



13th IWA
Specialized Conference on
Small Water and Wastewater
Systems

5th IWA
Specialized Conference on
Resources-Oriented Sanitation

Performance of air-cathode stacked microbial fuel cells systems for wastewater treatment and electricity production

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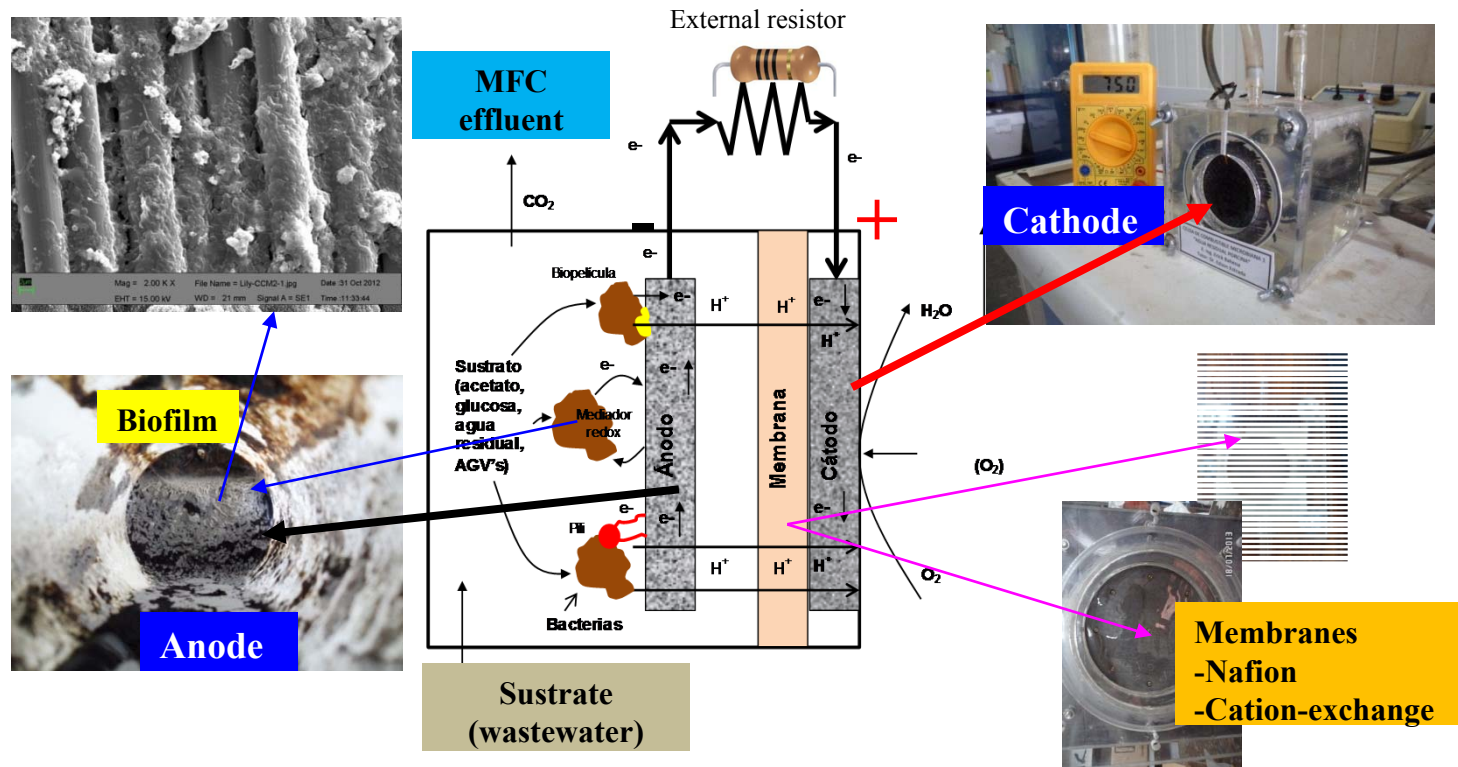
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Microbial Fuel Cells (MFCs)



The MFCs are bio-electrochemical systems, capable of generating electricity from the oxidation of organic matter from wastewater treatment and simultaneously the contaminant removal.

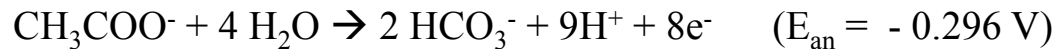


Advantages of MFC over current energy generating technologies from organics

- * Ambient temperature
- * No gas treatment
- * No energy input for aeration
- * High conversion efficiency
- * Application for wide locations and diverse fuels

Principles of MFC

- Biochemical degradation – microorganism growth : Oxidation-reduction
Electrochemistry, Catalyst reaction
Mass transport, Mixing of substrate
- @ anode, acetate (**substrate**) is oxidized by **bacteria** (catalyst), mixed culture.



Electrons produced are transferred to the anode by mediator/shuttles, direct membrane, or nano-wire.

- Electrons** flow to cathode through a conductive material containing a resistor or operated under a load to produce **electricity**.
- Protons** diffuse from anode to cathode through cation exchange membrane.
- @ cathode, the electrons are combined with proton and **oxygen** or chemical oxidizer. Diverse bacterial community is working as a catalyst.



