

# Valorisation of cellulosic sludge by fermentation process: VFAs production and phosphorus recovery



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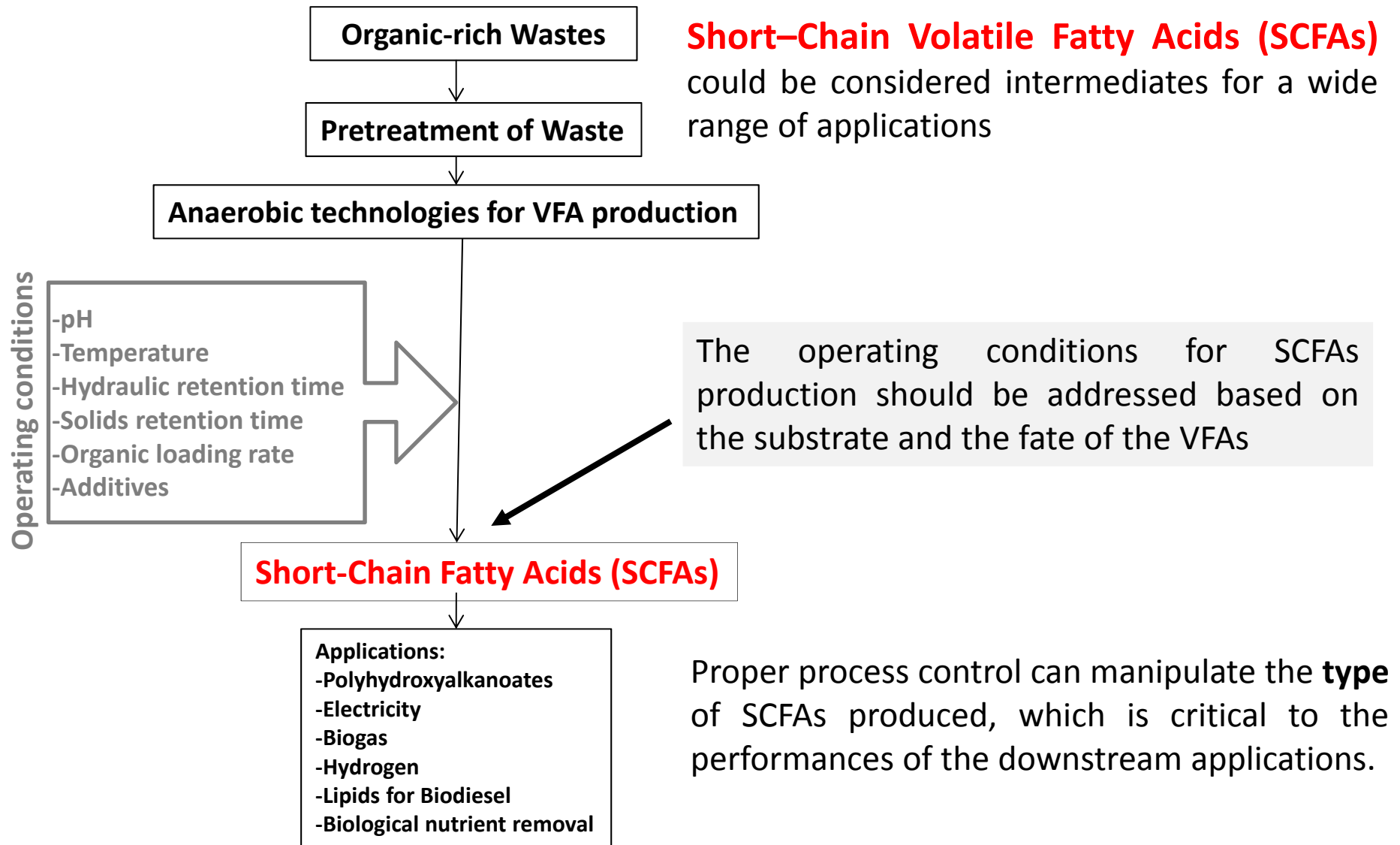
**13<sup>th</sup> IWA**  
Specialized Conference on  
Small Water and Wastewater  
Systems  
**5<sup>th</sup> IWA**  
Specialized Conference on  
Resources-Oriented Sanitation

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# Background

- Municipal wastewater contains around **100-120 gCOD/(inhabitant per day)**, however diluted in **250-350 L/(inhabitant per day)**;
- Around 50-80% of the suspended solid can be efficiently removed by the sieving of the wastewater, in which up to 35% is toilet paper (Ruiken et al., 2013, Water Research);
- The sewage sludge might be considered as a **challenging feedstock** to be processed for bio-based applications (waste-to-chemicals and bio-product value chain);

