



Novel Graphene Oxide Based Photocatalyst Glass Coating for Organic Removal Under Solar Light

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15.09.2016



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Water pollutants:

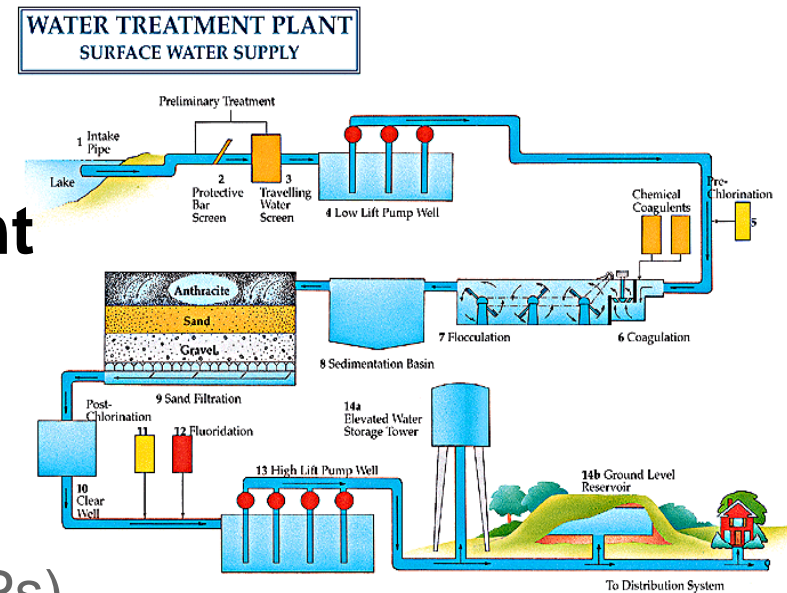
- Inorganic wastewater
- Emerging organic contaminants (EOCs)



- Microorganism

Conventional water treatment plants:

- Inadequate
- Harmful disinfection byproducts (DBPs)





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- Emerging treatment technology:

Semiconductor-mediated heterogeneous photocatalysis

- Advantages:
 - non-hazardous and economical semiconductors
 - Shorter reaction time
 - Complete mineralisation of the contaminants into harmless substances



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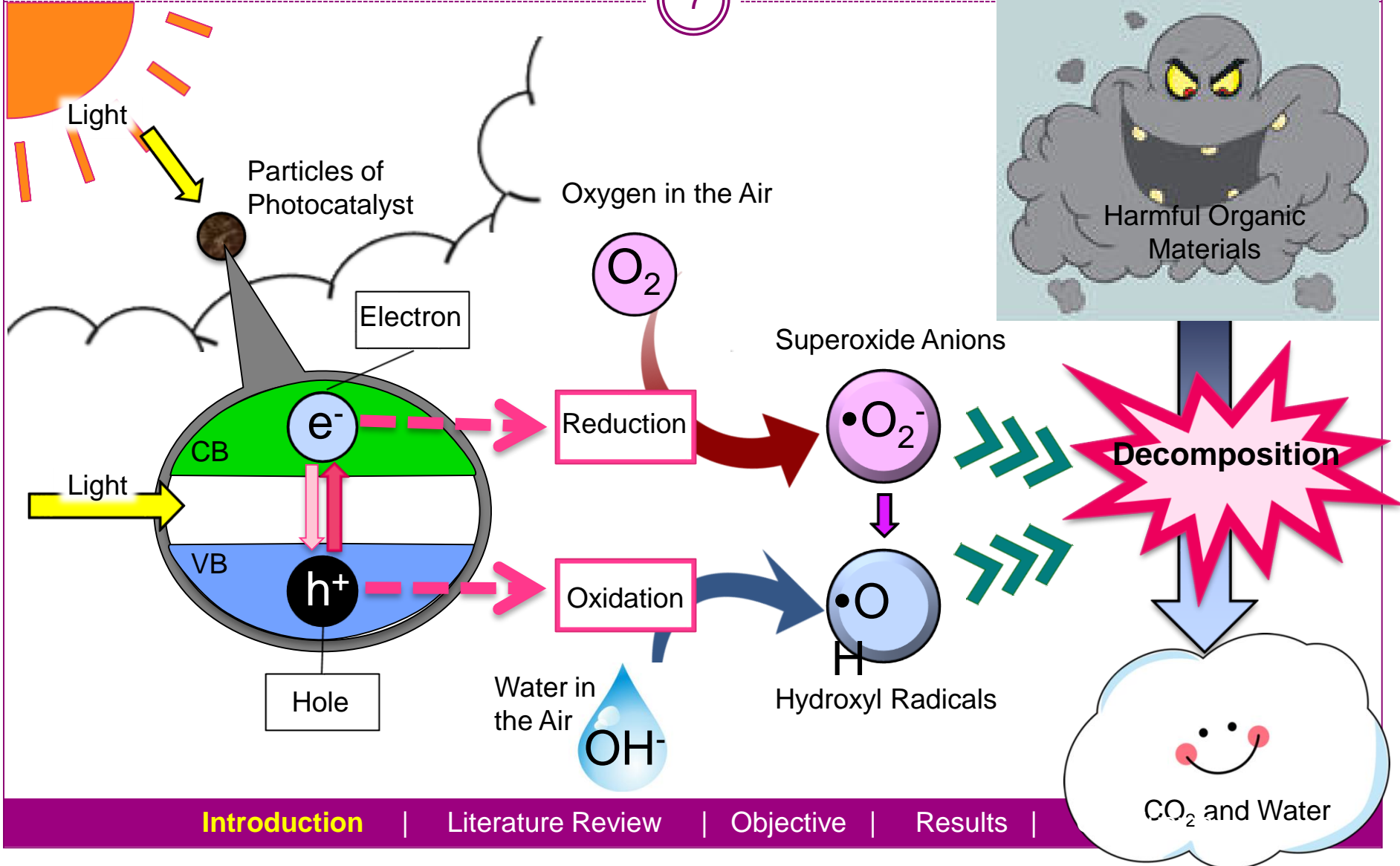


