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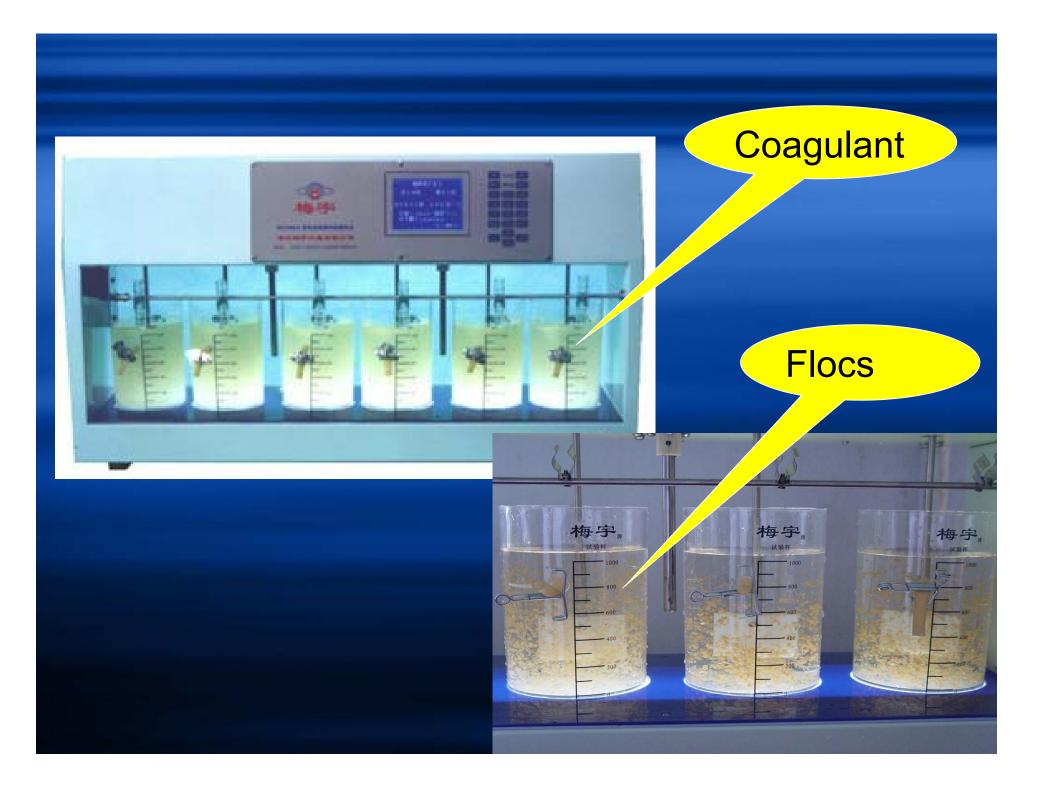
24/06/2016

Coagulant, also called water treatment agent, is used in coagulation processes. Coagulation process is an important unit in treating water and wastewater, which mainly removes colloids and small particles which can not precipitate by themselves. Coagulants can be divided into inorganic, organic and microbial types. Inorganic coagulants will hydrolyze quickly after it is once added into water samples. And then its hydrolysis products will react with the pollutants negatively charged, such as charge neutralization/destabilization, bridging, and sweeping, forming some flocs which can precipitate, thus achieving the purpose of eliminating pollutants after solid-liquid separation.

M-PTF coagulant

Coagulation behaviour of M-PTF

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Research background and significance



M-PTF coagulant

Coagulation behaviour of M-PTF



China, experiencing rapid development, has become one of the countries which were seriously polluted by solid wastes.









