



13th IWA Specialized Conference
on Small Water and Wastewater
Systems

5th IWA Specialized Conference
on Resources-Oriented Sanitation

**Topic: Advances in wastewater treatment by combined
microbial fuel cell-membrane bioreactor (MFC-MBR)**

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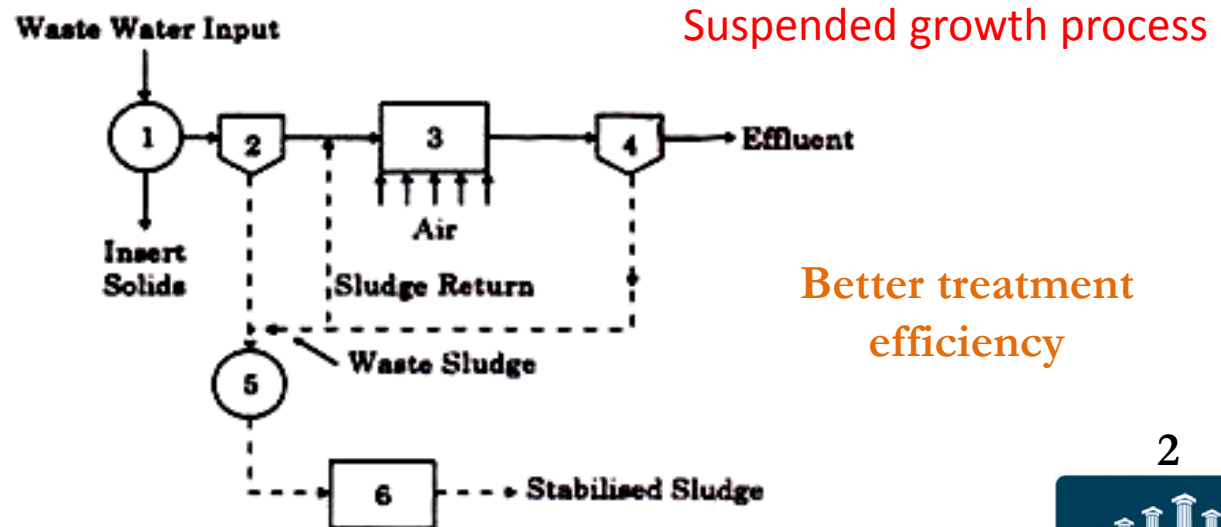
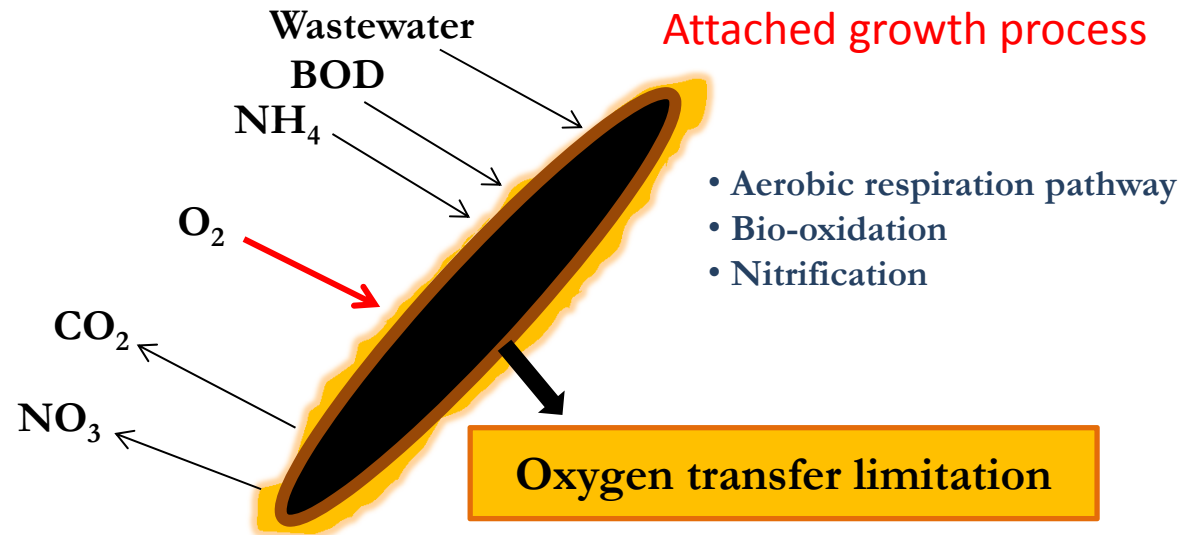
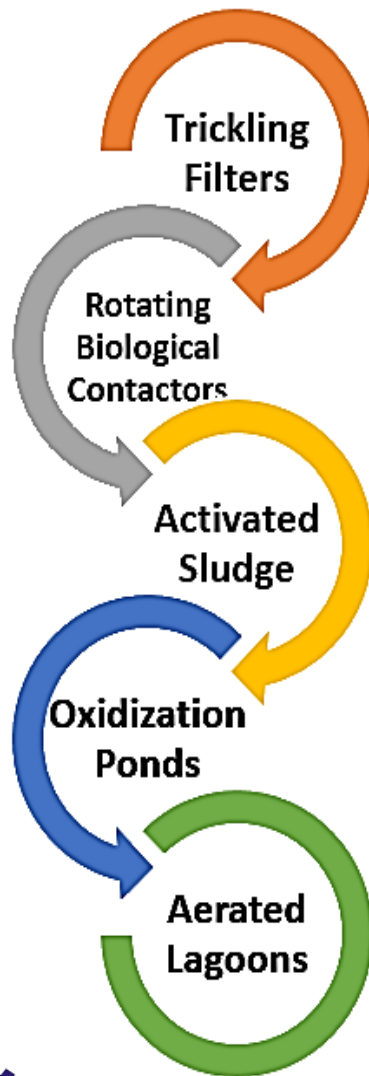
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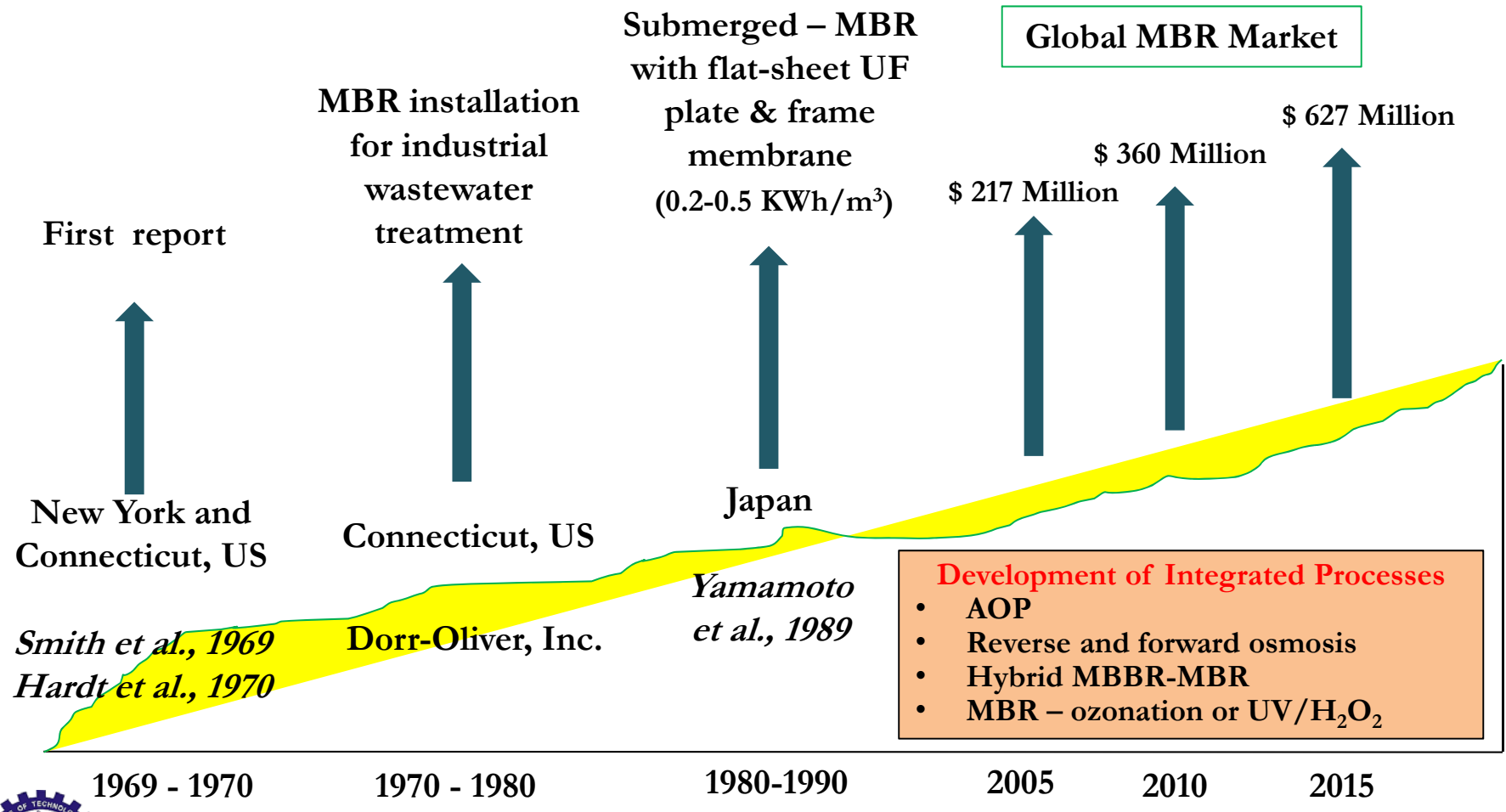


