20 kJ/ <u>mol</u>	<table><tr><th>Substance</th><th>Chemical Exergy</th></tr><tr><td>Ni (II)</td><td>232.7 (kJ mol⁻¹)</td></tr><tr><td>Zn (II)</td><td>339.2 (kJ mol⁻¹)</td></tr><tr><td>Cu (II)</td><td>134.2 (kJ mol⁻¹)</td></tr><tr><td><u>Pb</u> (II)</td><td>232.8 (kJ mol⁻¹)</td></tr></table>	Substance	Chemical Exergy	Ni (II)	232.7 (kJ mol ⁻¹)	Zn (II)	339.2 (kJ mol ⁻¹)	Cu (II)	134.2 (kJ mol ⁻¹)	<u>Pb</u> (II)	232.8 (kJ mol ⁻¹)
<u>Sustance</u>	<u>Chemical exergy</u>																							
Carbon Monoxide	275 kJ/ <u>mol</u>																							
Hydrogen	236 kJ/ <u>mol</u>																							
Methane	831 kJ/ <u>mol</u>																							
Carbon (graphite)	410 kJ/ <u>mol</u>																							
Carbon Dioxide	20 kJ/ <u>mol</u>																							
Substance	Chemical Exergy																							
Ni (II)	232.7 (kJ mol ⁻¹)																							
Zn (II)	339.2 (kJ mol ⁻¹)																							
Cu (II)	134.2 (kJ mol ⁻¹)																							
<u>Pb</u> (II)	232.8 (kJ mol ⁻¹)																							

Selected parameters

- CHP
- Exergy of CHP
- Exergy of Products
- Exergy of Metals

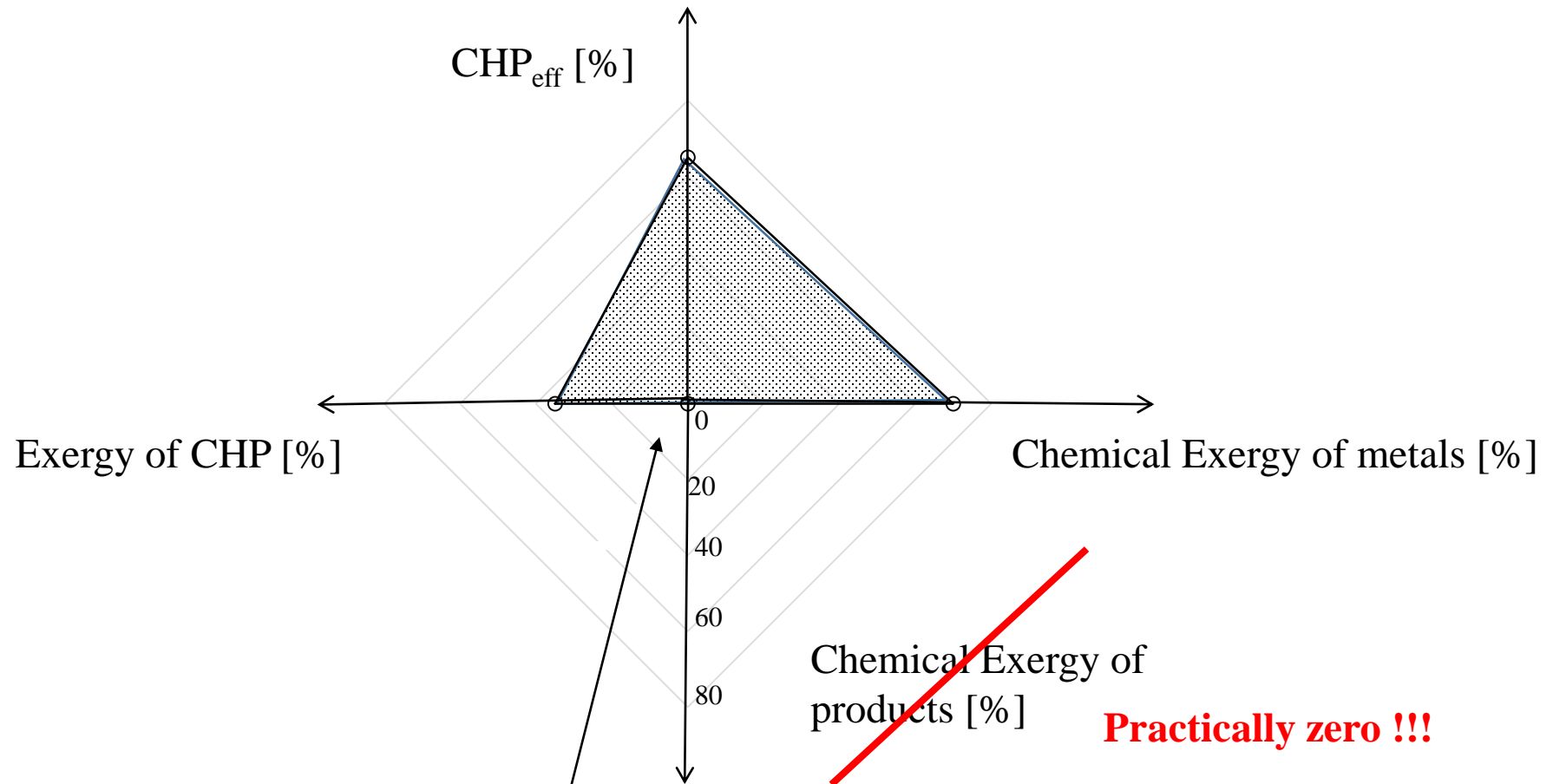
Introducing the 3T Method



Integrated efficiency index - General solution for all thermal treatments

$$\sin\left(\frac{\pi}{2}\right) / 2 * [(\text{Prod- } B_{\text{ch}_{\text{eff}}} * B_{\text{ph}_{\text{eff}}}) + (B_{\text{ph}_{\text{eff}}} * \text{CHP}_{\text{eff}}) + (\text{CHP}_{\text{eff}} * B_{\text{ch}_{\text{eff}}\{m\}}) + (\text{Prod- } B_{\text{ch}_{\text{eff}}} * B_{\text{ch}_{\text{eff}}\{m\}})]$$