# Production and Application of Waste-derived SCFAs

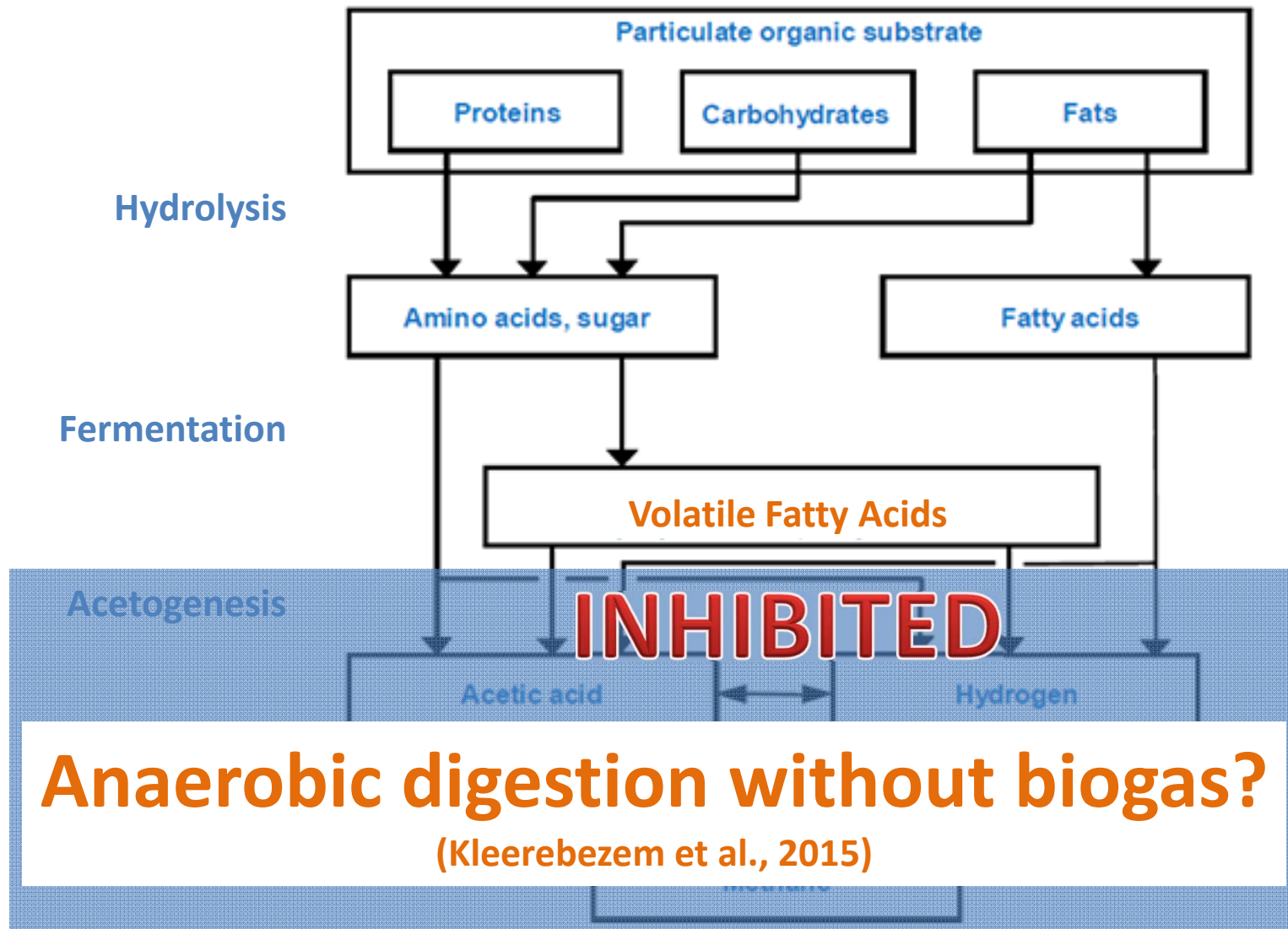


# Anaerobic biodegradation of toilet paper

Up to now, the utilization of cellulosic sludge has been investigated for the production of **biogas**, the production of thermal and electric energy and **compostable material**;