Commonly Used Aerobic Biological Wastewater Treatment Processes



Membrane bioreactor (MBR) Technology

Biological – ASP + Membrane Filtration



MBR technology involves high energy-consuming process

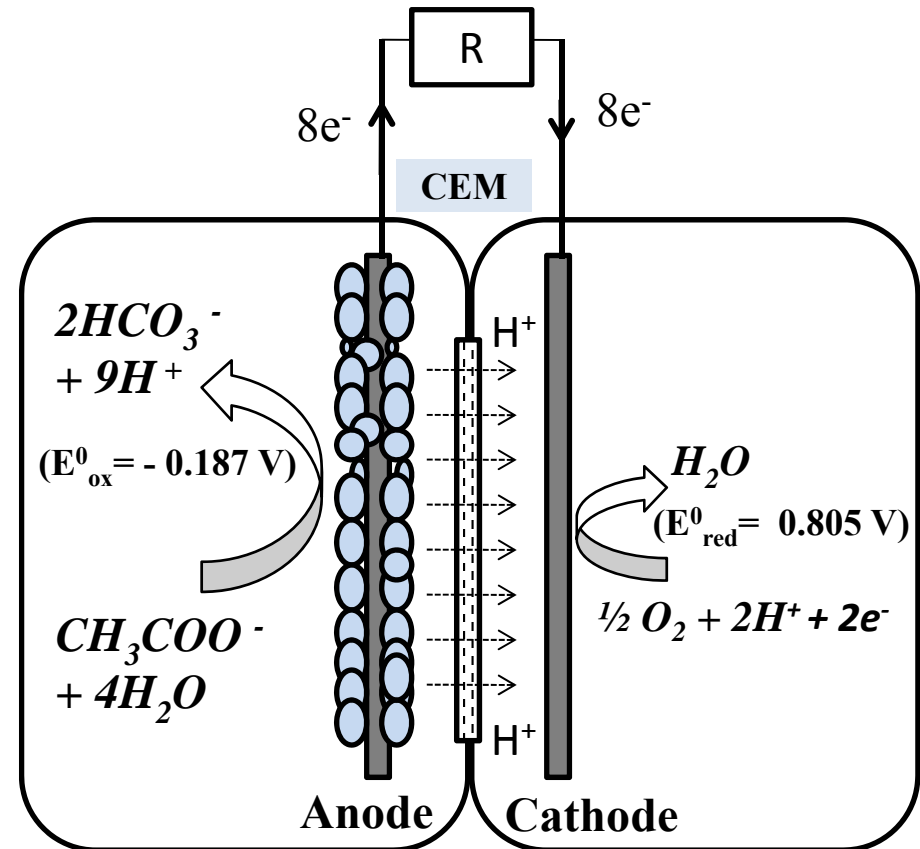
Energy consumption of MBR can be lowered by integrating it with
Microbial Fuel Cell (MFC) technology

Conversion of bio-chemical
energy to electrical energy



Bio-electricity – An Alternative
and Clean Energy

- How much electrical energy can be generated?
- Can we provide an efficient treatment?
- Can low-cost sustainable development of MFC-MBR technology be achieved?