The solid waste in China was over 6 billion tons currently and increased rapidly at annual growth of 10%

Most of solid wastes discharged in one point maybe have their values in other points, so, resource disposing becomes one of the main treating methods

Preparation of various inorganic and organic coagulants using various solid wastes has become a promising focus in the field of water and wastewater treatment

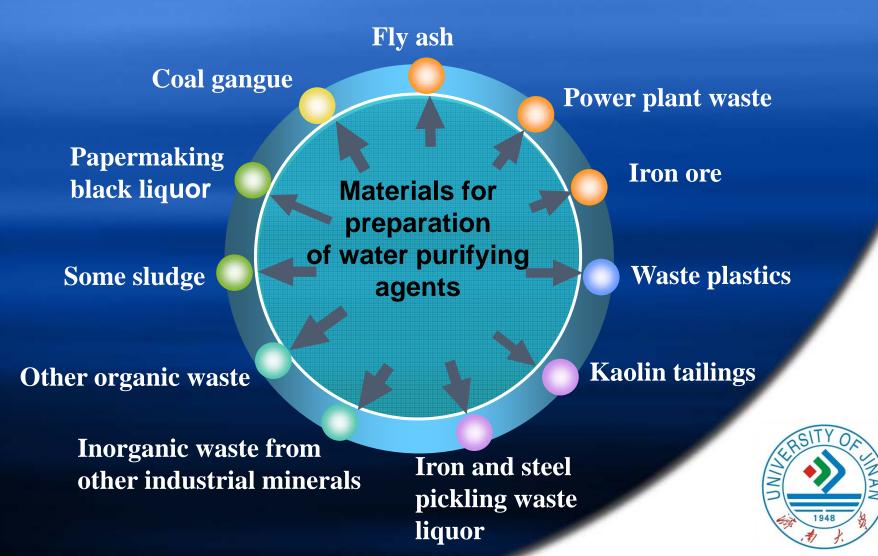








Preparation of various inorganic and organic coagulants using various solid wastes has become a promising focus in the field of water and wastewater treatment



Chine White Preparation of M-PTF using WAFC has its itanium theoretical and practical basis, according with the aim of "waste control by waste".

Lots of wastes and byproducts are generated

"Waste acid filter cake" (WAFC)" (containing elements of Ti, Fe, Al, Si, etc $(w(TiO_2)=70-73\%, and w(Fe_2O_3)=7-9\%)$) coming from the washing process

Fe, Al and Si are often important components for producing inorganic coagulants, while Ti is a new element for prepareing water purifying agents. Fe, Si and Ti are all non-toxic.



Fig.1 Pictures of (a) "Waste acid filter cake" (WAFC)