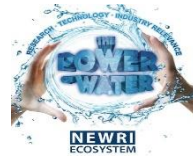
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Semiconductor-mediated heterogeneous photocatalysis

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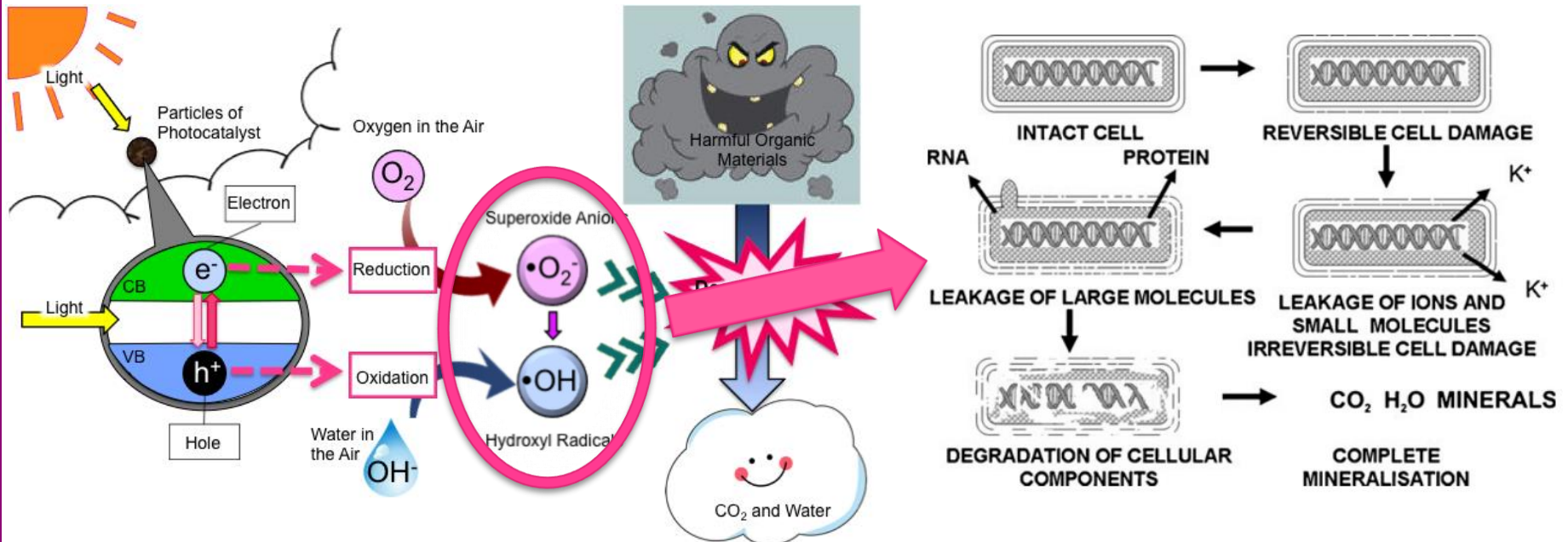