Microbial Fuel Cell (MFC)

Recent advances in MFC-MBR processes

Completely
anaerobic process

- **Electrochemical – MBR**
- **Up-flow integrated air-cathode MFC-MBR**

Wang, 2013 – *Sci. Rep.*

Ge, 2013 – *J. Chem. Technol. Biotechnol.*

Wang, 2012 – *Appl. Energy*

Wang, 2013 – *Chem. Eng. Technol.*

Lower energy consumption

Combination of
anaerobic –
aerobic process

MFC – Biocathode MBR

Wang, 2014 – *Bioresour. Technol.*

Consumption of electrical energy
to develop MFC-based biosensors



Aim of our research

Development of two-stage continuous process of combining MFC with MBR treatment technology for a highly-efficient and reliable wastewater treatment

- For treatment of organic wastewater, having COD of 3 g/l
- To achieve better treatment efficiency in terms of organic matter removal
- Recovery of high quality reusable effluent



Reactor fabrication and operating principle

MFC

Parameters	Operating conditions
Working volume	1.5 l
Electrode material	Anode: Carbon felt (untreated) Cathode: C/TiO ₂ suspension
Inoculum	Mixed anaerobic sewage sludge
Substrate	Synthetic wastewater – Sucrose as carbon source Jadhav & Ghangrekar, 2009 (<i>Bioresour. Technol.</i>)
Substrate conc.	3 g COD/l
HRT	2 days

Aerobic MBR

Parameters	Operating conditions
Working volume	1 l
MLSS	7.09 ± 0.48 g/l
F/M	0.08 kg COD/kg MLSS. day
HRT	10 h
Inoculum	Aerobic pond sediment
Substrate	MFC effluent
Membrane filtration	Hollow-fibre Polysulfone-made UF membrane (pore size 80 nm, OD 1 mm and ID 0.8 mm)
Membrane area	300 cm ² /l
Permeate flux	38 l/m ² .h

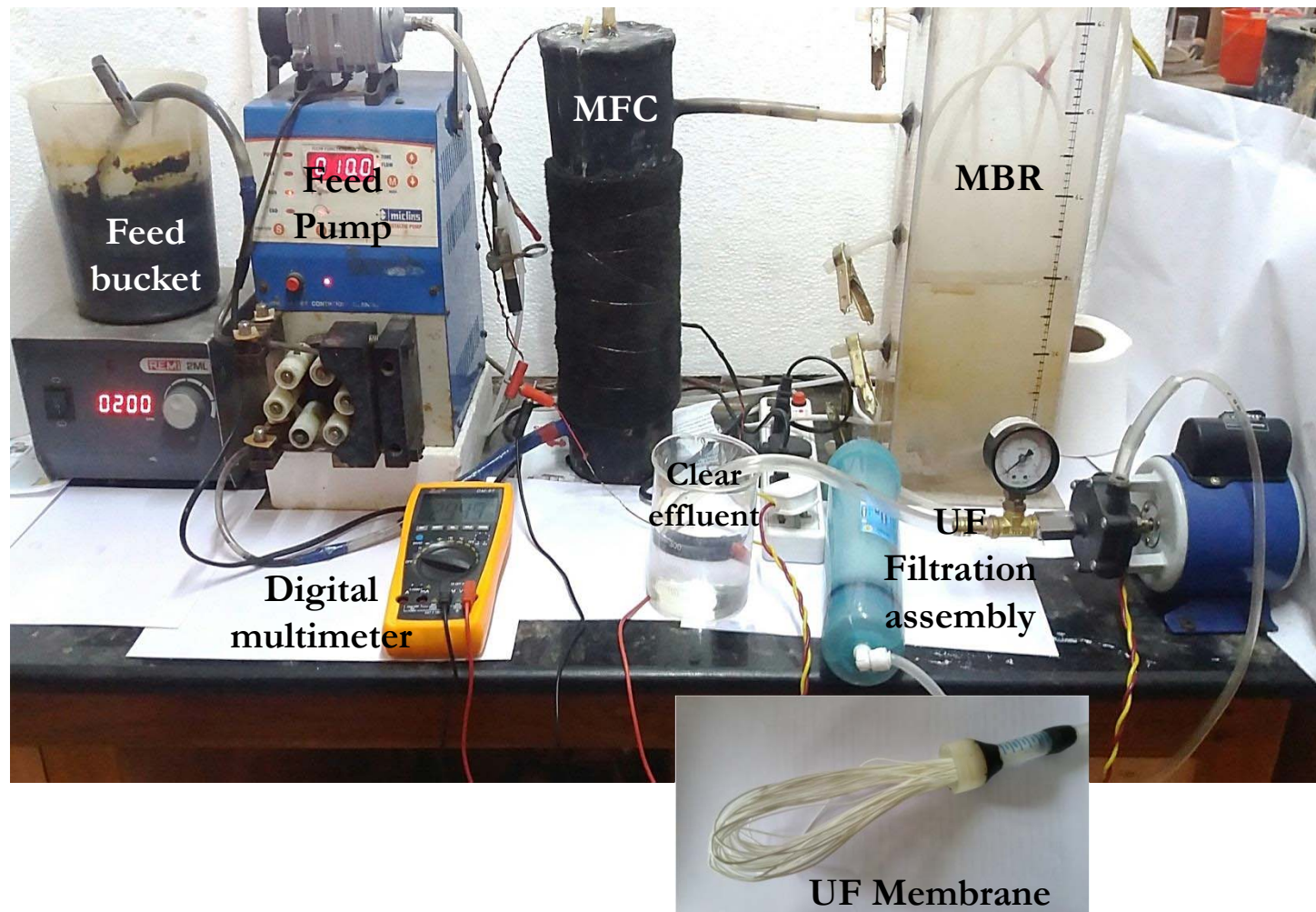
Electrochemical monitoring, polarization study and determination of coulombic efficiency (Logan, 2008 – John Wiley & Sons Inc.)