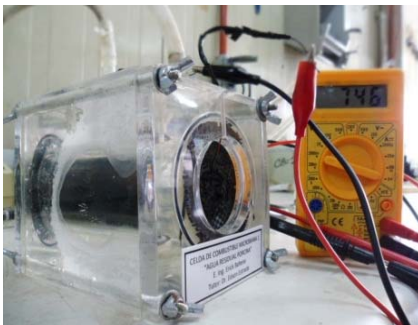
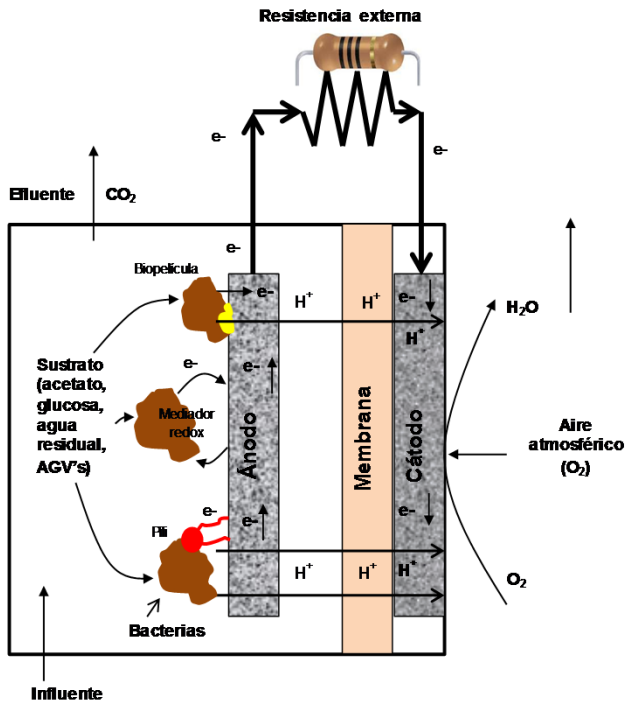
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2.2 kW/h.m³ considering that the energy of glucose molecule containing **4.4 kW/h.kg COD** (chemical oxygen demand (COD) of 500 mg/L)

Voltage productions by MFCs

Single chamber air-cathode MFC



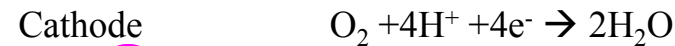
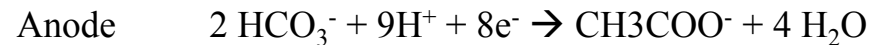
- Standard electrode potential, at SC (25 °C, 1 atm, 1 M) = zero, relative to normal hydrogen electrode (NHE)
- **Maximum attainable cell voltage** can be calculated by,

$$E_{emf} = E_{cat} - E_{an}$$

According to **Nernst equation**.

$$E_{emf} = E_{emf}^0 - \frac{RT}{nF} \ln \left(\frac{[C]^{v_C} [D]^{v_D}}{[A]^{v_A} [B]^{v_B}} \right)$$

Acetate oxidized at anode, oxygen used as e-acceptor at cathode



$$E_{an} = E_{an}^0 - \frac{RT}{8F} \ln \left(\frac{[\text{CH}_3\text{COO}^-]}{[\text{HCO}_3^-]^2 [\text{H}^+]^9} \right) = -0.296 \text{ V}$$

$$E_{cat} = E_{cat}^0 - \frac{RT}{4F} \ln \left(\frac{1}{p\text{O}_2 [\text{H}^+]^4} \right) = 0.805 \text{ V}$$

Oxidation-reduction potentials

$$E_{emf} = E_{cat} - E_{an} = 0.805 - (-0.296) = 1.101 \text{ V}$$

Maximum MFC voltage

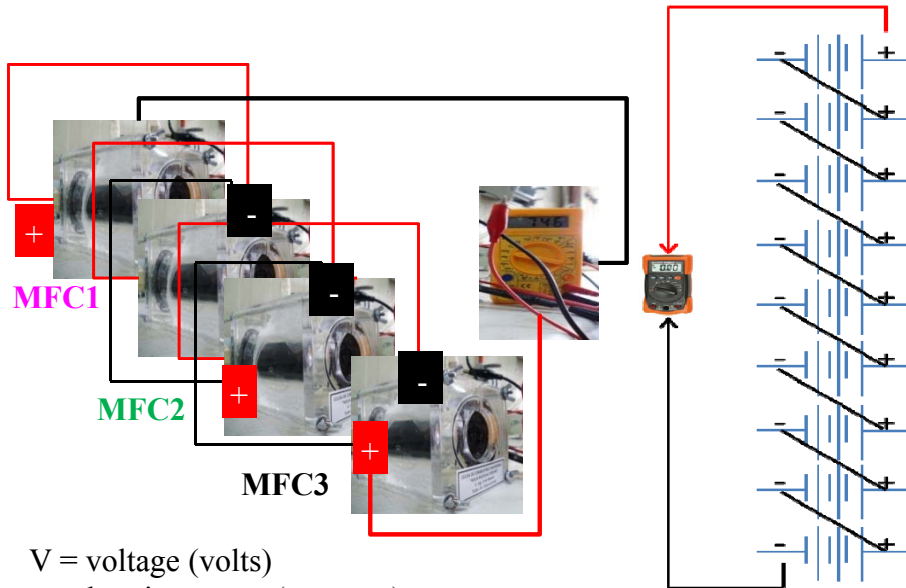
Theoretical = **1.1 V**

Open circuit mode (OCV) = **0.6-0.8 V (without current)**

Real voltage during current generation < **0.62 V**

Stacked MFC (Multi-electrode MFCs)

A single cell delivering about 0.2-0.8 V (too low for most applications). Just like batteries, individual cells are stacked to achieve a higher voltage and power. This assembly of cells is called a cell stack, multi-electrode or just a stack.



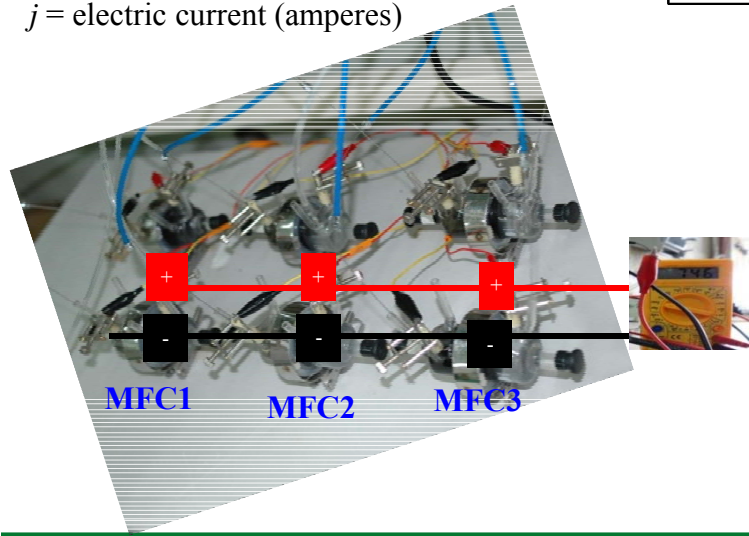
Series connections

Increasing voltage

$$\text{Stacked MFC}_{V_t} = \text{MFC1}_{V_1} + \text{MFC2}_{V_2} + \text{MFC3}_{V_3} \dots \text{MFCn} + I_{V_{n+1}}$$

The power density and voltages can be increased when MFCs are stacked in series or in parallel.