Speciacialized 3T Solution for incineration

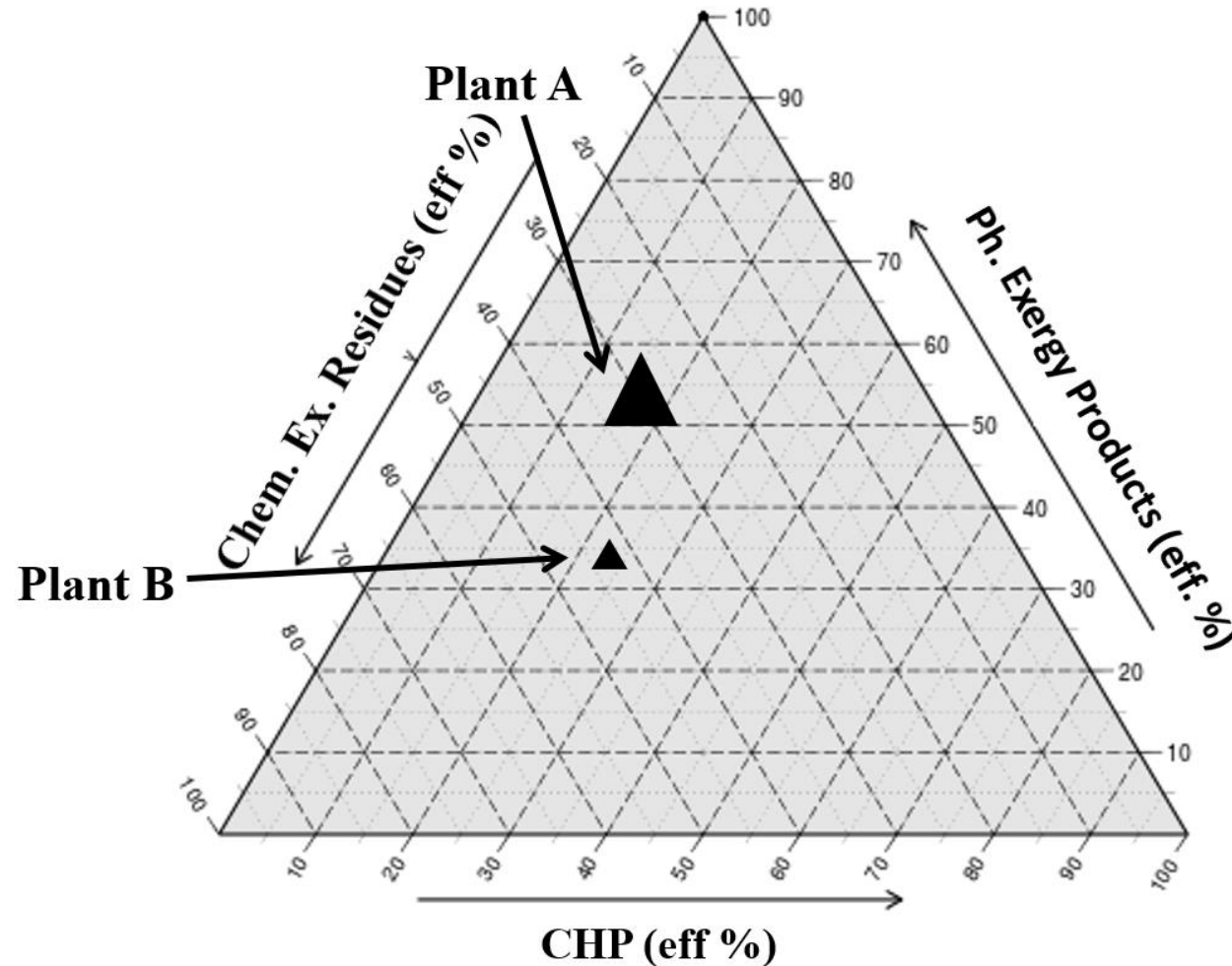


Integrated efficiency index - Specialized solution for combustion

$$[(B_{ph_{eff}} + B_{ch_{eff}} \{m\}) * CHP_{eff}] / 2$$

One more thing !

Mapping of waste-to-energy plants



- The individual efficiencies of each plant are normalized in order to add to 100.
- Placing each plant into a ternary diagram acts as visual mapping.
- The size of each plant's triangle corresponds to the overall value of the T3 value.

Conclusions

- R1 formula has been a great first tool for assessing waste-to-energy plants.
- But the assessment of novel waste-to-energy technologies requires the development of new tools that will be more compatible.
- This work proposes the 3T method where thermodynamic parameters are combined in a radar graph and the overall efficiency is calculated from the area of the trapezoid.
 - The comparison of different technologies becomes possible.
 - The specialized solution allows the data mapping of incineration WtE plants.
- The method includes also the recovery of metals and is in good agreement with the concept of “circular economy”.

ATHENS2017

Assessing the 3T method as a replacement to R1 formula for measuring the efficiency of waste-to-energy plants

Stergios Vakalis, Konstantinos Moustakas and Maria Loizidou

THANK YOU FOR YOUR ATTENTION

email: stergios.vakalis@outlook.com

**5th International Conference on
Sustainable Solid Waste Management**

21–24 June 2017 www.athens2017.uest.gr