Total and soluble COD, MLSS, MLVSS, TKN and alkalinity (APHA 1998)



Bench-scale working model

Two-stage wastewater treatment process combining microbial fuel cell and aerobic membrane bioreactor –



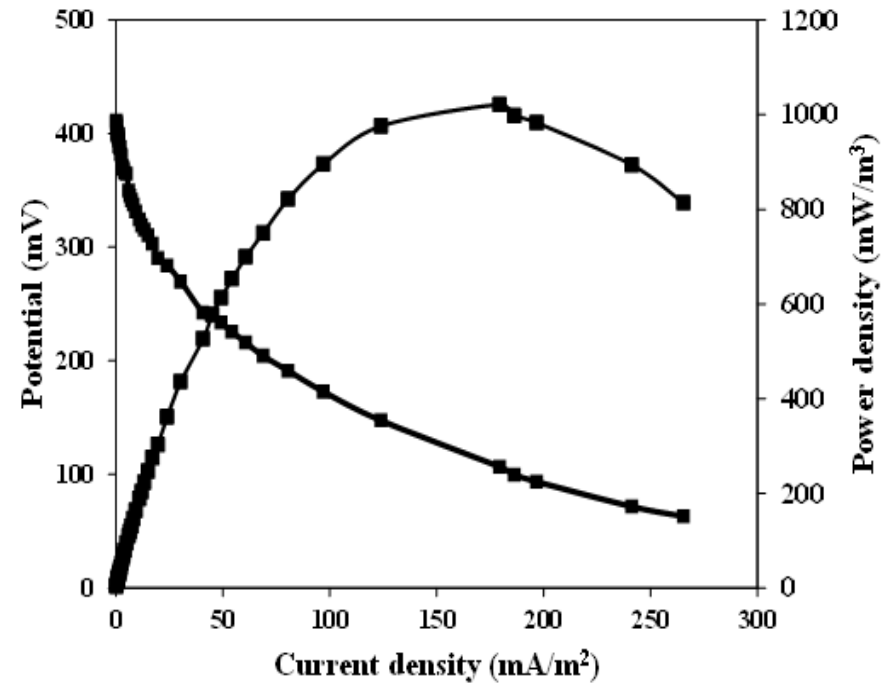
Results..

