



Valorization of hemicellulose-biomass side streams via catalytic hydrogenation into value added chemicals and fuels

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Utilization of Biomass



A successful commercial example of biomass derived plastic

replacing PET



https://www.avantium.com/yxy/yxy-technology/

Lignocellulosic Biomass

Structure



Source: Ritter S.K., Lignocellulose: A Complex Biomaterial, Plant Biochemistry, 86(49) (2008) 15



Hemicellulose: general formula(C₅H₈O₄)_n



C₅ & C₆ sugars, uronic acids, acetyl units

Lignin:





Phenolic monomers

Cellulose: 30-50%, Hemicellulose: 20-40%, Lignin: 15-25%

Others, 5-35% - Ash 3-10% (Si,Al,Ca,Mg,K.Na), Extractives: Resins, Phenols, Sterols, etc

Lignocellulosic biomass raw

materials

- Agricultural and forestry residues/waste (wheat straw, trimmings, tree branches)
- Industrial wood processing residues (e.g. sawdust)
- Food industry waste (e.g. kernels, shells)
- Municipal solid waste (e.g. waste paper)
- Perennial or annual crops with high yield 1-4 ton/1000m² year (e.g. eucalyptus, pseudoacacia, willow, miscanthus, switch grass, cellulosic sorghum,..)



Olive kernels

Agricultural & forestry Residues/wastes





Miscanthus



Robinia pseudoacacia



Biomass (agricultural) residues in EU-28 (2006-201<u>5)</u>





Maize (80.37Mt)

+ others (49.22 Mt)

Barley (50.10Mt)

Permanent Crops (21.86Mt)





Olive trees (17.11Mt) Vineyards (4.08Mt)



European Commission Report, 2018

Oil-bearing crops (73.10Mt)



 Rapeseed
 Sunflower

 (53.99Mt)
 (14.63Mt)

 + others
 (4.48 Mt)





Sugar beet (9.23Mt) Potatoes (4.18Mt)

+ others (0.68 Mt)

Biomass (agricultural) residues in Greece





non, Eurobionet-biomass survey in Europe, Country report of Greece, 2008nter for Renewable Energy Sources & Saving, Greece, 2



Integrated lignocellulosic biomass valorization (Biorefinery)



Hydrothermal pre-treatment (in pure H₂O)

Solid product

Cellulose +





reactor



Enzymatic Hydrolysis Glucose

40

Liquid product

Hemicellulose

monomers and oligomers, xylose, furfural, acetic, formic acid, etc.

Severity factor (logRo)

$$R_0 = t \cdot exp \frac{(T-100)}{14.75}$$

Experimental conditions:

Temperature : 130-220°C
Time : 15-180 min
LSR: 15
Stirring: 400 rpm

C.K. Nitsos, K.A. Matis, K.S. Triantafyllidis, ChemSusChem, 6 (2013) 110 – 122 C.K. Nitsos, T. Choli-Papadopoulou, K.A. Matis, K.S. Triantafyllidis, ACS Sust. Chem. & Engin. 4 (2016) 4529-4544 C. K. Nitsos, P. A. Lazaridis, A. Mach-Aigner, K. A. Matis, & K. S. Triantafyllidis, ChemSusChem (2019) 12 (6): 1179

Generalized reaction scheme

Hemicellulose hydrolysis at subcritical



Self-catalyzed hydrolysis (pH 5 [] 2.5)

□The catalyst (acetic acid) is a biomass

component

Cellulose hydrolysis at subcritical



Sugars dehydration products



Evolution of main structural components in hydrothermally treated solids



Xylose and furfural concentration vs. % hemicellulose removal



Catalytic hydrogenation of furfural: General reaction mechanism-possible



Dominant pathways/products depend on catalyst type, reaction parameters and solvent (acting or not as H-donor for inducing transfer hydrogenation)

Y. Wang, P. Prinsen, K.S. Triantafyllidis, S.A. Karakoulia, A. Yepez, C. Len, R. Luque, ChemCatChem 2018, 10, 3459–34 Wang, Y., Prinsen, P., Triantafyllidis, K. S., Karakoulia, S. A., Trikalitis, P. N., Yepez, A., Christophe Len, Luque, R. . ACS Sustainable Chemistry & Engineering, 2018, 6(8), 9831-9844

Furfural derived chemicals and fuels



R. Mariscal, P. Maireles-Torres, M. Ojeda, I. Sádaba, M. López Granados, Energy Environ. Sci., 2016,9, 1144-1189

Catalytic hydrogenation experiments of hemicellulose stream



□ Solvent: Ethyl acetate, H₂O, EtOH & IPA (as H₂ donor - transfer

hydrogenation)

- \Box H₂ gas: 30 bar at room temp.
- □ Temperature: 180 °C
- Catalyst: Ru Pd Pt Cu Ni supported on Micro/mesoporous Activated Carbon