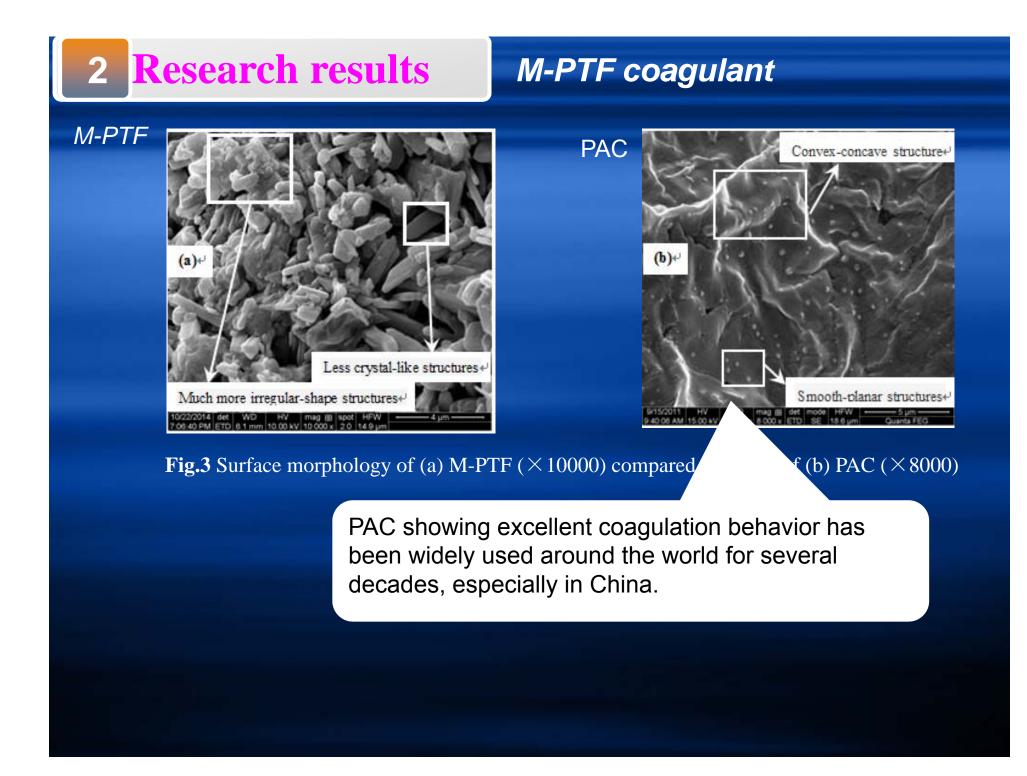
M-PTF coagulant

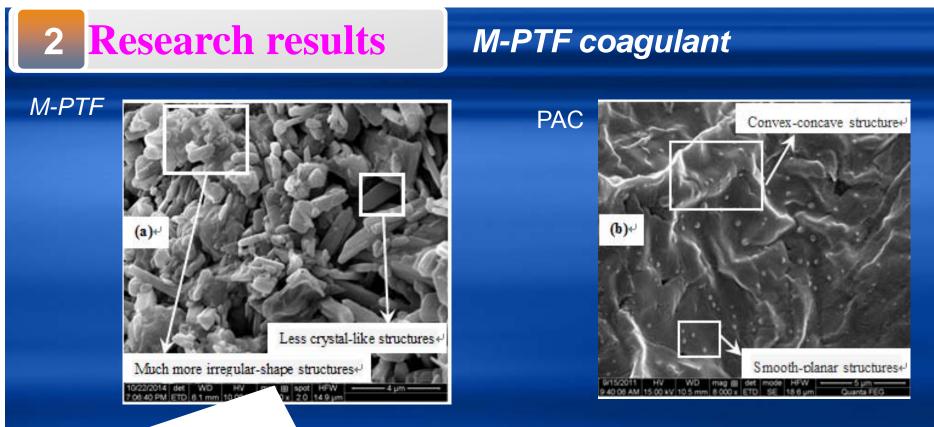
M-PTF



Fig.1(b) Pictures of solid M-PTF and liquid M-PTF







f(a) M DTE (\times 10000) compared with that of (b) PAC (\times 8000)

The surface morphology of M-PTF is so complex, and appeared to be some sort of network structure which was built up by a variety of structures having large surface area and mainly consisted of some irregular and crystalline-like structures, in which the dominant structure was irregular type.

It can be inferred from the complex surface structure that M-PTF was a complex polymer copolymerized by Fe, Ti and many other ions.

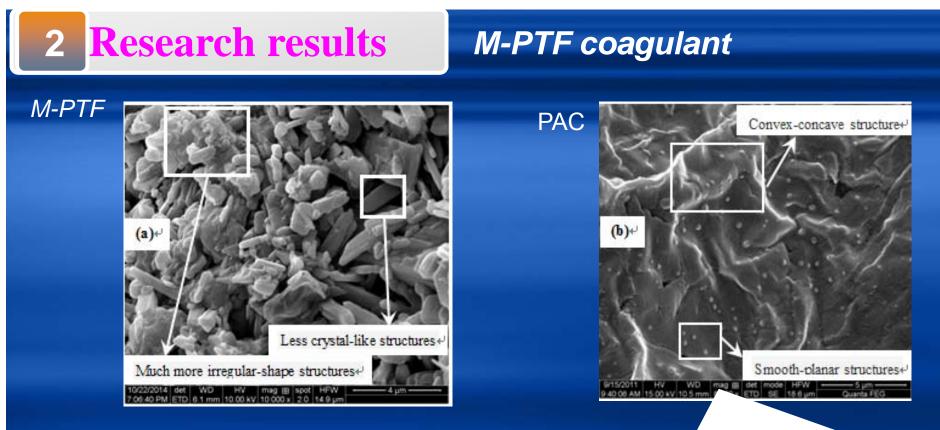


Fig.3 Surface morphology

PAC was composed of a sort of irregular mountain appearance of convex-concave structure, in which some morphology presented a large area of smooth plane, thus leading to smaller surface area of PAC than that of M-PTF.