Solar Photo-disinfection

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- With the aid of the **reactive oxygen species (ROS)**, microorganisms such as bacteria and viruses could be effectively **destroyed**.





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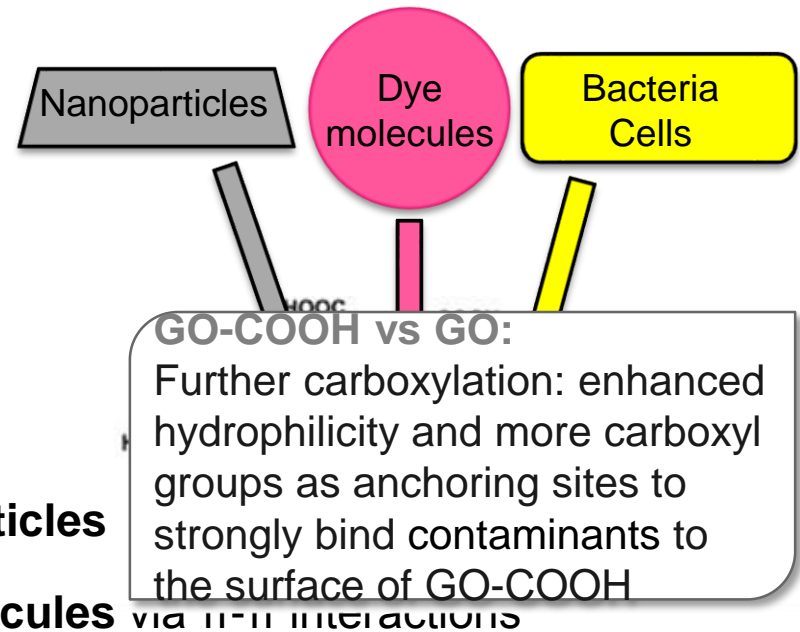
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Carboxylic acid functionalised Graphene Oxide (GO-COOH) Sheets

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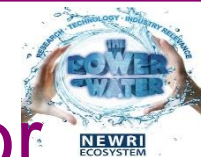
• Advantages:

- Excellent photocatalytic conductivity
- Large surface area (2600 m²/g)
- Carboxyl functional groups
 - ✦ Effective support for inorganic **nanoparticles**
 - ✦ Strong binding ability with the **dye molecules**
 - ✦ **Bacteria cells** can be assembled and adsorbed on the GO-COOH sheets



• Disadvantages:

- Aggregation as suspensions in solution



Copper Sulfide (CuS) as Semiconductor

Advantages	Limitations
<ul style="list-style-type: none">• Nontoxic• Stable under ambient conditions• Low cost• Promote photocatalysis under visible light irradiation (narrow band gap 2.1eV)	<ul style="list-style-type: none">• Low photogenerated charge transfer rate on the photocatalyst surface.• Tendency for nanoparticles aggregation• Disinfection ability not so strong





Considerations for CuS-based nanocomposite

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Glass coating increases the ability in recovery, recycling and isolating these nanoparticles

Novel copper sulfide-based nanocomposites (GO-COOH-CuS-Ag) for water remediation: Synergistic effect

GO: Suitable support and enhanced photocatalytic ability
Ag: Enhanced disinfection ability