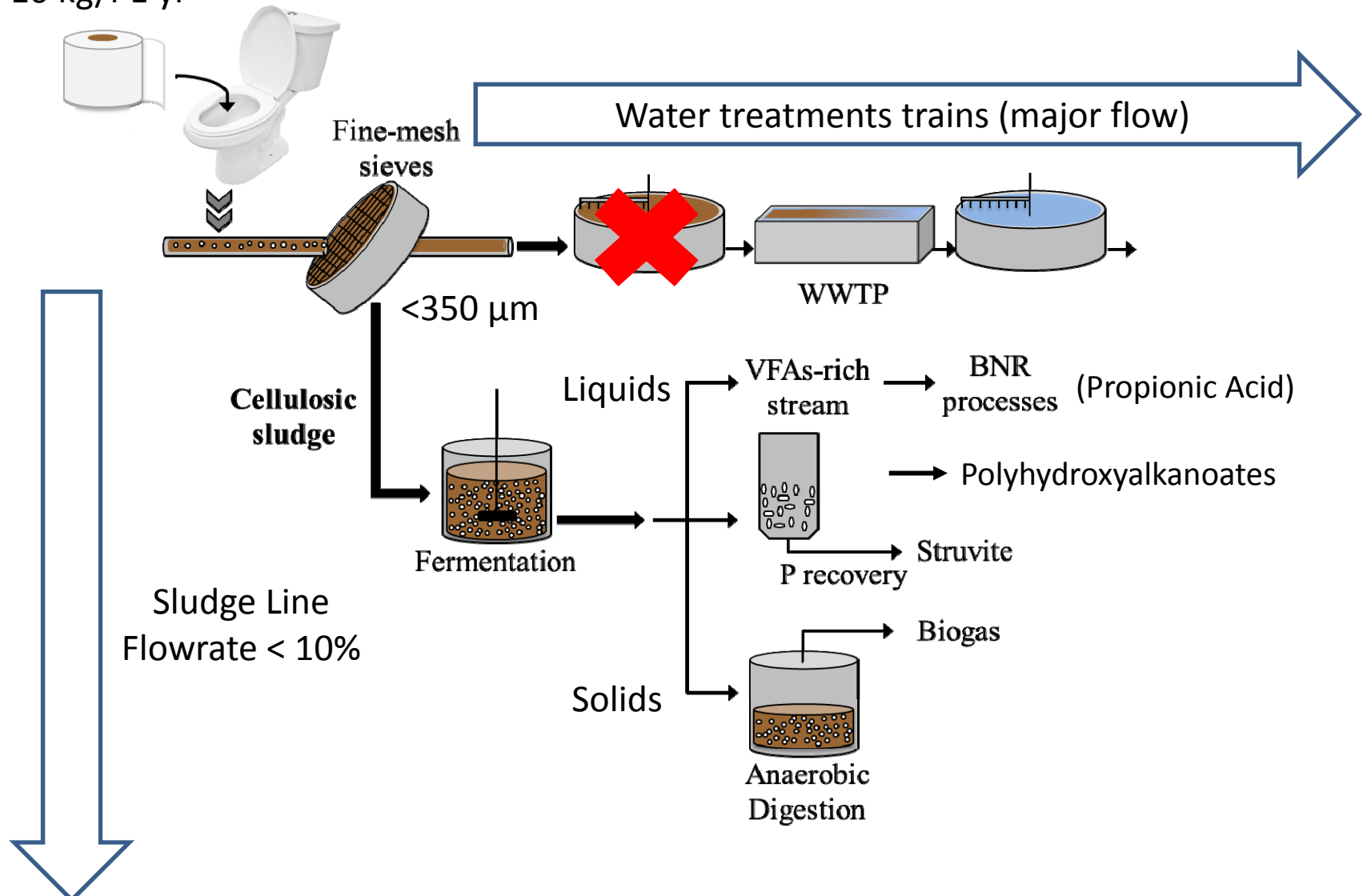
Condition	pH	°C	Type fibre	Removal%	Days
Anaerobic	7.0	30	Toilet paper from recycled paper	100	8
Anaerobic	7.0	24	Toilet paper from recycled paper	100	12
Anaerobic	7.0	9	Toilet paper from recycled paper	10	20
Anaerobic	7.0	24	4 layer type 1 toilet paper	100	15
Anaerobic	7.0	24	4 layer type 2 toilet paper	100	20
Anaerobic	7.0	24	Sieving material 50% SS removal	100	20
Anaerobic	7.0	17	Sieving material 50% SS removal	50	20
Anaerobic	5.0	24	Toilet paper from recycled paper	40	20

# Anaerobic biodegradation of cellulosic sludge



# Valorization of cellulosic sludge

Up to 10 kg/PE yr



# Objective

- To investigate the production of Volatile Fatty Acids (VFAs) and the phosphorus ( $\text{PO}_4^{3-}\text{-P}$ ) release from the fermentation of cellulosic sludge.
- To study the potential recovery of phosphorus by struvite crystallisation.