Influence of dosage

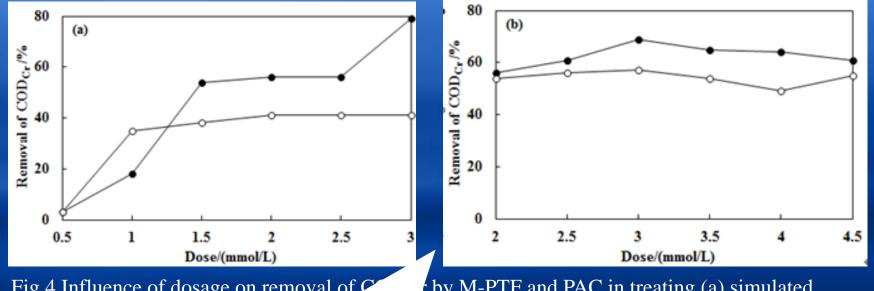
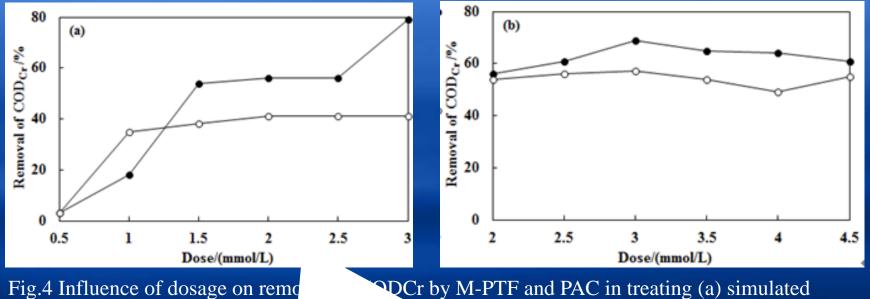


Fig.4 Influence of dosage on removal of O dyeing wastewater and (b) sewage. More represent the standard error of the by M-PTF and PAC in treating (a) simulated) and PAC (\bigcirc). The error bars for all the data points ee experiments

M-PTF gave higher COD removal than PAC for the two types of wastewaters.



Influence of dosage



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Fig.4 Influence of dosage on remo

 \bigcirc). The error bars for all the data points

greatest COD removal by M-PTF The achieved about 80% at dosage 3 mmol/L, more 40% than PAC at the same dosage. While PAC posed the greatest COD removal (35%) at dosage 1 mmol/L, and then almost unchanged with the increasing of dosages.



Influence of dosage

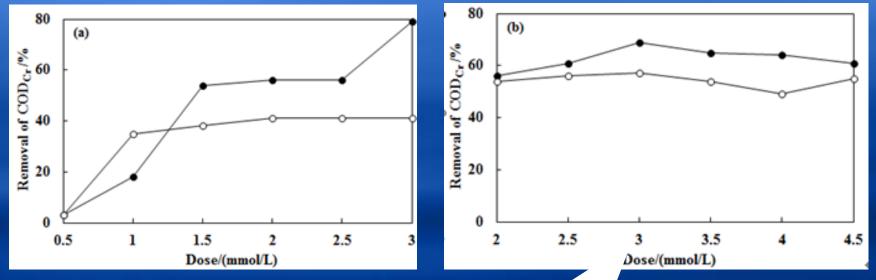


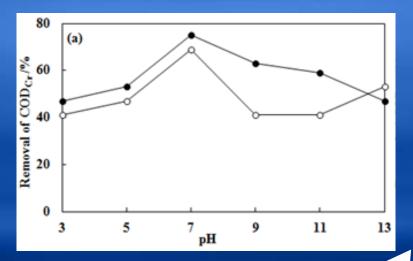
Fig.4 Influence of dosage on removal of CODCr by M-PTF and dyeing wastewater and (b) sewage. M-PTF (\bigcirc) and PAC (\bigcirc) represent the standard error of the mean of three experi-