Copper Sulfide (CuS) as Semiconductor

CuS

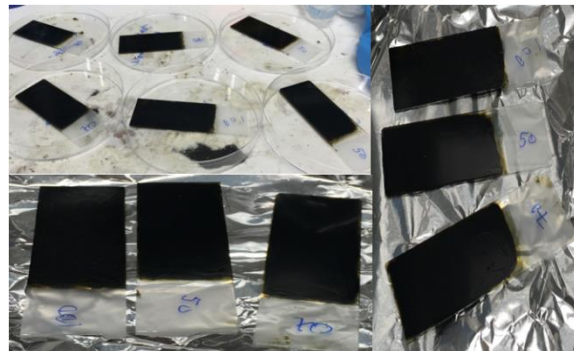
Limitations

- Nontoxic
- Transforms to a superconductor at high temperature (1.6K)
- Stable under ambient conditions
- Low cost
- Able to produce photocatalytic activity under light irradiation

photogenerated charge transfer rate

photocatalyst surface.

tendency for nanoplate aggregation





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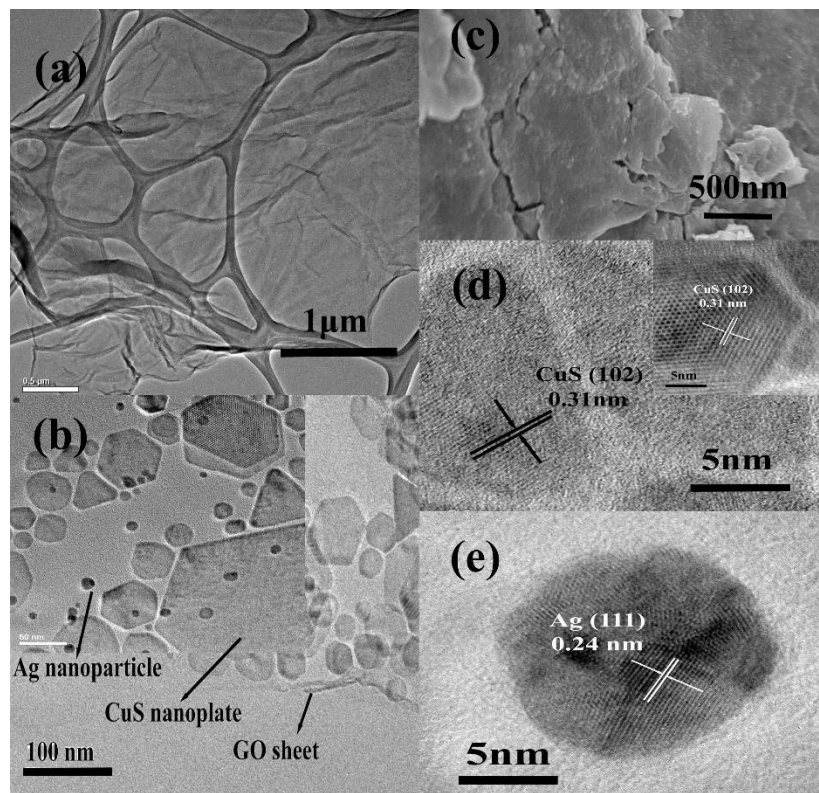


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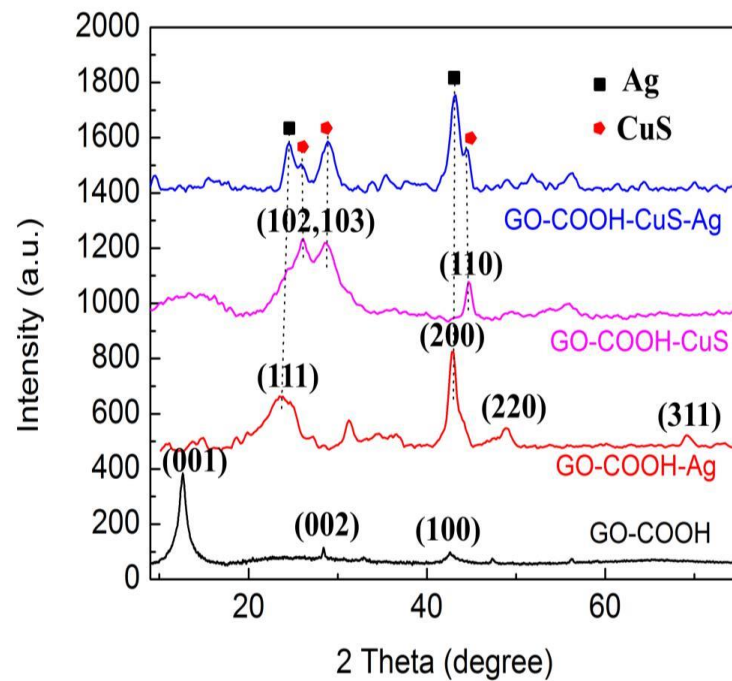