V = voltage (volts)
j = electric current (amperes)



Parallel connections

Increasing electric current

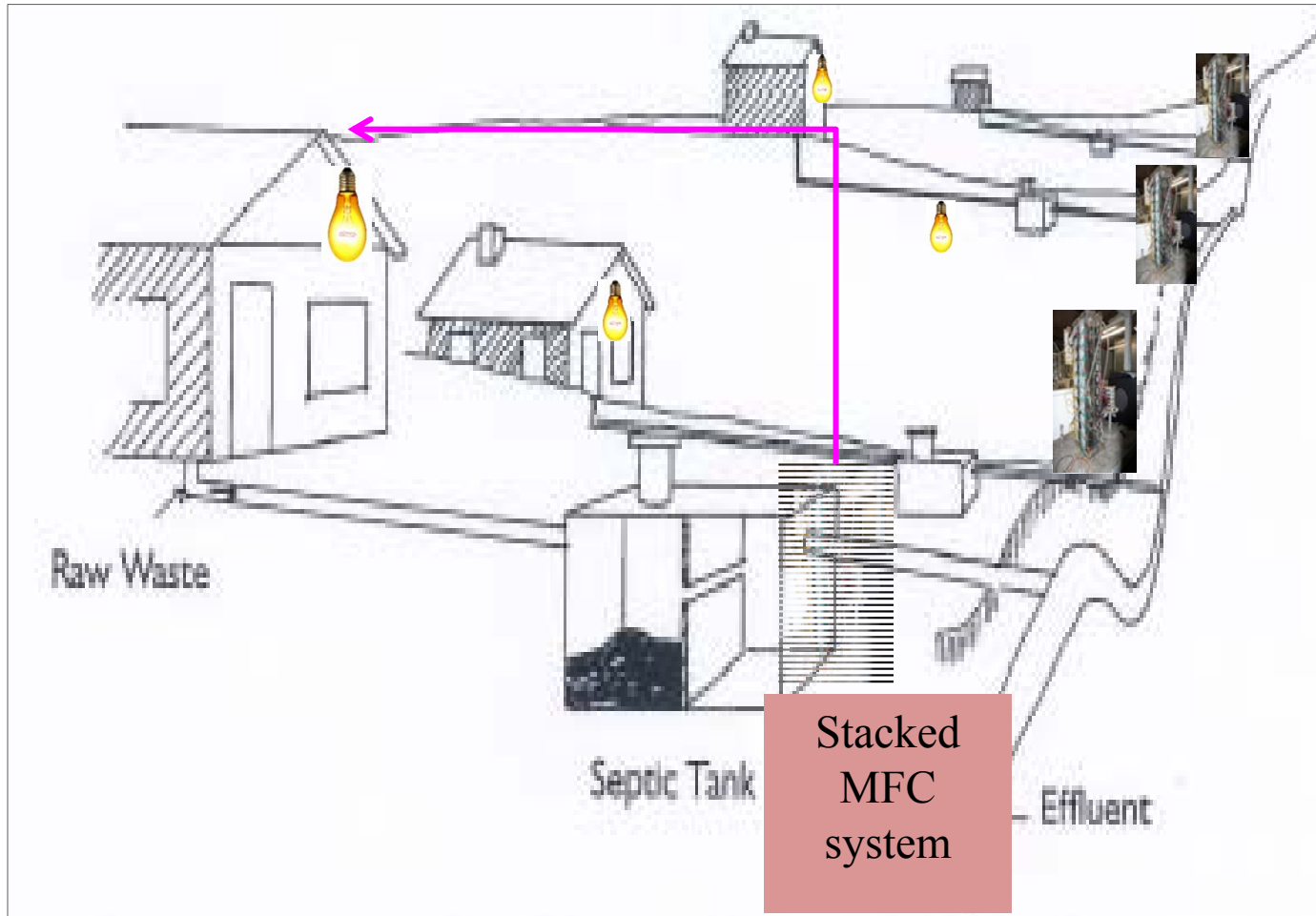
$$(\text{Stacked MFC}_j = \text{MFC1}_{j_1} + \text{MFC2}_{j_2} + \text{MFC3}_{j_3} \dots \text{MFCn} + I_{j_{n+1}})$$

The schematic shows three cells connected in parallel between a common top positive rail and a common bottom negative rail. A voltmeter is connected across the parallel arrangement to measure the total current.

Decentralized wastewater treatment system



MFCs as alternative for wastewater treatment.



New Process-
Microbial
Electrochemical
Technologies

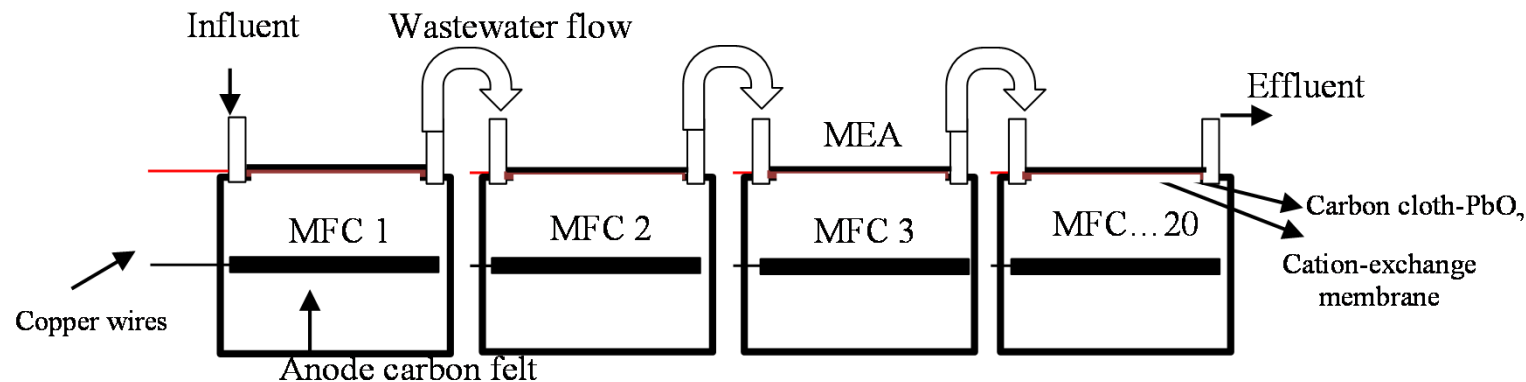
- * Removal organic matter
- * Electricity direct