In treating (a) simulated fror bars for all the data points

M-PTF gave the greatest COD removal (around 70%) at dosage 3 mmol/L, more 10% than PAC.

Coagulation behaviour of M-PTF

Influence of coagulation pH



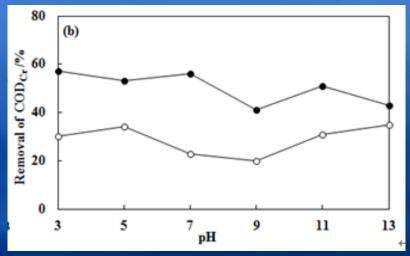


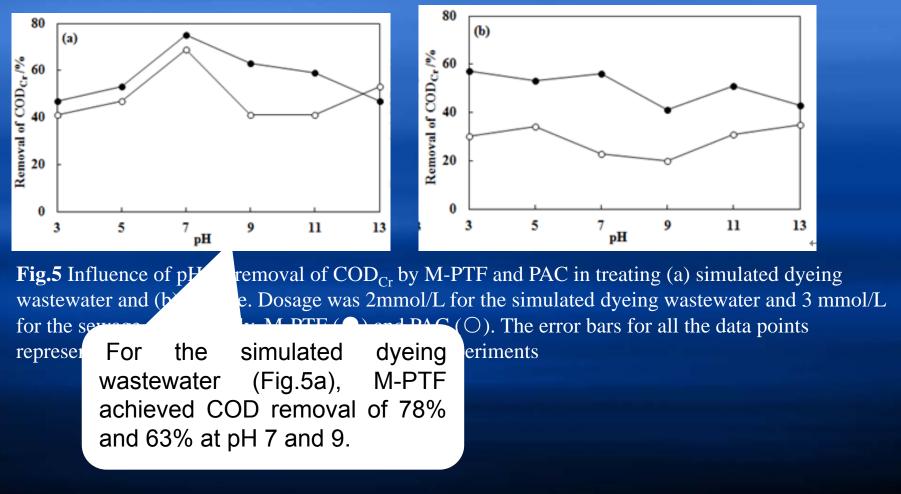
Fig.5 Influence of pH on removal of **CO** wastewater and (b) sewage. Dose for the sewage_respectivel

by M-PTF and PAC in treating (a) simulated dyeing nol/L for the simulated dyeing wastewater and 3 mmol/L PAC (\bigcirc). The error bars for all the data points

represent the M-PTF almost gave higher COD removal than PAC for the two types of wastewaters over the tested pH range.