# Characteristics of the cellulosic sludge

- Origin: WWTP of Verona municipality (North of Italy);
- Cellulosic sludge used was prepared by mixing primary sludge + toilet paper;
- The toilet paper was kept in wastewater for 4 h to achieve a cellulosic sludge with similar characteristics to those found in WWTPs

Parameter	Units	Cellulosic Sludge
TS	g/L	50-55
VS	%VS	80-85
total COD	gCOD/gVS	1.0-1.1
soluble COD	mgCOD/gVS	45-50
TN	mgN/gVS	30-35
TP	mgP/gVS	10-15
Content of cellulose	%TS	35-38





# Outline of the batch cellulosic fermentation experiments

N° Experiment	Temperature (°C)	Initial fermentation pH
1-4	37	6.3 (Uncontrolled),8,9,10
5-8	55	6.3 (Uncontrolled),8,9,10
9-12	70	6.3 (Uncontrolled),8,9,10

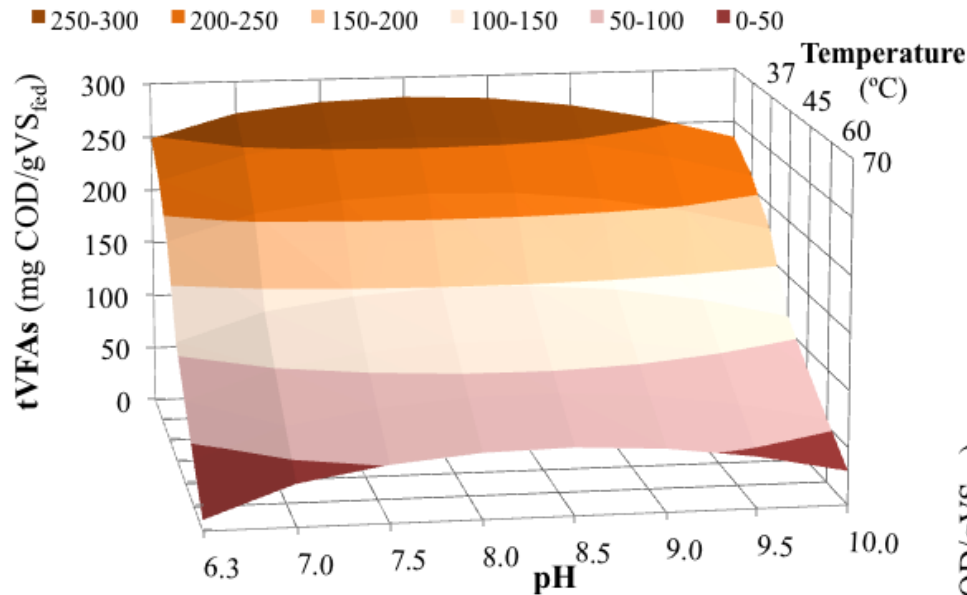
The response surface methodology (RSM) was applied:

$$Y(z) = \beta_0 + \beta_1 x_1 + \beta_2 x_2 + \beta_{1,2} x_1 x_2 + \beta_{1,1} x_1^2 + \beta_{2,2} x_2^2$$

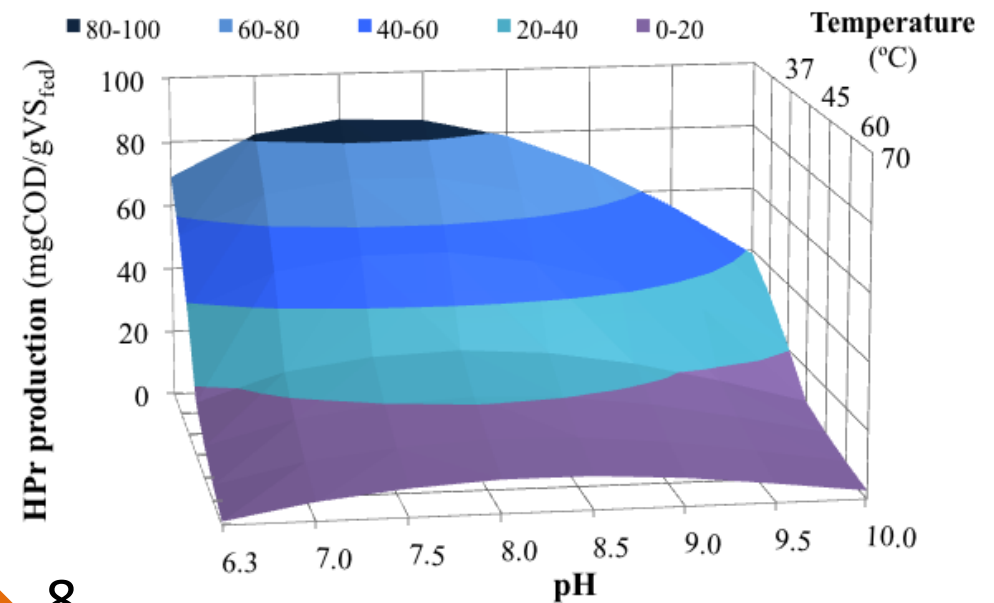
- $Y(z)$ : propionic acid production (mg COD/g  $V_{s_{fed}}$ )
- $X_1$ : initial pH
- $X_2$ : temperature