Objective

The main objective of this study was to evaluate the performance of two air-cathode stacked MFC systems at different HRT (3, 1 and 0.5 d) during wastewater treatment and electricity production.

Architecture of stacked MFCs

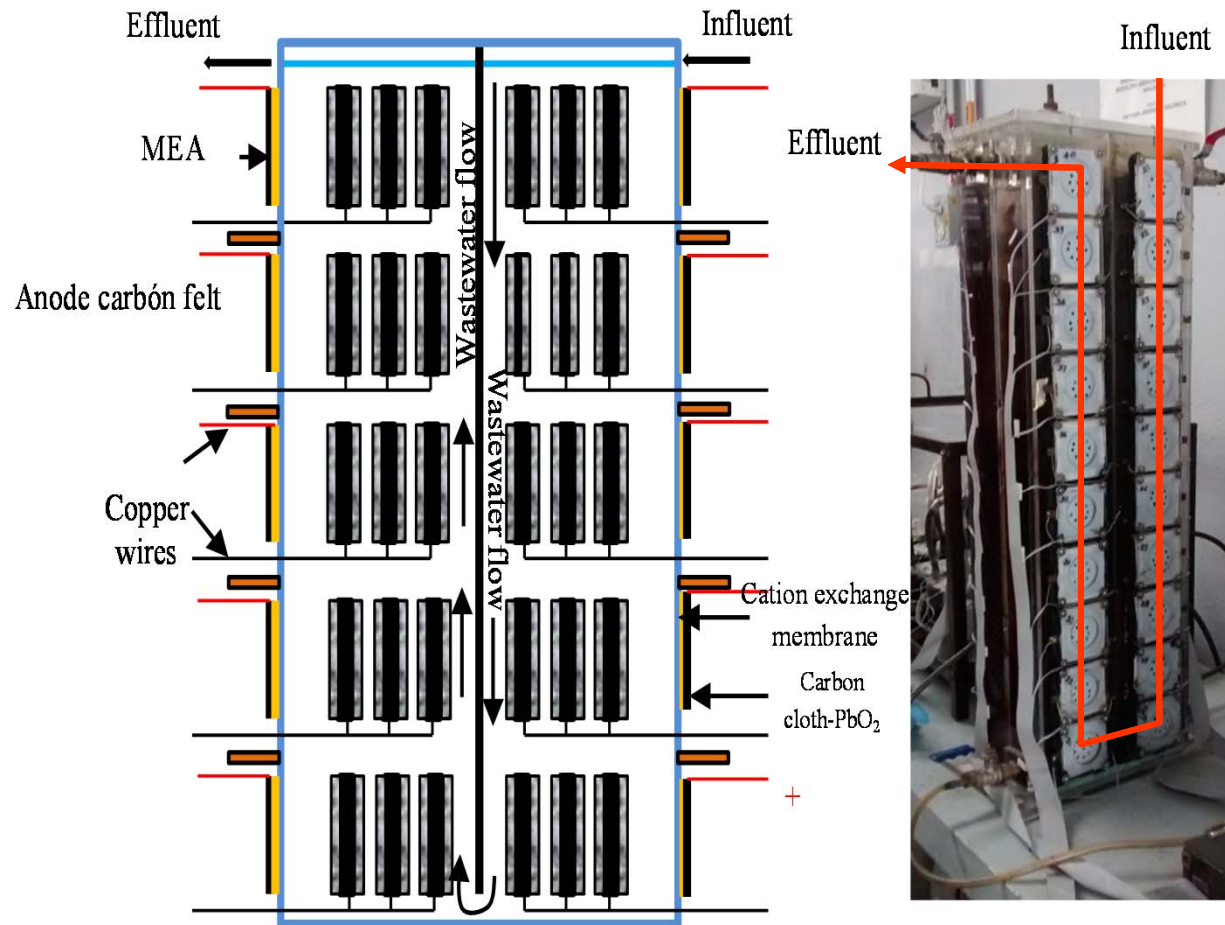
Stacked MFC 1 (un-shared reactor)-shared analyte



