

# PRETREATMENT OF DISPOSABLE NAPPIES TOWARDS VALORIZATION OF THE FERMENTABLE FRACTION THROUGH ANAEROBIC DIGESTION

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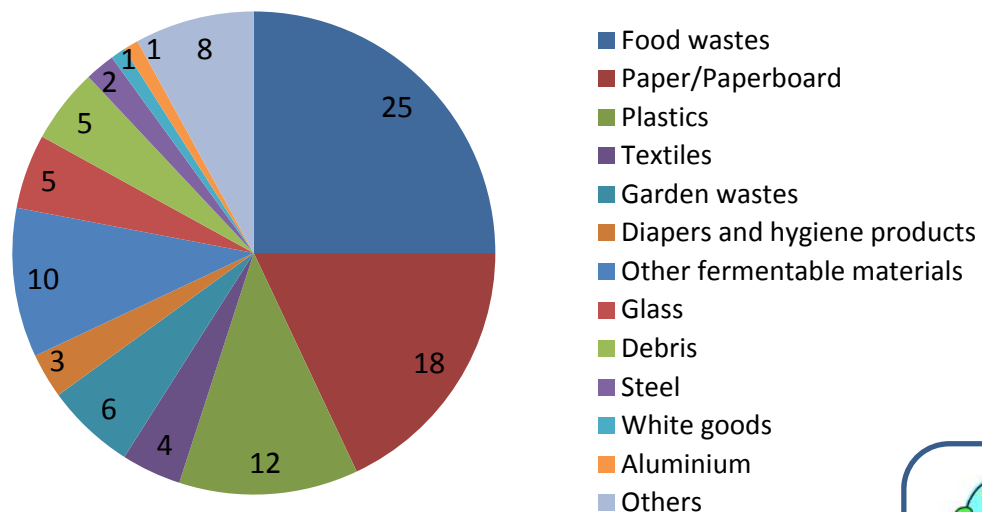


This project has received funding from the European Union's Horizon 2020 research and innovation programme under grant agreement 688995  
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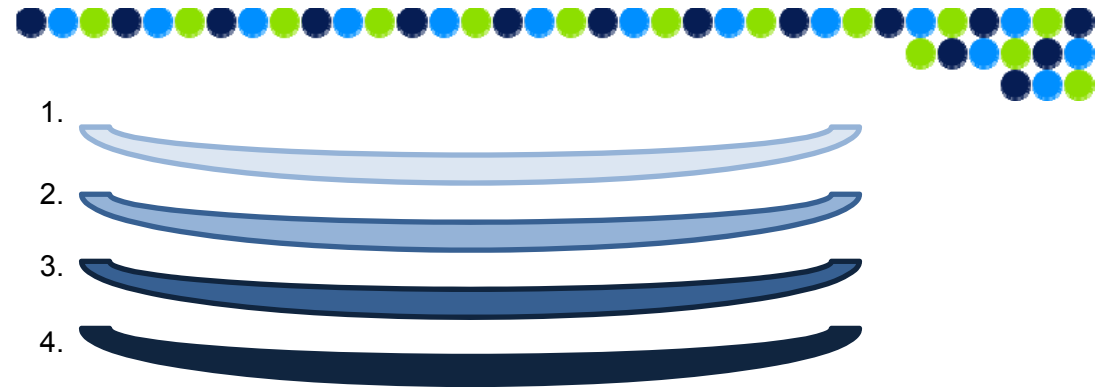


## Composition of MSW generated in EU-27



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## Overview of a nappy



1. Top-sheet layer next to the skin
  - Allows fluid pass through the next layer
  - Materials: Usually PP, sometimes replaced by PLA
2. Acquisition and distribution layer (ADL)
  - Transfers the liquid to the storage part of ADL before it is diffused within the absorbent core structure
  - Materials: PLA, PP, Nonwoven
3. Absorbent core layer
  - Superabsorbent polymer (SAP) and a blend of fiberized fluffy pulp is used, in order to avoid gel blocking
  - SAP: sodium polyacrylate, potasium polyacrylate etc
4. Backsheet
  - Provides a fluid impervious barrier so that moisture is contained within the structure
  - Materials: Usually PE, polyester and nonwoven

Adhesives, hydrophobic fibers, elastics, Velcro.