# 3D Surface plots: SCFA and Propionic acid production

## SCFA production

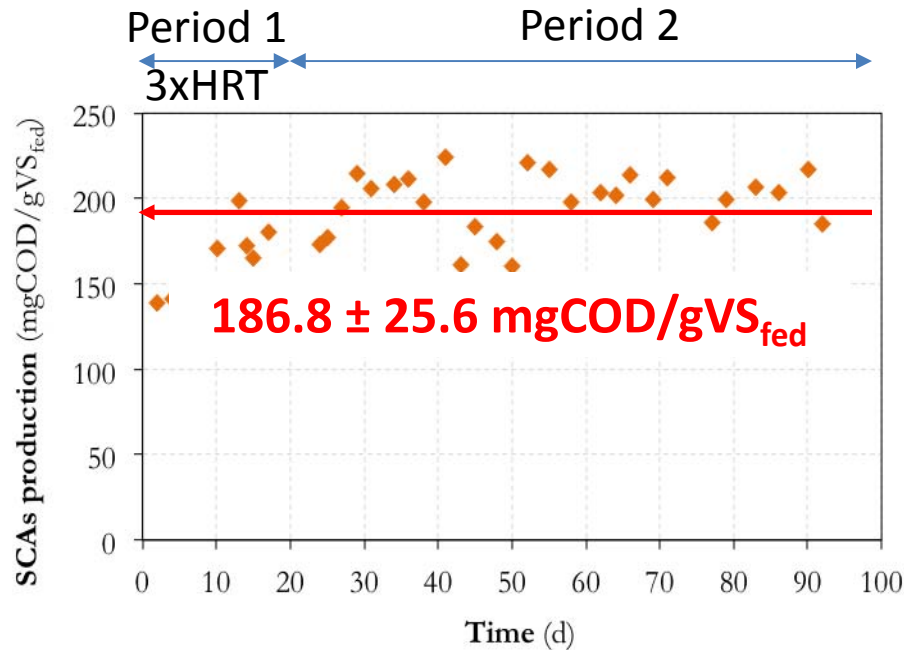


## Propionic acid production

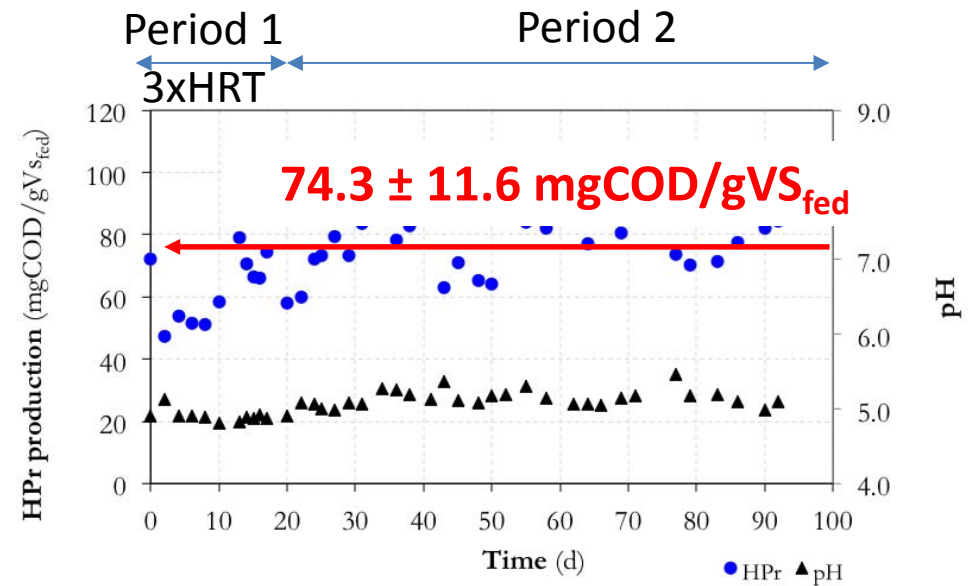


- High SCFAs and HPr production:
  - T: 37 °C
  - HRT: 4 d
  - Initial pH range: 7.5 - 8.5 → 8