Coagulation behaviour of M-PTF

Influence of coagulation pH

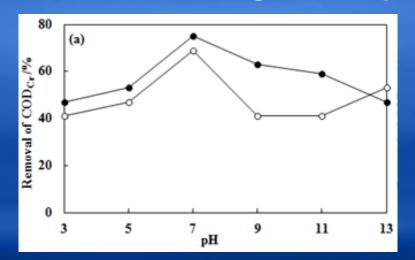


Coagulation behaviour of M-PTF

nulated dyeing

mmol/L

Influence of coagulation pH



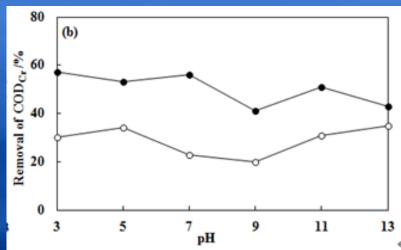
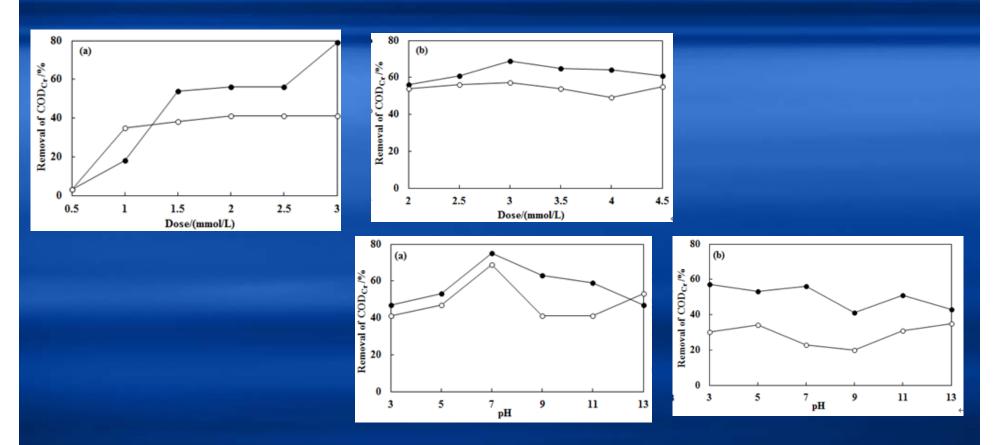


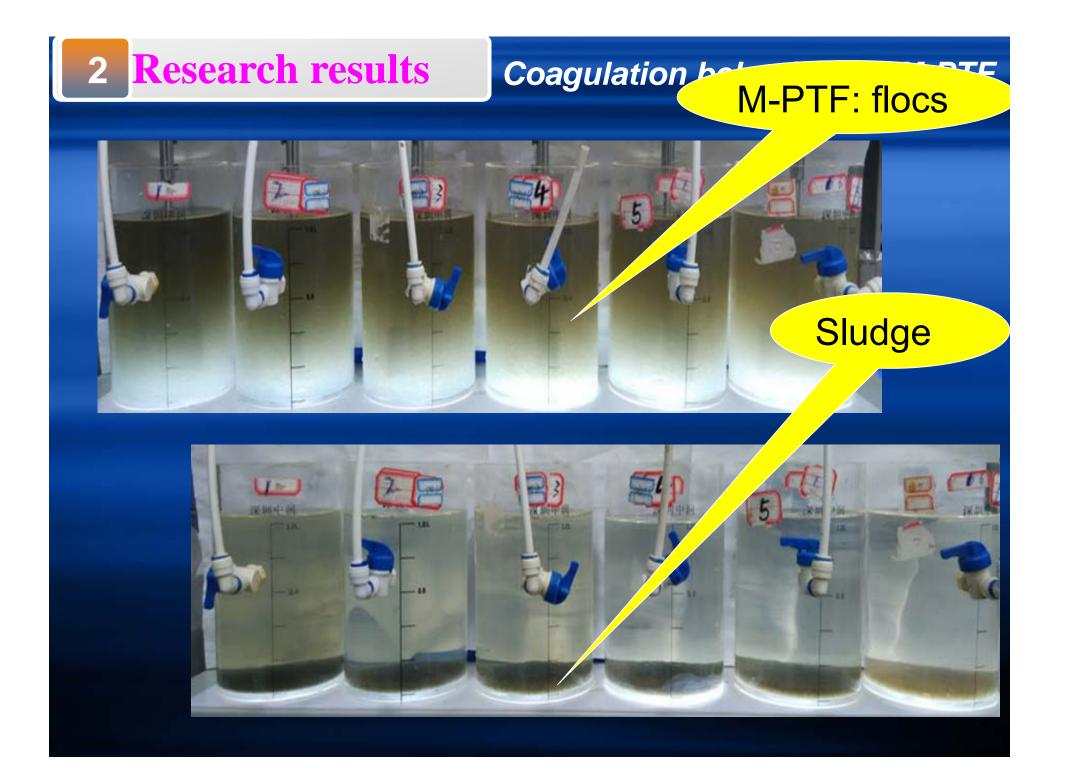
Fig.5 Influence of pH on removal of COD_{Cr} by M-PTF and PAC in trewastewater and (b) sewage. Dosag for the sewage, respectively. M-PT represent the standard error of the COD by M-PTF achieved a