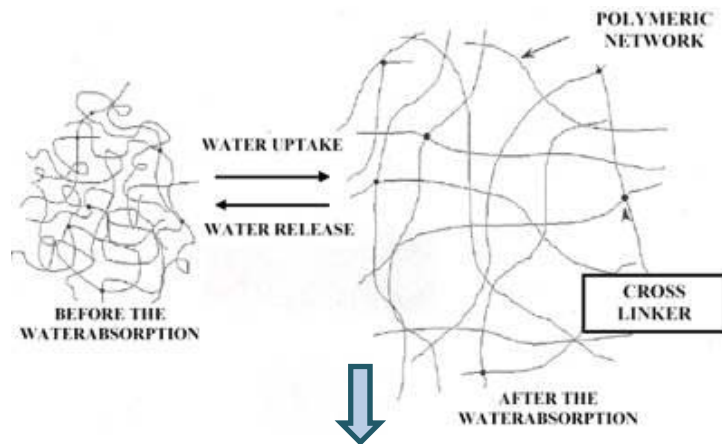


## Composition of an unused baby diaper

Material	%
Fluff Pulp	24
SAP	33
Nonwoven	21
Elastics & adhesive tape	13
PE	5
Adhesives	3
Others	1

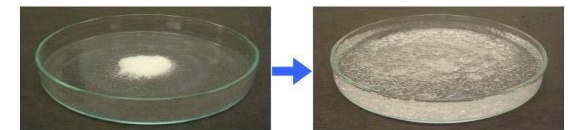


## The main problem



### SAP swelling depends on:

- **time**     $\uparrow$  time  $\rightarrow$      $\downarrow$  swelling ratio (g/g)
- **pH**     $\downarrow$  pH  $\rightarrow$      $\downarrow$  swelling ratio (g/g)
- **temperature**  $\downarrow$  temperature  $\rightarrow$   $\downarrow$  swelling ratio (g/g)
- **Salinity**     $\uparrow$  salinity     $\rightarrow$   $\downarrow$  swelling ratio (g/g)





## Purpose

The development of an alternative approach to landfilling of used nappies, including energy recovery through biofuels production and plastics recycling.

SAP volume minimization is the key for disposable nappies treatment, since it facilitates fluffy pulp recovery and other biodegradable baby by-products in a form of nappies hydrolyzate for further deployment through anaerobic digestion.

# Materials and Methods



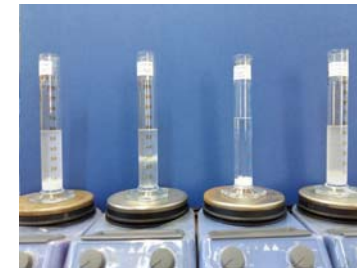
Unused nappies of 2 Greek best-selling brands



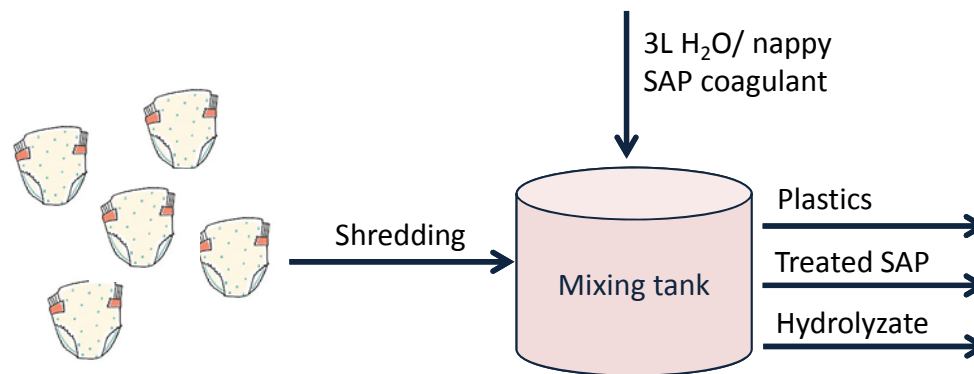
CaCl<sub>2</sub>

MgCl<sub>2</sub>

AlCl<sub>3</sub>



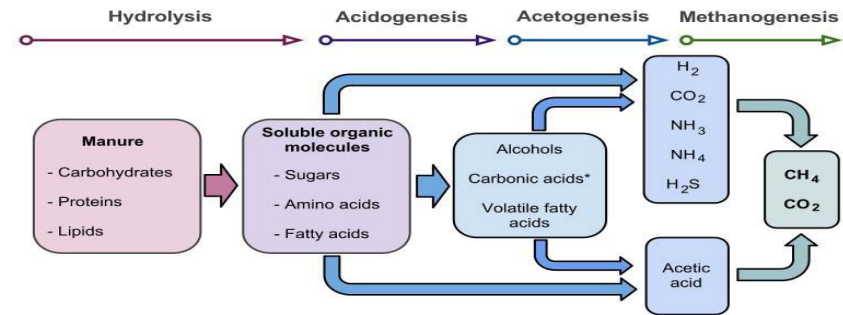
Experimental set up for the SAP deswelling optimization