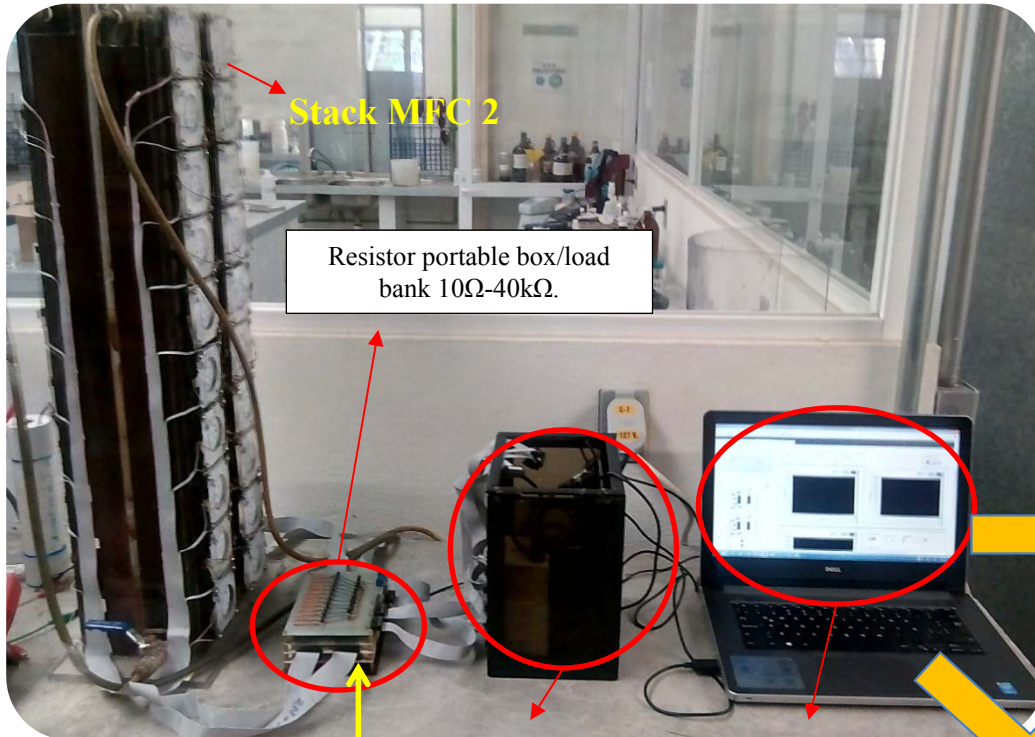
Anode and **cathode**
superficial area of 0.0036 m^2

- 20 individual MFC unit (800 mL MFC); Total volume 16 L.
- Single chamber air-cathode MFC
- Stacked MFC 1 was fed in continuous flow cascade mode (flow was transported through of the each MFC compartment)

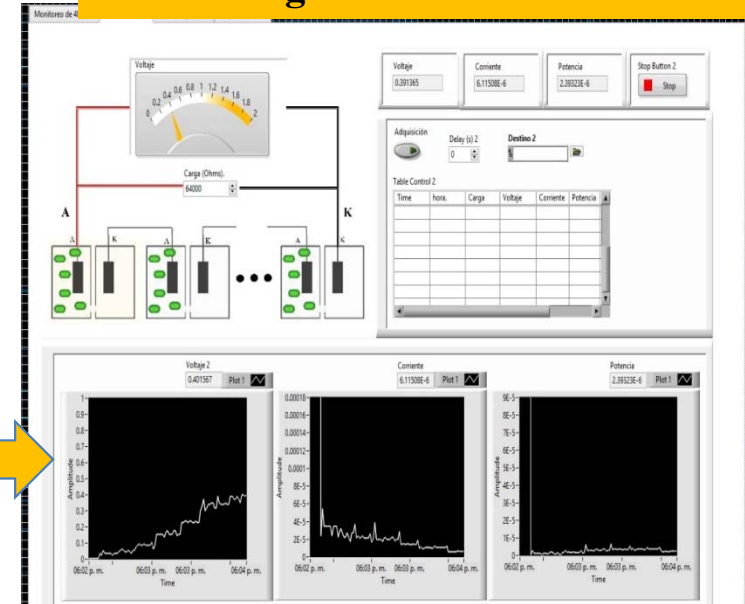
Stacked MFC 2 (shared reactor)-shared anolyte



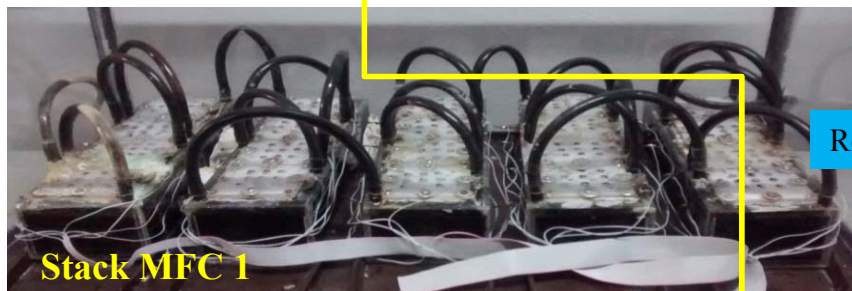
- 40 MFC unit into shared reactor.
- Total volume 16 L.
- Single chamber air-cathode MFC.
- Stacked MFC 2 was fed in continuous flow (not separator between in each cell was used).



Monitoring in series connection



Monitoring in individual MFC unit



Resistor change switch

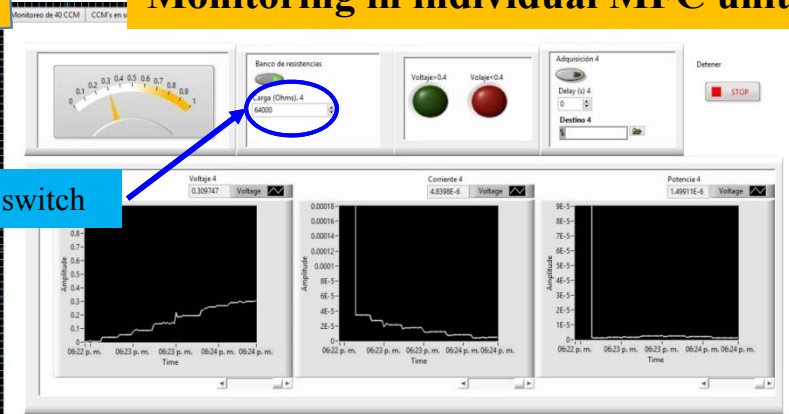


Table 1. Operational conditions

	Open Circuit Mode	Closed Circuit Mode (1,000 Ω)
Biomass acclimation	HRT 10 d	-
HRT (d)	3, 1 and 0.5	3, 1 and 0.5