Catalysts for furfural hydrogenation

Catalyst	Total SSA (m²/g)	Total pore volume (cc/g)	Micropore area (m²/g) / volume (cc/g)	Meso/macro- pore & external area (m²/g) / volume (cc/g)	Crystal size (nm)
Activated carbon (AC)	1281	0.946	841/0.343	440 / 0.603	-
3%Pt/AC	1180	0.847	759 / 0.309	421 / 0.538	13.6
3%/Pd/AC	1338	0.947	886 / 0.362	452 / 0.585	16.6
5%Ni/AC	1251	0.884	831/0.343	420 / 0.541	6.8
10%Ni/AC	1246	0.895	806 / 0.329	440 / 0.566	Ni(0) 23.5- NiO 6.1
10%Cu/AC	1172	0.828	768 / 0.313	403 / 0.515	Cu(0) 23.2 - Cu ₂ O 16.6
5%Ni-15%W/AC	1025	0.720	678 / 0.276	347 / 0.444	Ni(0) 7.8 - WO ₂ 9.9 - NiWO ₄ 15.5





(a) 5%Ni/AC, (b) 3%Pt/AC, (c) 3%Pd/AC, (d) 10%Cu/AC

Effect of reaction time & temperature

	Solven	Time	т	H ₂	X				2-
Catalyst	t	(h)	(°C)	(bars)	(%)	FAL	THFAL	2-MF	MTHF
						OF OF	Н Дорон	H ₃ C	CH ₃
3%Pd/AC	EtOAc	1	180	30	15.6	10.1	Ŭ.U	43.4	Ŭ.Ŭ
3%Pd/AC	EtOAc	3	180	30	19.6	6.0	0.0	58.4	0.0
3%Pd/AC	EtOAc	6	180	30	29.3	3.6	0.0	58.6	0.0
3%Pd/AC	EtOAc	9	180	30	34.8	5.8	1.1	74.6	11.5
3%Pd/AC	EtOAc	6	180	30	19.6	6.0	0.0	58.4	0.0
3%Pd/AC	EtOAc	6	220	30	43.4	4.4	3.8	69.4	13.2



Effect of catalyst type

	Solven	Time	т	H_2	X				2-
Catalyst	t	(h)	(°C)	(bars)	(%)	FAL	THFAL	2-MF	MTHF
							Ч Домон	H ₃ C	
3%Pd/AC	EtOAc	3	180	30	19.6	6.0	Ŭ.U	58.4	0.0
3%Pt/AC	EtOAc	3	180	30	72.9	3.5	1.5	74.3	0.0
10% Ni/AC	EtOAc	3	180	30	<u>19.3</u>	21.7	1.3	75.9	0.0
10%Ni/15%W-									
AC	EtOAc	3	180	30	53.7	18.0	5.4	42.1	0.0

Pt based catalyst were very reactive and selective towards 2-MF (polar, aprotic solvent)

Ni based catalysts exhibit also high selectivity to 2-MF but activity improvement is needed

Catalytic transfer hydrogenation of furfural (solvent acting as hydrogen donor)



Ni, Cu, Pt, Pd on micro/mesoporous carbon



			Yield (%)						
Entry	Catalyst		FA	THFA	MF	MTHF	iPrOMF	Mass	
		Conversion (%)		H OH		$\langle $		balance (%)	
1		2	6	0	0	0	0	104	
2	10%Cu/AC	24	22	0	2	1	1	103	
3	3%Pd/AC	47	21	1	5	2	5	87	
4	3%Pt/AC	93	47	1	24	3	5	87	
5	5%Ni/AC	85	6	1	66	2	3	93	
6	5%Ni/AC ^b	10	10	1	1	0	0	102	
7	5%Ni/AC ^c	95	20	1	50	1	1	78	
8	5%Ni/AC ^d	87	13	1	9	2	0	38 ^e	
9	5% Ni/AC ^f	67	38	1	17	1	13	103	

^a 200 °C, 5 h, 0.35 M furfural in 60 mL isopropanol, 30 bars H_2 , ^b 0 bar $H_2/200$ °C, ^c 0 bar $H_2/260$ °C, ^d In methanol, ^eUnknown compound eluting at 3.8 min in GC analysis, not included (48 % of total peak area), ^f Spent catalyst recovered after the experiment in entry 5

An example of the successful collaboration between Greece, France and Spain, involving training/exchange of young scientists within the frame of European COST Action "LIGNOVAL"



Y. Wang, P. Prinsen, K.S. Triantafyllidis, S.A. Karakoulia, P.N. Trikalitis, A. Yepez, C. Len, R. Luque, ACS Sustainable Chem. Eng. 2018, 9831–9844

Y. Wang, P. Prinsen, K.S. Triantafyllidis, S.A. Karakoulia, A. Yepez, C. Len, R. Luque, ChemCatChem 2018, 10, 3459-3468

Catalytic hydrogenation experiments of "real"

hemicellulose stream



Enzymatic hydrolysis optimization (beech sawdust)





Chemo-catalytic processes

A synergy between thermochemical pretreatment, chemoand bio-catalysis is necessary for more efficient biomass valorization

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