Morphological and Structural Analysis

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- TEM and FESEM images

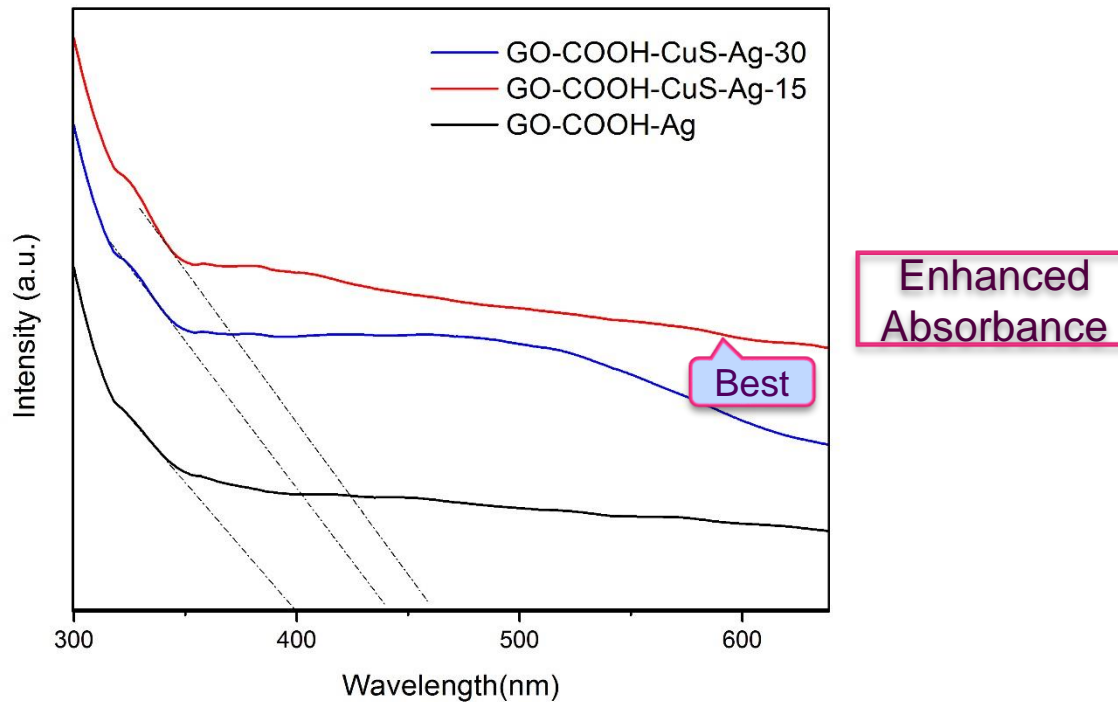


- XRD patterns



Optical Properties

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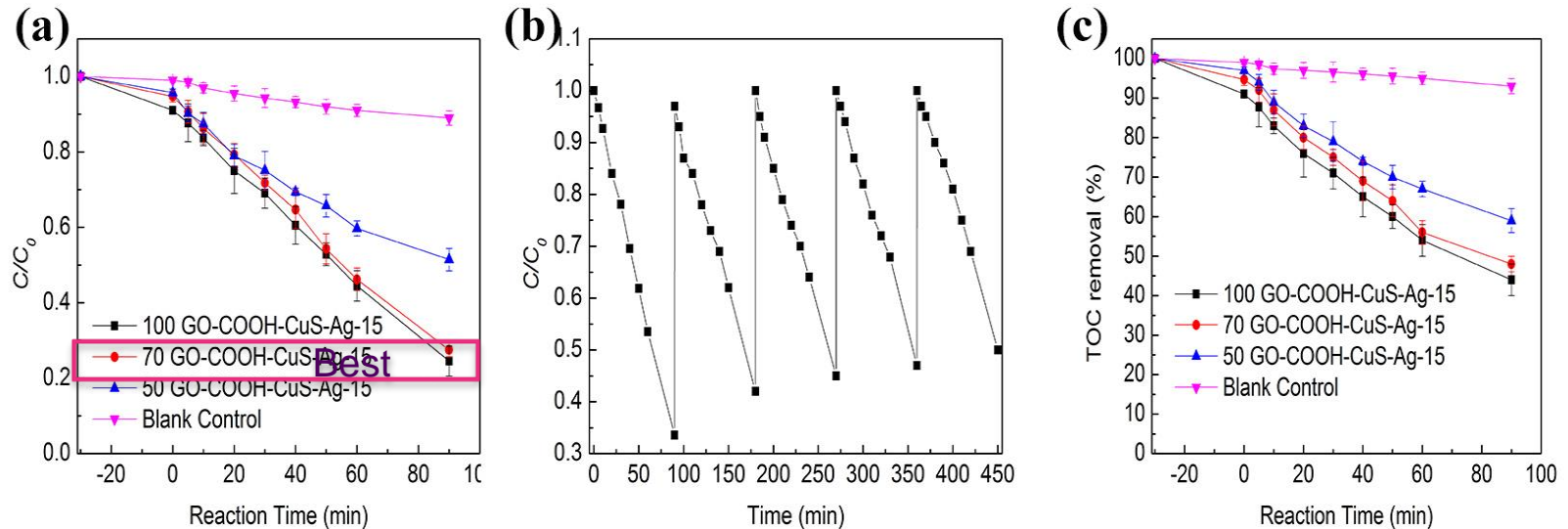