# Salinity effect for Anaerobic Digestion

- Toxic and inhibition effects by
- Sulfide/ Sulfate ions
  - Ammonia
  - Heavy metals

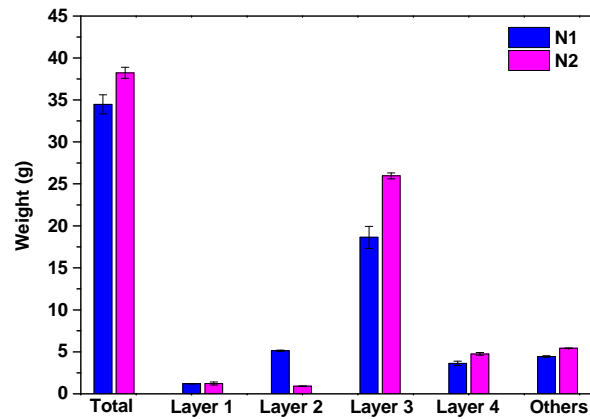


Ion	Concentration (mg/L)	Comments/Observations
Ca <sup>2+</sup>	200 100-200 2500-4000	Ideal concentration for CH <sub>4</sub> production from CH <sub>3</sub> COOH Beneficial for sludge granulation Moderately inhibitory
Mg <sup>2+</sup>	≤7200 ≥400	Cultures could be adapted without changes in growth rate Culture growth ceased, lysis of cells
K <sup>+</sup>	<400	Enhancement in performance
Na <sup>+</sup>	3500-5500	Moderately inhibitory
Al <sup>3+</sup>	≥1000	Methanogenic and acetogenic activity decreased by 50% and 72% respectively