INDIVIDUAL MFC UNIT SERIES CONNECTION



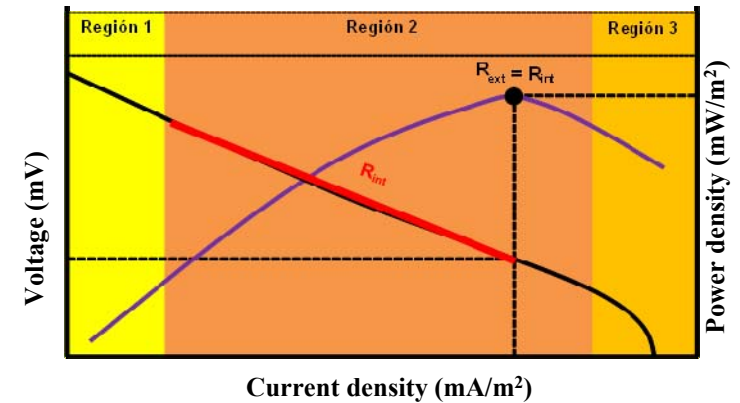
$$P = V * I = \frac{V^2}{R_{ext}} = I^2 * R_{ext}$$

$$P^* = \frac{V^2}{aR_{ext}} = \frac{P}{a}$$

$$I = \frac{dq}{dt} = \frac{V}{R_{ext}}$$

$$J = \frac{q}{at} = \frac{1}{a} \left(\frac{q}{t} \right) = \frac{I}{a}$$

Polarization curves



COD, TN, and TP were measured using standard methods (APHA, 2005).

Characteristic of municipal wastewater

COD = 209 ± 41 mg/L.

Total Nitrogen (TN) = 38 ± 11 mg/L.

Total Phosphorus (TP) = 15 ± 3 mg/L.

pH = 6.9-7.5

Total Suspended Solids (TSS) = 200 mg/L

Residential housing



Results

	Stacked MFC 1 (HRT 3, 1 and 0.5 d)	Stacked MFC 2 (HRT 3, 1 and 0.5 d)
Individual MFC unit (OCV)	2-440 mV	263-600 mV
Series connection (OCV)	20 cells 580 ±65 mV	40 cells 540 ±35 mV
Individual MFC unit (1,000 ohms)	6-50 mV	1-30 mV
Series connection (1,000 ohms)	46 ±28 mV	30 ±8 mV

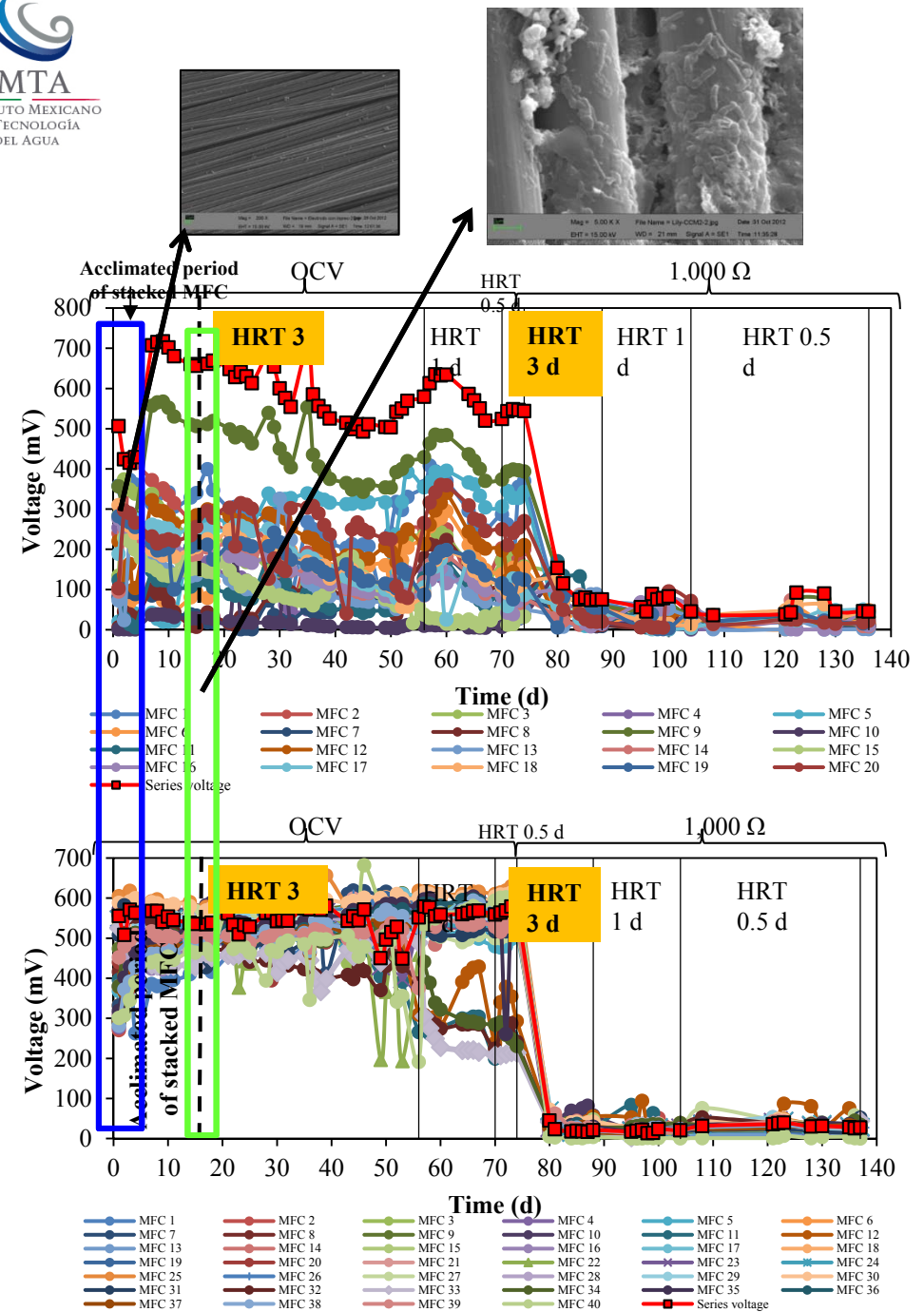
$$V_{total} = V_{MFC1} + V_{MFC2} + V_{MFCn+1}$$