Generation of bio-electricity in MFC

Parameters	Responses
Open circuit potential	536 ± 25 mV
Working potential (100 Ω)	260 ± 12 mV
Power density	1.021 W/m ³
Internal resistance (Whole cell)	17.8 Ω
CE	4.35 %

Treatment of wastewater in MFC

The COD removal efficiency of 78.4 ± 2.14 % was observed during MFC treatment. The total COD concentration of MFC effluent was 0.71 ± 0.04 g/l.



Polarization and power curves for MFC

Treatment of MFC-effluent in MBR with submerged UF membrane

Parameters	Wastewater (MFC reactor influent)	MFC reactor effluent	MBR effluent (Permeate)
Total COD	3.02 (0.03)	0.71 (0.04)	-
Soluble COD	2.65 (0.02)	0.59 (0.03)	0.04 (0.003)
TKN	0.31 (0.05)	0.147 (0.02)	0.010
TS	3.67 (0.05)	5.09 (0.08)	-
TSS	-	-	< 0.005
MLVSS	NA	0.9 (0.02)	ND
pH	7.53 (0.14)	7.31 (0.11)	7.4 (0.1)

^a All units are in g/L, except pH; numbers in the parenthesis are standard deviation

NA= Not applicable; ND= Not detectable

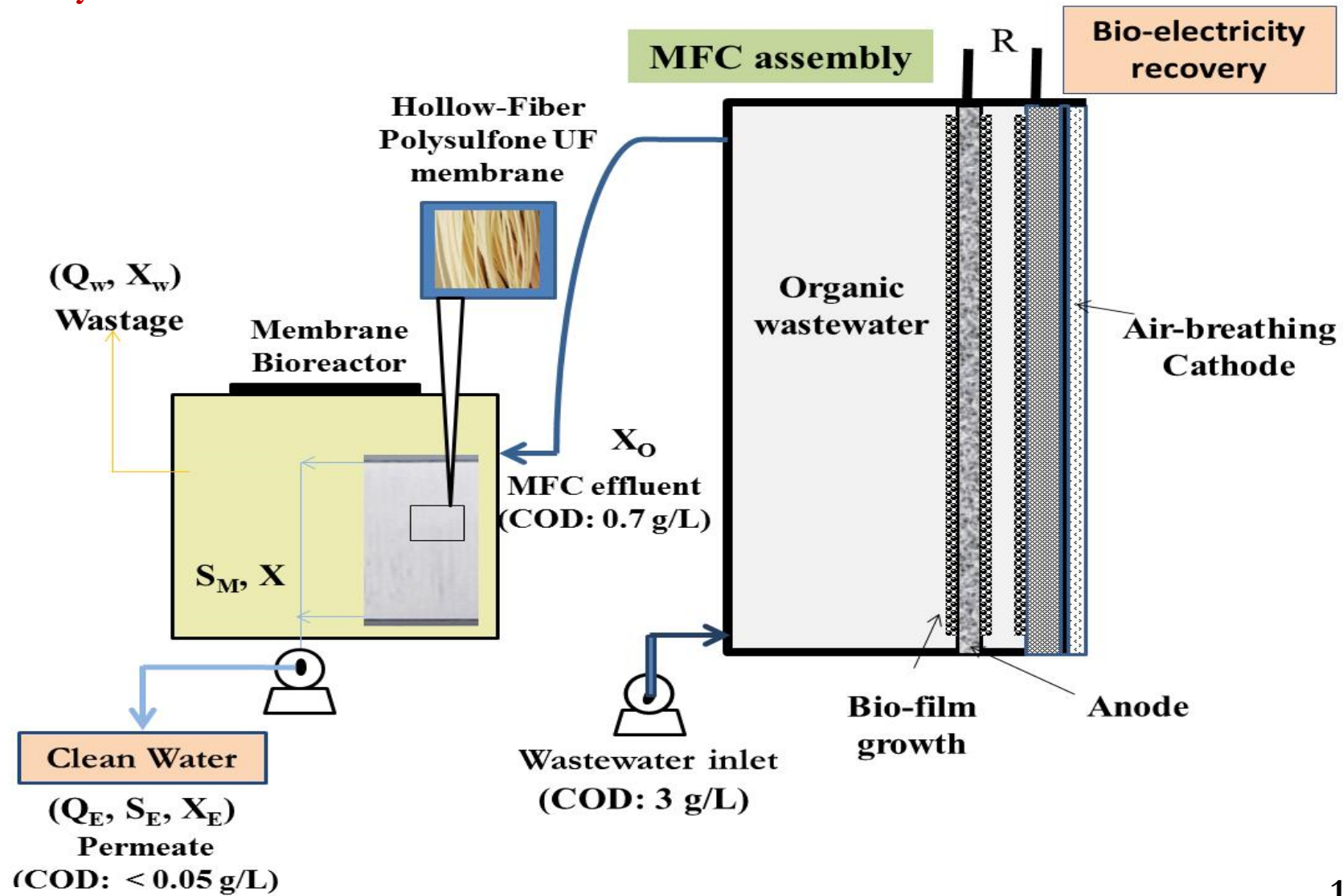
Characteristics of effluent at different stages of MFC-MBR treatment