- (a) UV-vis diffuse reflectance spectra (DRS)



Photocatalytic Degradation of MB

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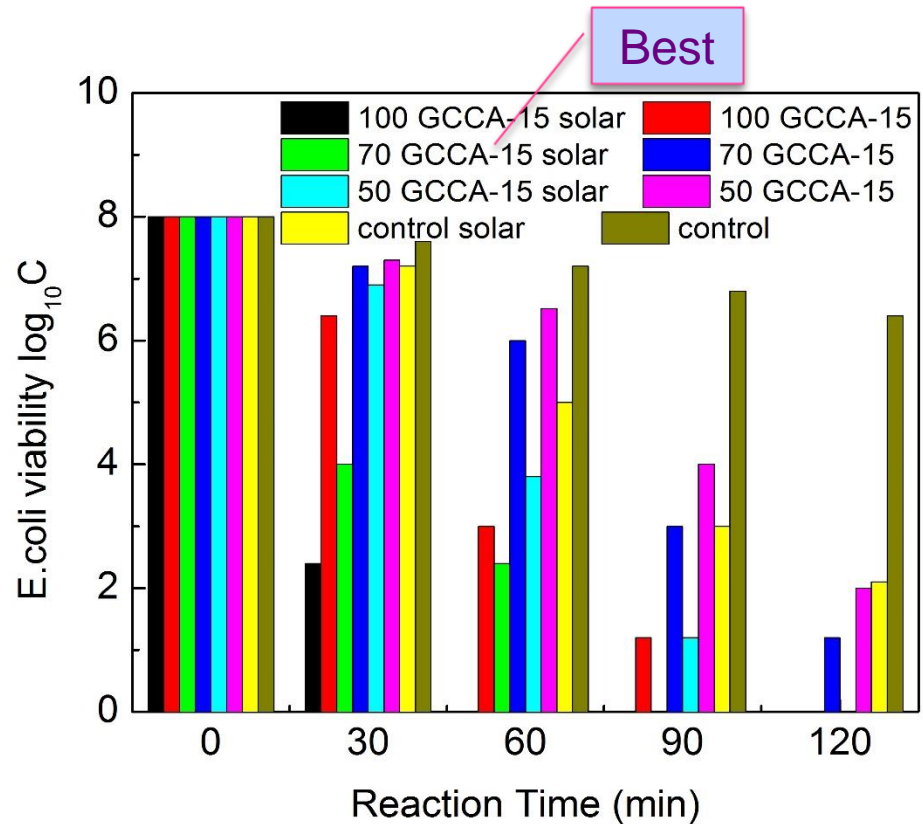
- (a) Photocatalytic performance for MB (40 ppm)