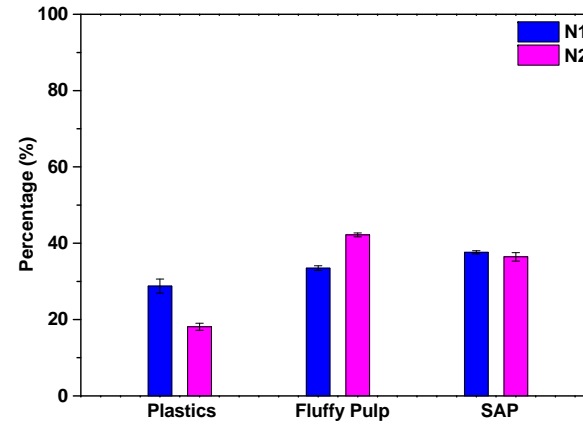




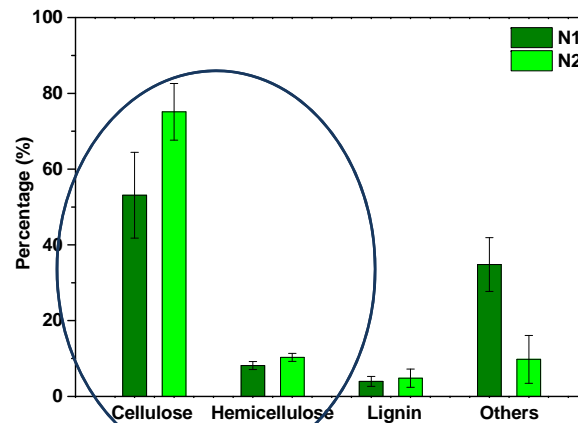
# Results



Weight (g) of unused nappies' layers

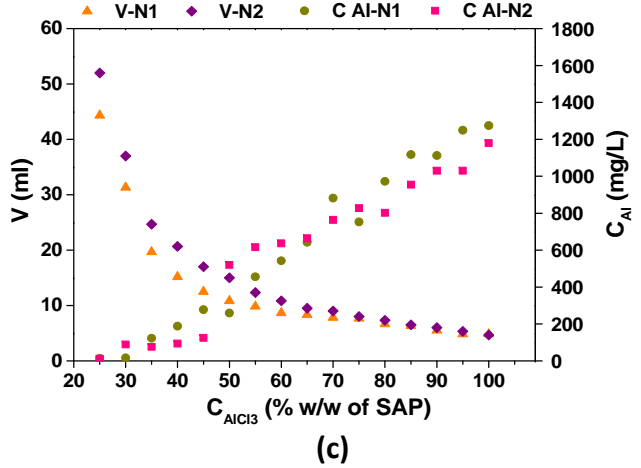
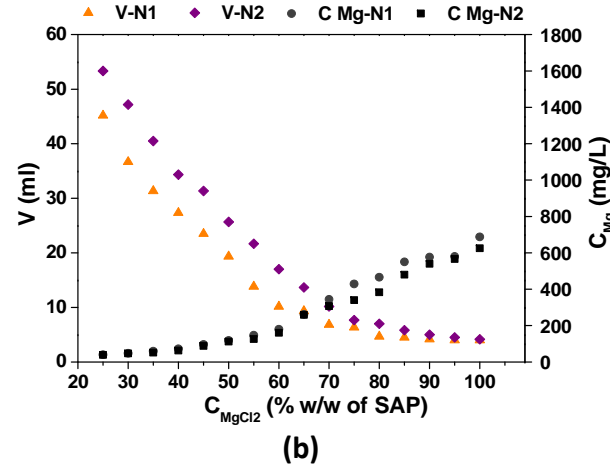
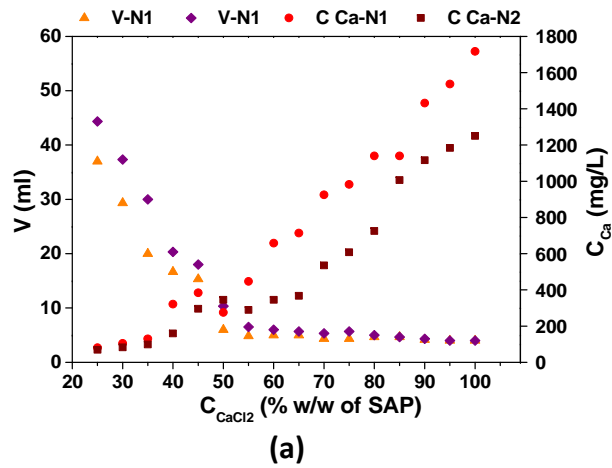


Percentages of plastics, fluffy pulp and SAP in unused nappies.



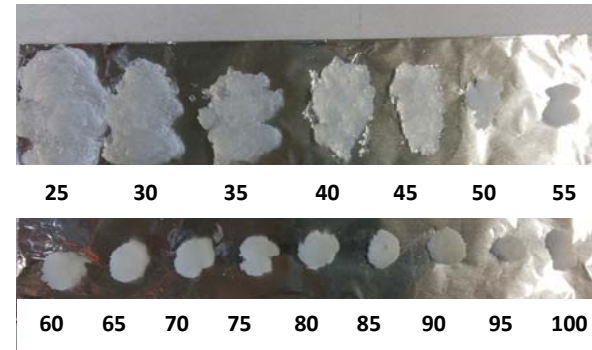
Cellulose, hemicellulose and lignin content in the absorbent layer of unused nappies

# Results



Volume of SAP (left) in correlation with different added concentrations of: (a) CaCl<sub>2</sub>, (b) MgCl<sub>2</sub>, (c) AlCl<sub>3</sub> and residual cation concentration in the bulk solution (right)

1% w/v wet SAP  
50 ml solution  
25-100% w/w reagent  
t=30 min



# Results



Mixture combinations of  $\text{CaCl}_2$  and  $\text{MgCl}_2$  concentrations for SAP deswelling optimization treatments, final SAP volume after a 30 min treatment, residual concentration of salts in the supernatant and final cost of reagents. Data are mean values  $\pm$ SD (n=3).