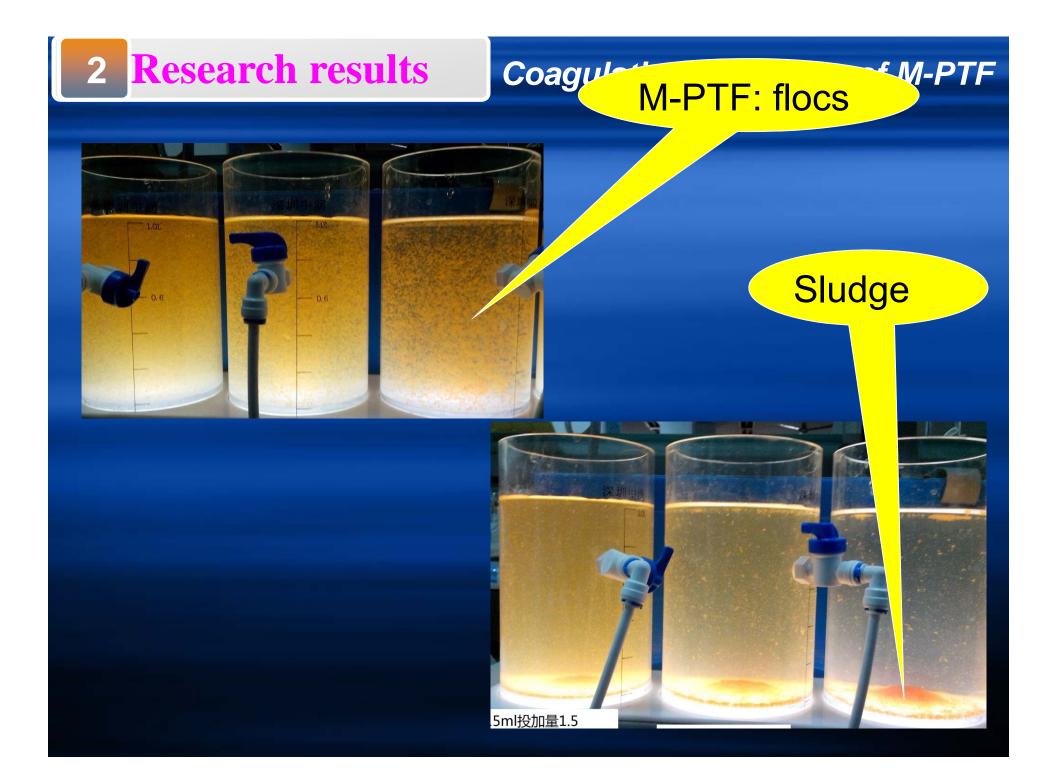
For the sewage (Fig.5b), the removal of COD by M-PTF achieved about 53, 56 and 51% at pH 5, 7 and 11, respectively, more 29, 33 and 20% than PAC, respectively.

2 **Research results**



M-PTF can be adapted to a greater range of dosage at the same pH values and a wider pH range at the same dosages than PAC, which makes M-PTF have much more application significance in treating complex wastewater samples for organic matters removal.







Research conclusions

M-PTF prepared from WAFC is an excellent water treatment agent, and posed higher COD removal than PAC for different wastewaters at the tested pH and dosage ranges. So M-PTF can be adapted to a greater dosage range and a wider pH range than PAC.

M-PTF giving better removal of organic matters mainly depended on the complex micro-characteristics and the characteristics of hydrolysis products deduced.