voltage dropped phenomenon

Same analyte
Stacked MFC 1 and 2

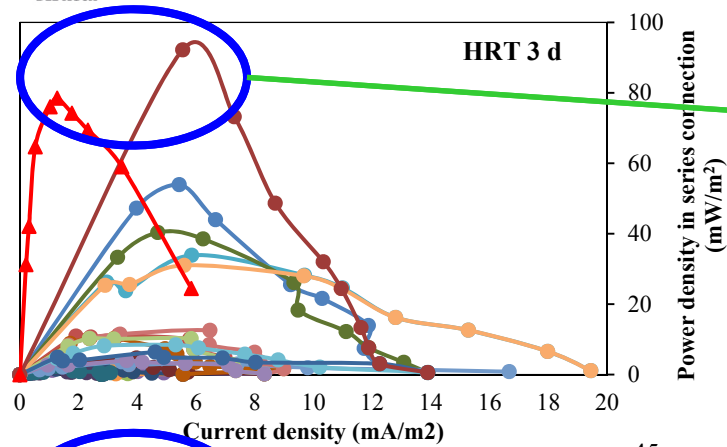
Shared reactor
(Stacked MFC 2)

ionic cross-conduction between units

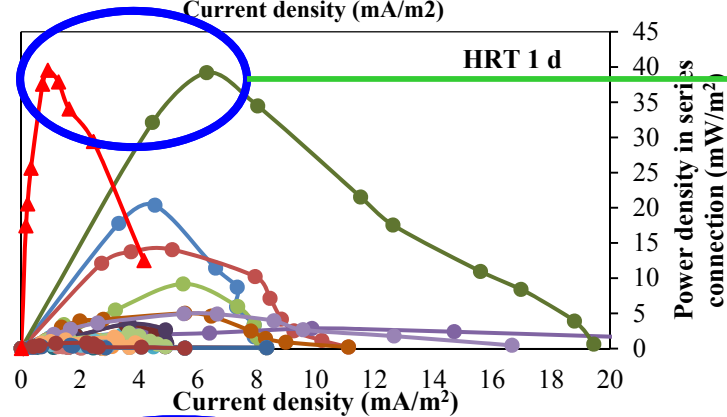


Stacked MFC 1

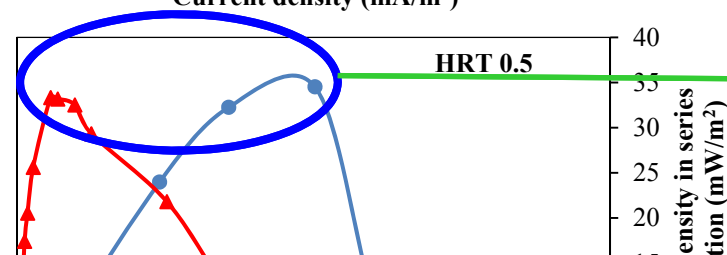
High Power at HRT 3 d



-Individual cell 20 Power = 1,107 mW/m²
-Series connection Power = 79 mW/m²



-Individual cell 9 Power = 870 mW/m²
-Series connection Power = 40 mW/m²



-Individual cell 1 Power = 430 mW/m²
-Series connection Power = 33 mW/m²

Linked MFC 2

Actual cell 11 Power = 472 mW/m²

Actual cell 34 Power = 292 mW/m²

Actual cell 34 Power = 275 mW/m²

connection

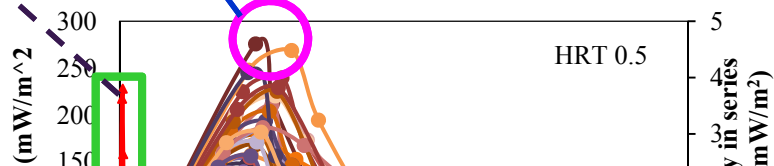
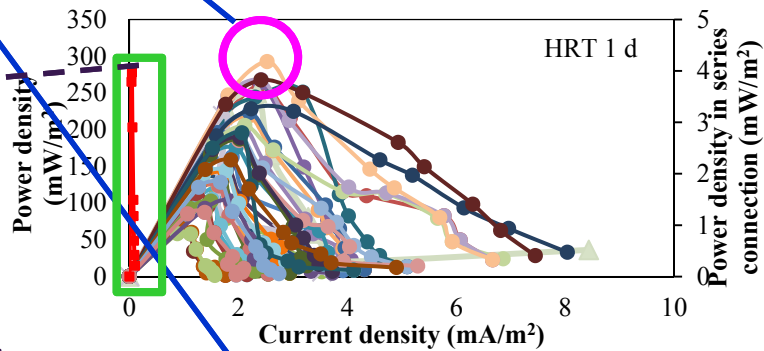
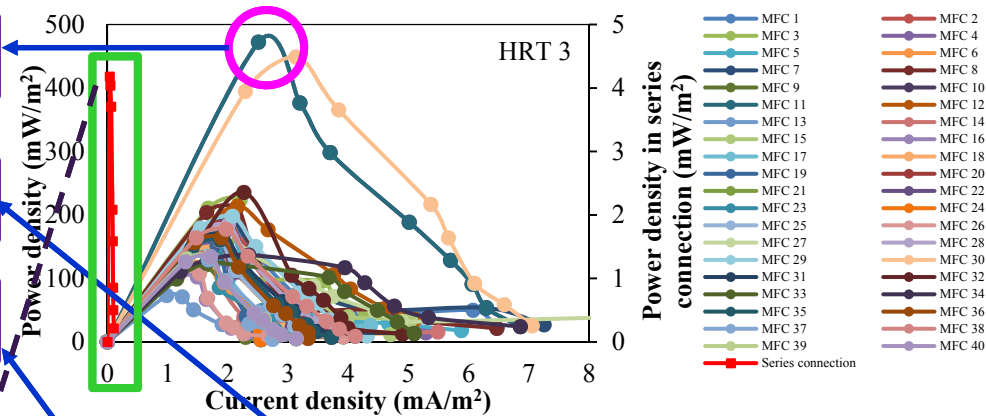
3.8-4.2 mW/m²