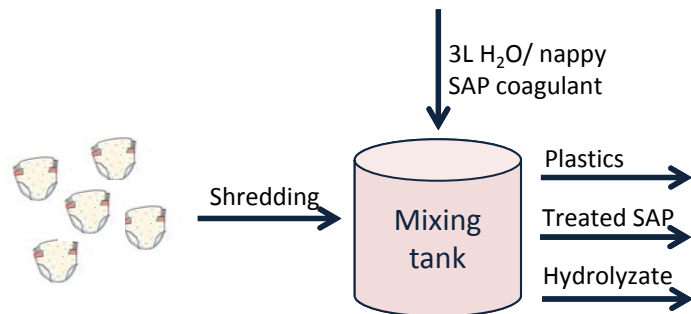
Test No	$\text{CaCl}_2$ (% w/w of SAP)	$\text{MgCl}_2$ (% w/w of SAP)	SAP (ml) $\pm$ SD	Residual $\text{Ca}^{+2}$ (mg/L) $\pm$ SD	Residual $\text{Mg}^{+2}$ (mg/L) $\pm$ SD	Final Cost (€/g treated SAP)
1	20	20	25.00 $\pm$ 1.00	59.30 $\pm$ 10.98	44.32 $\pm$ 1.65	0.1874
2	20	40	7.00 $\pm$ 0.00	84.92 $\pm$ 8.84	99.73 $\pm$ 4.95	0.2638
3	20	60	4.33 $\pm$ 0.58	169.67 $\pm$ 16.40	374.33 $\pm$ 9.09	0.3402
4	20	80	4.00 $\pm$ 0.00	262.00 $\pm$ 27.75	563.50 $\pm$ 31.22	0.4166
5	40	20	5.33 $\pm$ 0.58	353.42 $\pm$ 94.71	80.88 $\pm$ 3.06	0.2984
6	40	40	4.00 $\pm$ 0.00	467.83 $\pm$ 59.10	184.83 $\pm$ 7.42	0.3748
7	40	60	4.00 $\pm$ 0.00	702.83 $\pm$ 108.16	511.33 $\pm$ 21.00	0.4512
8	40	80	4.33 $\pm$ 0.58	487.00 $\pm$ 118.30	582.50 $\pm$ 22.49	0.5276
9	60	20	5.33 $\pm$ 0.58	671.83 $\pm$ 64.73	139.27 $\pm$ 9.58	0.4094
10	60	40	4.83 $\pm$ 0.29	969.00 $\pm$ 56.76	373.17 $\pm$ 20.68	0.4858
11	60	60	5.00 $\pm$ 0.00	898.00 $\pm$ 407.08	555.83 $\pm$ 54.00	0.5622
12	60	80	4.50 $\pm$ 0.50	1224.83 $\pm$ 76.29	704.33 $\pm$ 38.39	0.6386
13	60	50	4.17 $\pm$ 0.29	666.67 $\pm$ 21.69	455.33 $\pm$ 50.58	0.5240
14	40	50	4.10 $\pm$ 0.22	595.70 $\pm$ 108.08	401.50 $\pm$ 36.31	0.4130
15	20	50	4.33 $\pm$ 0.58	160.13 $\pm$ 7.67	276.50 $\pm$ 14.55	0.3020



# Results

Physicochemical characterization of used nappies' hydrolysates following SAP recovery. Data are mean values  $\pm$ SD (n=3).

COD : N : P  
350 : 7 : 1



Parameter	Hydrolysate (GR)	Hydrolysate (SP)
	Value $\pm$ SD	Value $\pm$ SD
pH	7.77 $\pm$ 0.04	7.51 $\pm$ 0.05
Humidity (%)	99.19 $\pm$ 0.10	99.29 $\pm$ 0.13
Alkalinity (g CaCO <sub>3</sub> /L)	0.53 $\pm$ 0.08	0.44 $\pm$ 0.02
t-Carbohydrates (g/L)	1.37 $\pm$ 0.13	1.52 $\pm$ 0.17
d-Carbohydrates (g/L)	0.24 $\pm$ 0.01	0.16 $\pm$ 0.02
Phenols (g/L)	0.04 $\pm$ 0.00	0.01 $\pm$ 0.00
t-COD (g/L)	5.87 $\pm$ 1.26	2.56 $\pm$ 0.11
d-COD (g/L)	1.51 $\pm$ 0.22	1.32 $\pm$ 0.23
Fats and Oils (g/L)	0.06 $\pm$ 0.01	0.02 $\pm$ 0.00
t-Phosphorus (g/L)	0.024 $\pm$ 0.00	0.04 $\pm$ 0.00
d-Phosphorus (g/L)	0.017 $\pm$ 0.00	0.02 $\pm$ 0.00
TKN (g/L)	0.26 $\pm$ 0.02	0.07 $\pm$ 0.04
NH <sub>3</sub> -N (g/L)	0.26 $\pm$ 0.02	0.07 $\pm$ 0.03
TSS (g/L)	4.45 $\pm$ 0.18	3.52 $\pm$ 0.05
VSS (g/L)	4.30 $\pm$ 0.32	3.44 $\pm$ 0.03
TS (g/L)	8.07 $\pm$ 0.97	7.06 $\pm$ 0.67
VS (g/L)	4.99 $\pm$ 0.98	4.56 $\pm$ 0.48
Ca <sup>2+</sup> (g/L)	0.12 $\pm$ 0.01	0.17 $\pm$ 0.01
Mg <sup>2+</sup> (g/L)	0.14 $\pm$ 0.00	0.11 $\pm$ 0.02

