



ammonia
trapping



NH₃ recovery from digestate using gas-permeable membranes: Effect of wastewater pH

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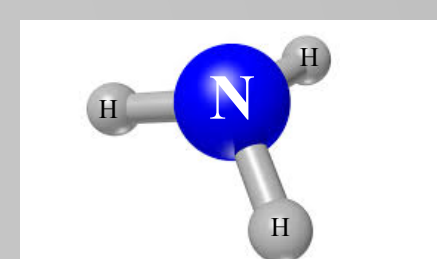
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Introduction



Figure 1: On-site pilot plant.

Agriculture is the largest source of NH₃ emissions in Europe, with a contribution of a 91% of the total NH₃ emissions in 2016. This gas has harmful effects on the environment and the human health and the current regulation on air quality states a reduction commitment for NH₃ annual emissions for each European country. **Gas-permeable membranes** have been proved as an efficient technology to capture NH₃, reducing emissions while recycling nutrients in the form of a valuable ammonium salt. The EU project **Ammonia Trapping** is aimed at reducing NH₃ emissions in livestock wastes facilities by using gas permeable membranes in *on-site* pilot plants.



- Objectives :**
1. recovering NH₃ from digestate by a pilot plant, installed at the biogas facility
 2. evaluating the effect of wastewater pH on the NH₃ recovery rate.