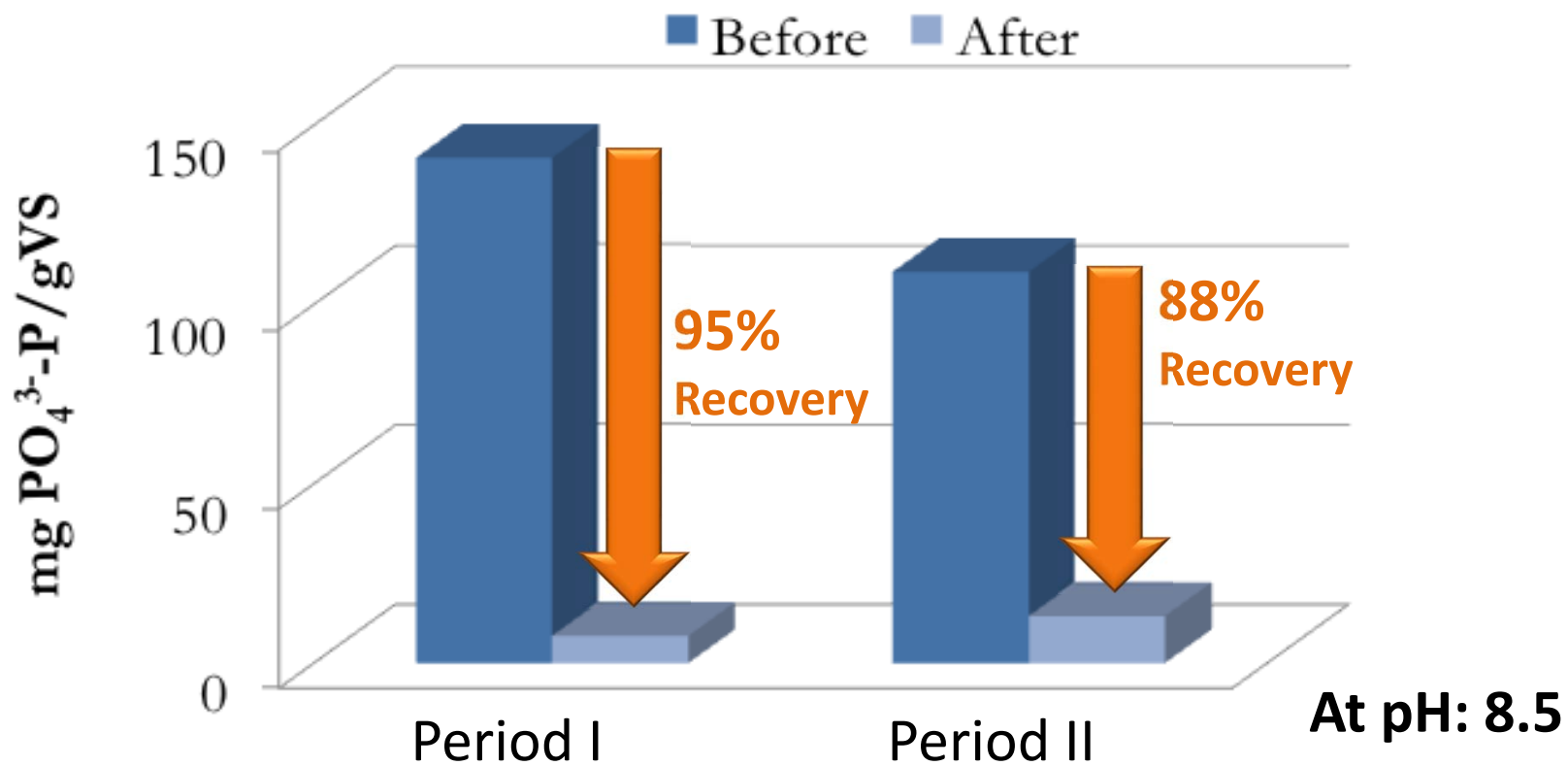
# Sequencing Batch Fermentation Reactor (SBFR): SCFA and propionic acid (HPr) production