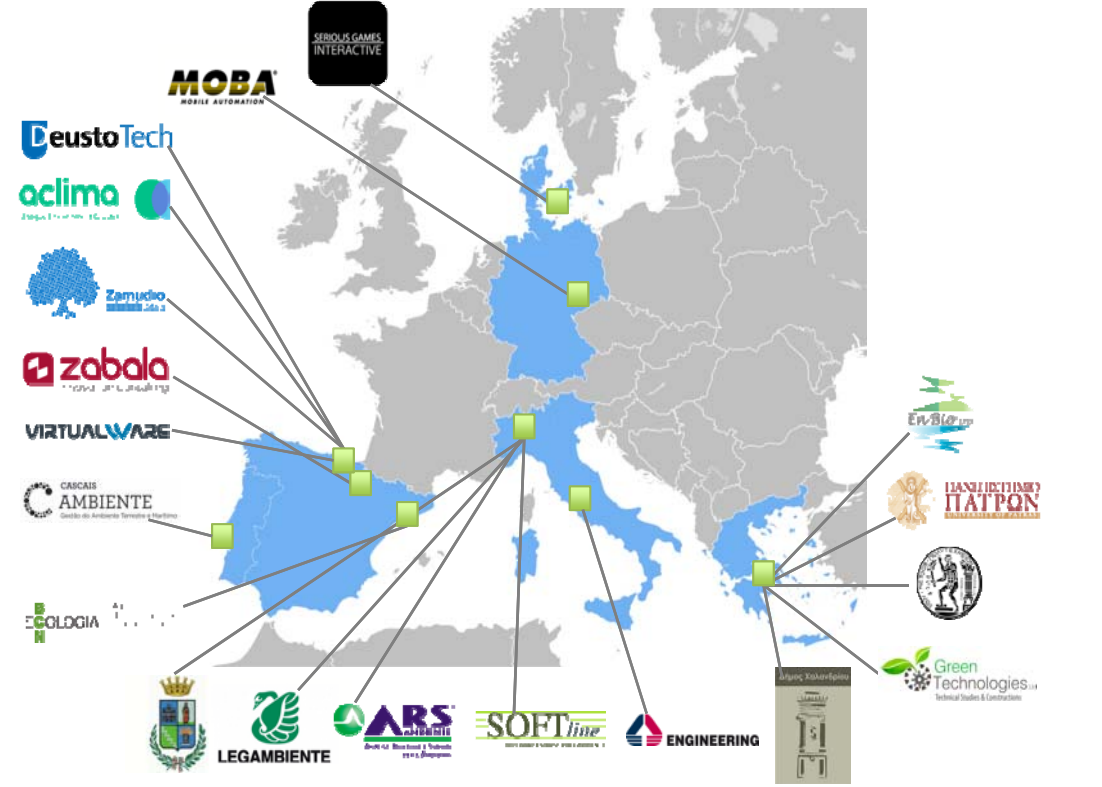



## Conclusions

- Three salts namely  $\text{AlCl}_3$ ,  $\text{MgCl}_2$  and  $\text{CaCl}_2$  were tested for SAP deswelling
- Concerning the environmental impacts  $\text{AlCl}_3$  was rejected as SAP coagulant
- The combination of 20% w/w  $\text{CaCl}_2$  and 50% w/w  $\text{MgCl}_2$  resulted to SAP volume of  $4.33 \pm 0.58$  ml (for 0.5g SAP) with low enough residual ion concentrations and final cost
- The hydrolyzate can be used as substrate in high rate anaerobic digestion systems or as co-substrate in conventional systems without limitations and inhibition risks



Thank you for your attention !!!



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**ATHENS2017**  
 5<sup>th</sup> International Conference on Sustainable Solid Waste Management  
 21-24 June 2017  
 Athens, Greece