Material and methods

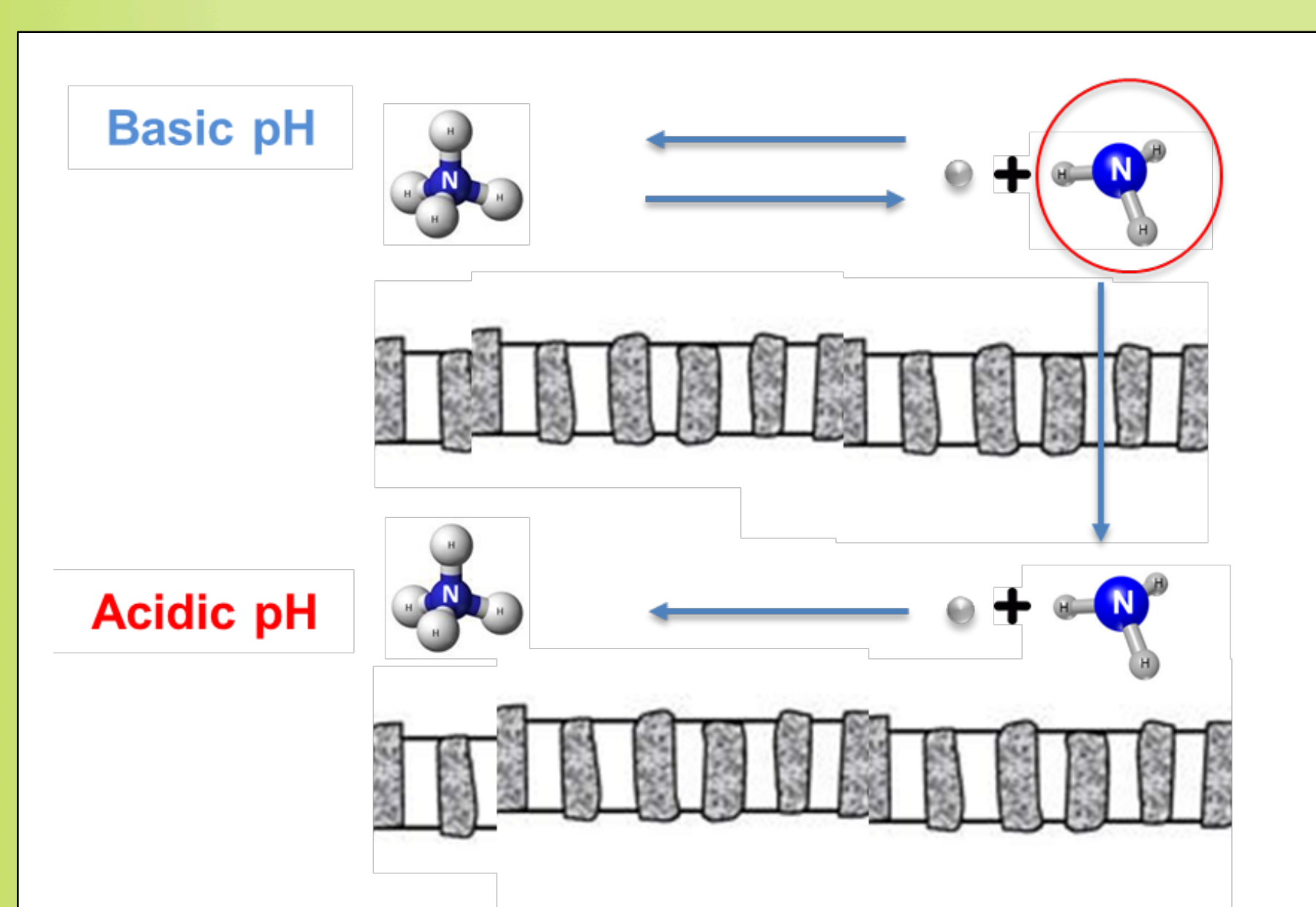


Figure 2: Principle of NH₃ capture by the gas-permeable membranes

The pilot plant was placed inside a mobile container, located in the biogas plant (Fig. 1). The pilot plant consisted of a **NH₃ separation reactor tank** of 5.85 m³ (working volume 5 m³) that contained 16 **membrane modules** in vertical configuration. The total membrane surface was 8.85 m². The membrane was tubular and made of e-PTFE with an outer diameter of 5.2 mm. A solution of 1N H₂SO₄ was contained in a 0.25 m³ **NH₃ concentration tank** (working volume 0.15 m³). This acidic solution was used as a trapping solution and it was continuously recirculated through the membrane. A PLC controlled the pilot plant. The digestate was mixed in cycles of 20 seconds on and 10 seconds off. The pH in the trapping solution was maintained below 2 by adding concentrated H₂SO₄ whenever the pH of the trapping solution increased up to 2. Aeration in on /off cycles of 180 seconds each was provided (Fig. 3).

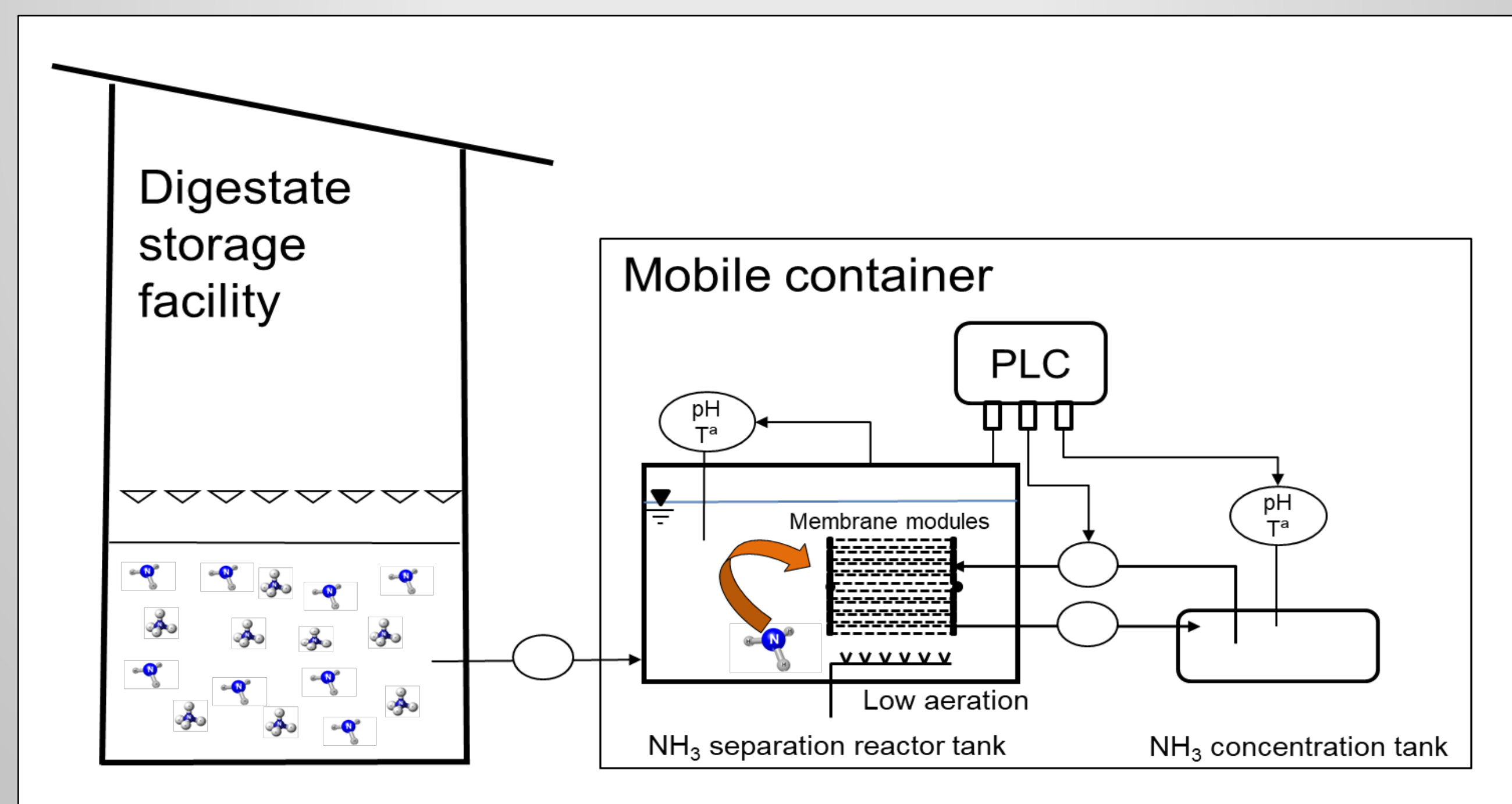


Figure 3: Scheme of the pilot plant.