Comparison of Antibacterial capabilities

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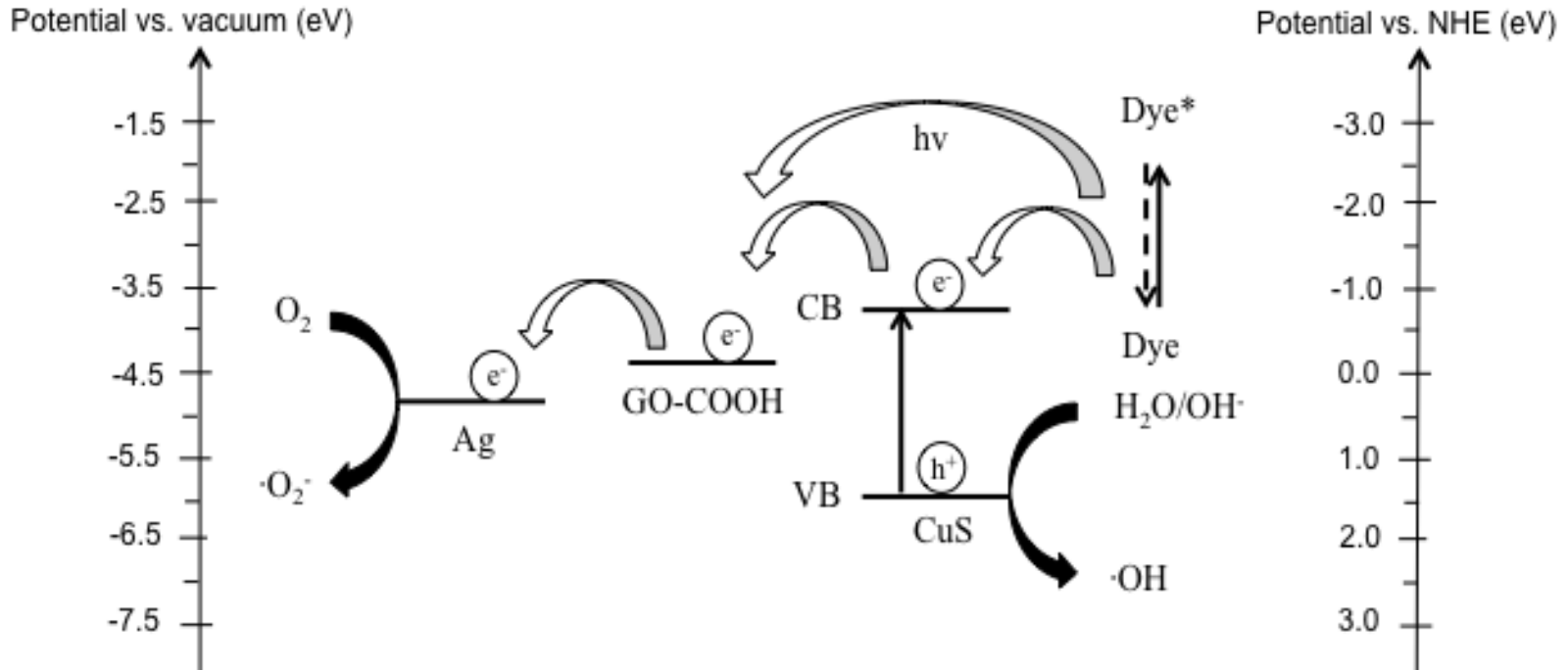
- Antibacterial activities of GO-COOH-CuS-Ag for *E. coli*





Photocatalytic Mechanism

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- Schematic illustration of proposed electron transfer mechanism of GO-COOH-CuS-Ag nanocomposites.



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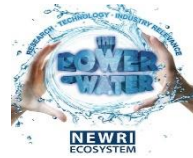
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- High performance potential in photodegradation and photo-disinfection processes of GO-COOH-CuS-Ag glass coatings

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- Advantage of simplifying the recovery and reuse in comparison to powdered forms.

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- Reducing problems including the agglomeration of powdered particles that may cause blockages.



Acknowledgements



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Paper was submitted to the Journal of Hazardous Materials

- Nanyang Technological University (M4081044)
- Ministry of Education of Singapore (M4011352)
- Nanyang Environment & Water Research Institute



