# Groundwater Denitrification Using Hydrogen Gas in a New Reactor Type

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### Short background

- Nitrate pollution (>10 mg NO3<sup>-</sup>-N/L according to EPA)
- Physicochemical methods (RO, IEX, Electrodialysis)
- Biological denitrification:
  - Nitrate reduction to N<sub>2</sub> gas by denitrifying bacteria under anoxic conditions.
  - Electron donor can be organic or inorganic and must be added to water.
  - Cell Yields are lower for autotrophic bacteria using inorganic electron donors (30% of heterotrophs).

### Hydrogenotrophic denitrification

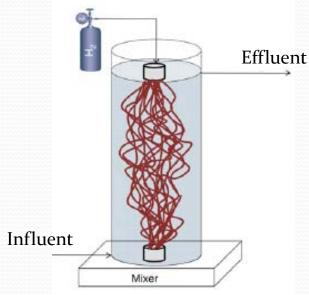
- Why hydrogen?
  - Clean: Low cell yield, no organics addition, no harmful by-products
  - Does not persist in water
  - Cheap
- So why **not**?
  - Mass transfer limitations -> low rates
  - Low hydrogen utilization-> financial asp
  - Hydrogen flammability/explosiveness -> safety

## How can hydrogen transfer be increased economically and safely?

One possible solution:

Membrane biofilm reactor (MBFR)
Lee & Rittman, 2002

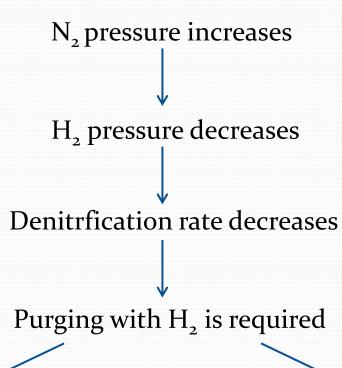
Denitrification rates  $< 1 \text{ g N} / (L_{reactor} x \text{ day})$ 



#### Our alternative solution:

A new reactor type with closed headspace!!!

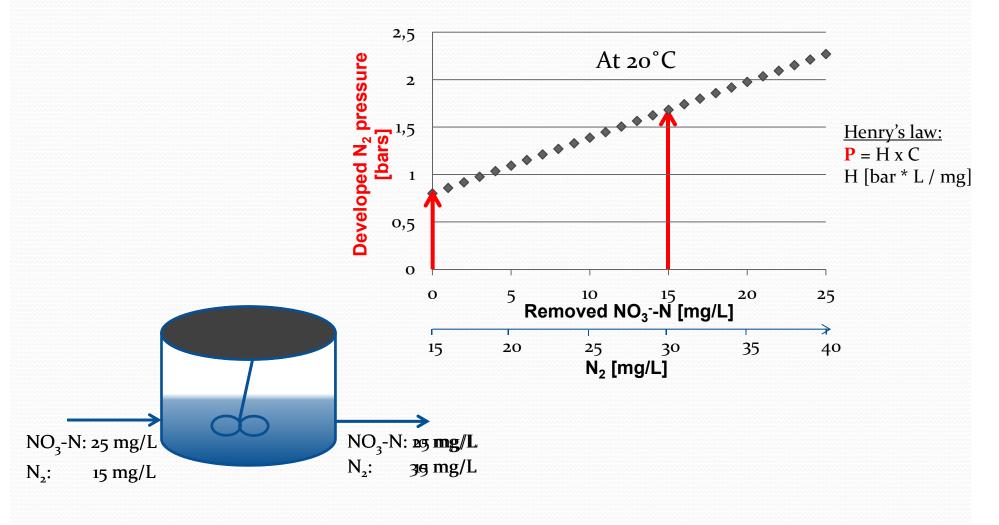
# The common misconception of closed headspace denitrifying reactor



Safety problem

Financial problem

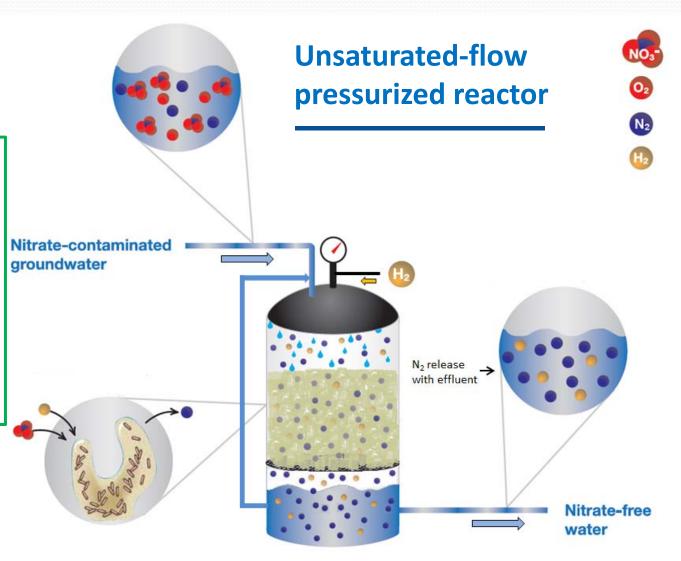
## N<sub>2</sub> accumulation in closed headspace – does it really happen???



### Our proposed system

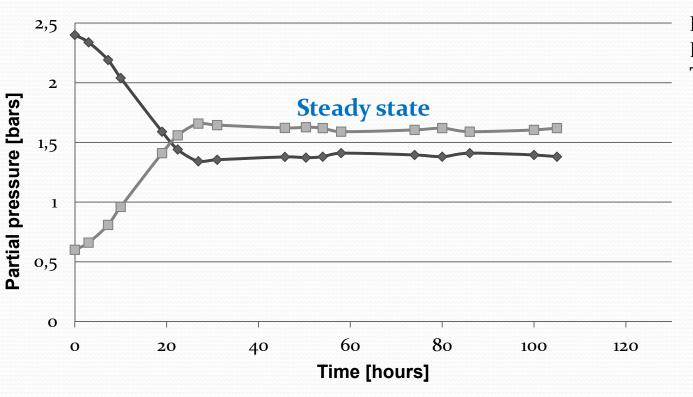
#### Main Characteristics:

- Closed Headspace (economic and safe)
- High surface area for bacterial growth
- High mass transfer
- No N<sub>2</sub> accumulation under continuous operation



### Proof of concept with GC analysis

#### Reaching gas-liquid equilibrium



Influent: 25 mg N/L Effluent: 10 mg N/L Total pressure: 3 bars

→H<sub>2</sub>

 $H_2$  utilization > 90% Max. denitrification rate ≈ 9 g  $N/(L_{reactor} \cdot d)$ 

#### Summary

- Main features of the pressurized reactor:
  - High H<sub>2</sub> utilization (>90%)
  - Safe operation
  - High denitrification rates (up to 9 gN/[L<sub>reactor</sub> x day])
- Competitive alternative to existing technologies due to simplicity and higher rates.
- Future improvements: increasing H<sub>2</sub> utilization, treatment of water with high nitrate concentration

### Thank you!