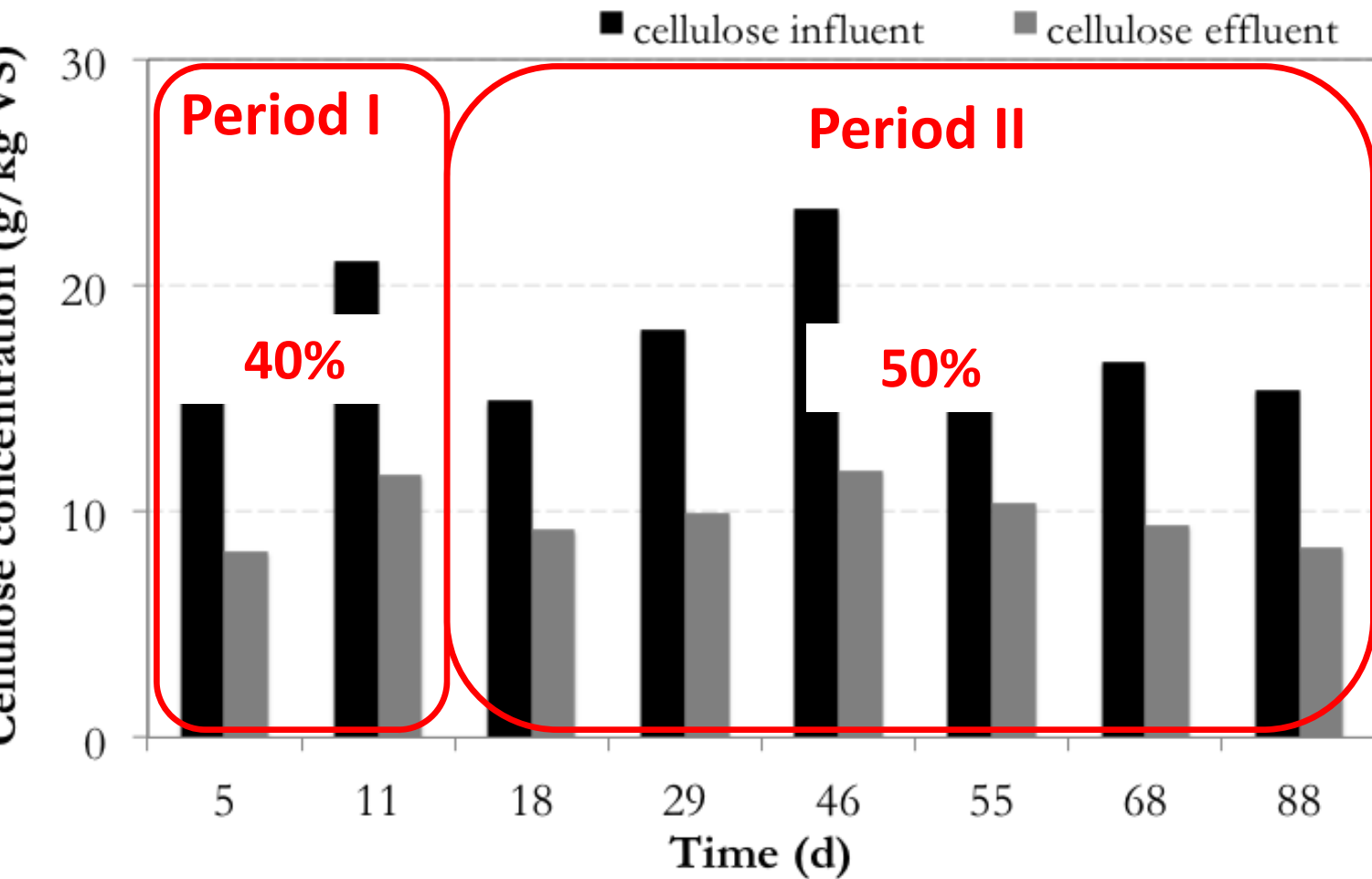
Period	Cellulosic Sludge content (%)	Influent pH
I	34.6	6.2 (Uncontrolled)
II	33.9	8.0