Current density < 0.1 mA/m²

ohmic and activation losses

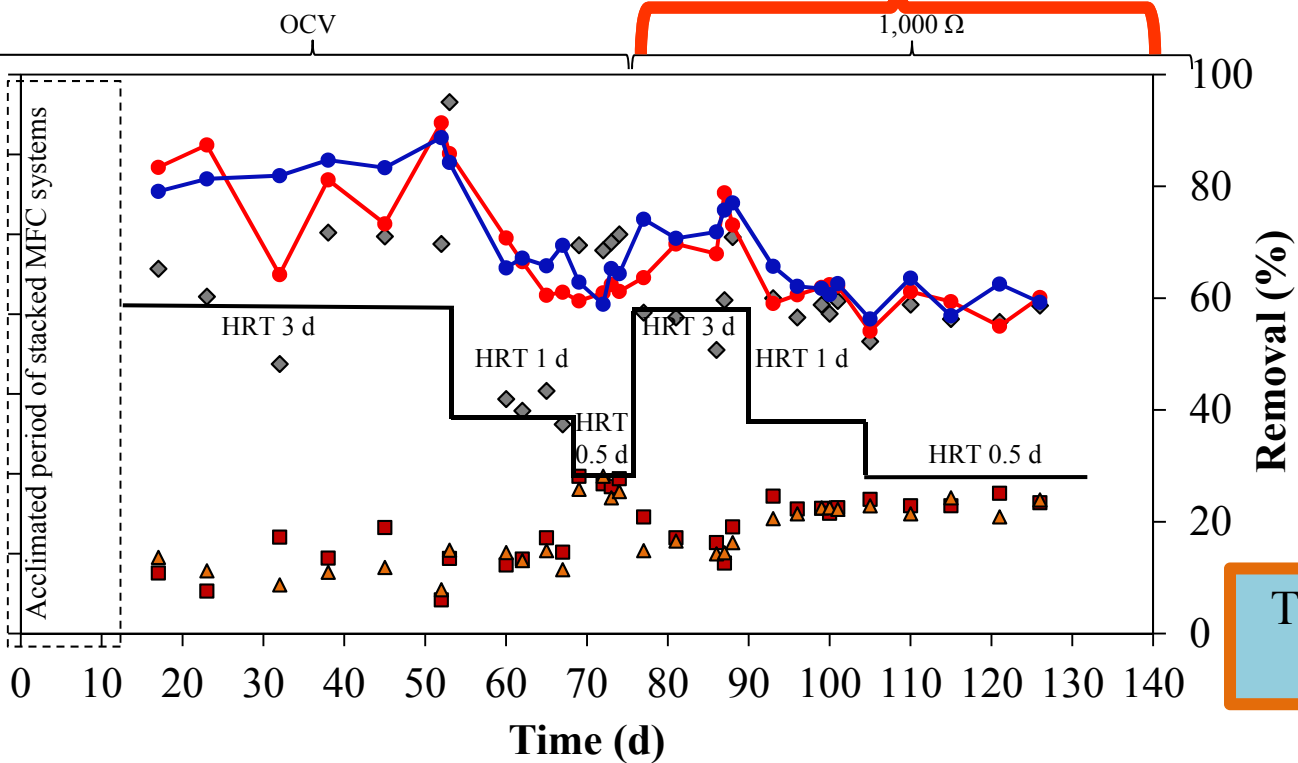


High Power at HRT 3 d



Removal COD from Stacked MFC 1 and 2 at different HRT (3, 1 and 0.5 d)

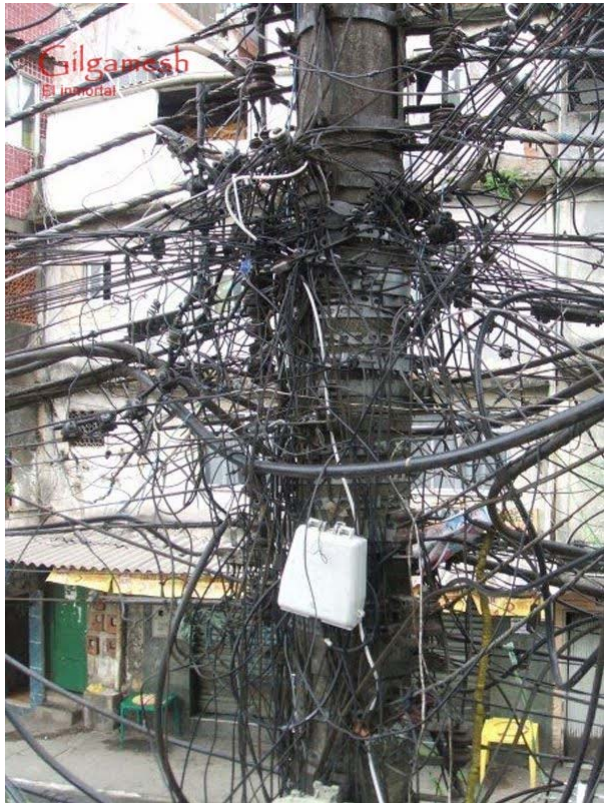
At closed circuit mode, COD removal were decreased



TN and TP removal were < 47 %

Conclusions

- The two stacked MFC systems tested were not effective for power production in series connection under OCV and closed circuit mode.
- Voltages dropped (voltages reversal) in the two systems were generated due to the architecture of the systems (shared reactor) and the same anolyte in all MFC units.
- The maximum power density in series connection of the stacked MFC 1 was $79 \pm 0.65 \text{ mW/m}^2$ (current density of $1.3 \pm 0.4 \text{ mA/m}^2$). For the individual MFC unit (no-series connection), the maximum power density was $1,106 \pm 1.2 \text{ mW/m}^2$ (current density of $5.5 \pm 0.6 \text{ mA/m}^2$) at HRT of 3 d.
- Power production in stacked MFC 2 ($4.2 \pm 0.6 \text{ mW/m}^2$) and the current density ($0.04 \pm 0.006 \text{ mA/m}^2$) were lower compared with the power generated from stacked MFC 1.
- The results showed that the COD removal (up to 84%) were increased when the HRT were increased from 0.5 to 3 d.



Thank you for your attention

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