Organic removal efficiency in combined MFC-MBR process

Soluble COD, TKN and SS removal efficiency was $98.49 \pm 0.28 \%$, $96.77 \pm 0.12 \%$ and $99.75 \pm 0.18 \%$, respectively.



Analysis of Bio-kinetic Parameters of MBR



Kinetic Equations and Results

Monod equation for biomass growth rate: $\mu = \mu_m \frac{S}{K_s + S}$

The rate of change of biomass in MBR: $V \cdot \frac{dX}{dt} = \mu XV - k_d \cdot XV - Q_w X - Q_E X_E$

At steady state condition, $dX/dt = 0$: $\mu = k_d + \frac{Q_w}{V} + \frac{Q_E}{V} \cdot \frac{X_E}{X}$

$$\text{Sludge retention time, } SRT (\theta_c) = \frac{VX}{Q_w X + Q_E X_E}$$

Hence, $\mu = k_d + \frac{1}{SRT}$

Thus, the final equation for substrate utilization: $S = \frac{K_s \left(\frac{1}{SRT} + k_d \right)}{\mu_m - \left(k_d + \frac{1}{SRT} \right)}$ The substrate balance

equation to demonstrate the expression for biomass generation in MBR:

$$X = \left[\frac{Q(S_0 - S) - S_E \cdot Q_E}{\left(k_d + \frac{1}{SRT} \right)} \right] \frac{Y}{V}$$

- The **SRT** was calculated as 15 days.
- **Endogenous decay constant (k_d)** and **sludge-yield coefficient (Y)** was calculated as 0.07 d^{-1} and $0.216 \text{ g VSS/g of COD}$, respectively.



Summary..

- How much electrical energy can be generated?

Authors	Anode	Cathode	Maximum power density (W/m^3)
Wang, 2013 (<i>Water Res.</i>)	Graphite rod	Stainless steel mesh	1.43
Ge, 2013 (<i>Sci. Rep.</i>)	Carbon brush	Carbon cloth coated with 10% Platinum (Pt)	2
Li, 2014 (<i>J. Chem. Technol. Biotechnol.</i>)	Carbon cloth	Carbon cloth coated with 10% Pt	0.15
Liu, 2014 (<i>Int. J. Hydrogen Energy</i>)	Graphite granules	Stainless steel mesh	0.15
Li, 2014 (<i>Sep. Purif. Technol.</i>)	Graphite granules	Polyester filter cloth, modified by in situ formed PANi (polyaniline)-phytic acid (PA)	0.78
This Study	Carbon felt	C/TiO₂ ink cathode	1.02



- **Can we provide an efficient treatment?**

The treated effluent generated in two-stage combined MFC-MBR process has the following characteristics:

Soluble COD: In the range of **30 – 40 mg/l**

BOD: Less than **5 mg/l**

TKN: **10 mg/l**

TSS: Less than **5 mg/l**

- **Can low-cost sustainable development of MFC-MBR technology be achieved?**

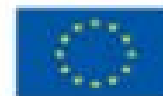
1. Generation of high quality effluent – Membrane retains most particulate matter.
2. Combined process has smaller footprint for medium-scale organic wastewater treatment.
3. Easy operation and less space is required for reactor set-up

Acknowledgement



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Thank You