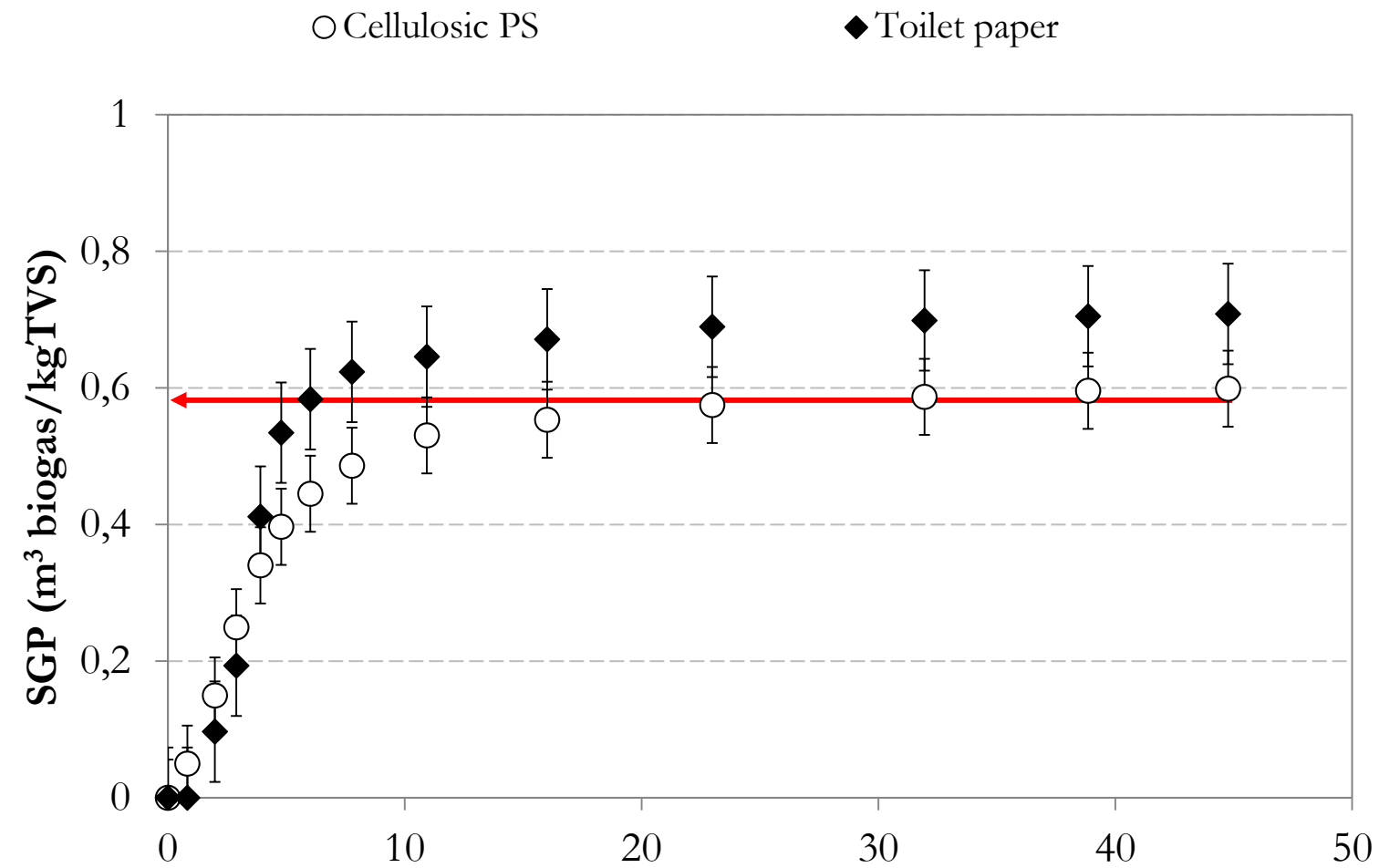
# Recovery during cellulosic sludge fermentation



# Cellulose biodegradation



# Specific gas production of the cellulosic sludge



## Conclusions

The VFAs produced from the cellulosic sludge fermentation are low cost carbon source and intermediates for a wide range of applications

The fermentation of cellulosic sludge at **mesophilic conditions** and at **influent pH of 8** can be a suitable way to produce **high concentrations of VFAs** ( $186.8 \pm 25.6$  mg COD/g  $VS_{fed}$ );

At these fermenting conditions the main VFAs produced were **acetic and propionic acids**, with a molar ratio of 0.84 mol HPr/mol HAc;

During the fermentation high concentrations of  **$PO_4^{3-}$ -P** and  **$NH_4^+$ -N** are released, which could be **recovered** by **struvite crystallisation (up to 10-15 Struvite/kgTVS fed )**.

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**Thank you for your attention**

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