Results & Discussion

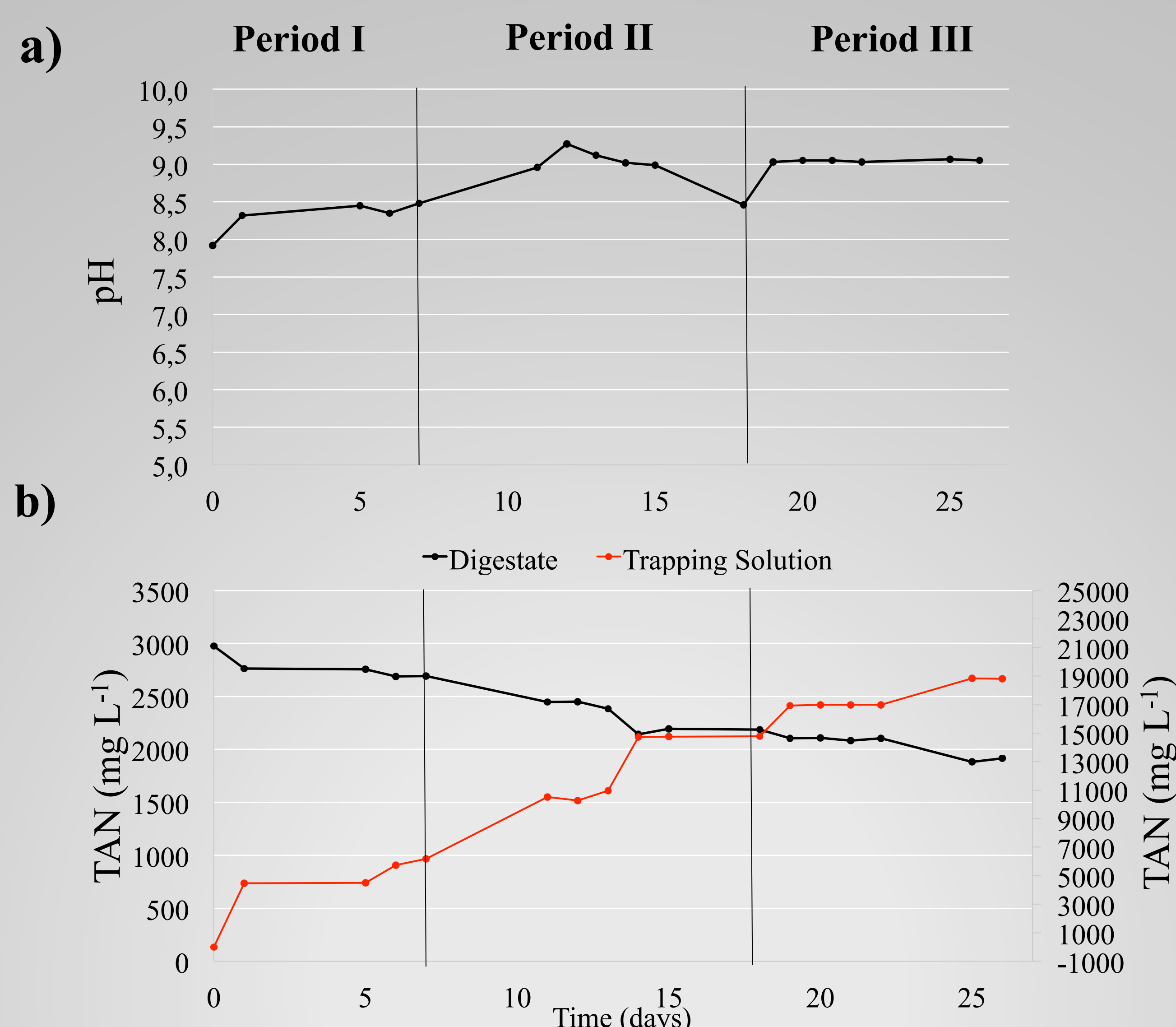


Figure 4: pH in the digestate (a) and TAN removal and recovery (b) during periods I, II and III.

Increasing pH values were tested, corresponding to pH values of 8.3 ± 0.2 (period I), 8.9 ± 0.3 (period II) and 9.0 ± 0.0 (period III) (Fig. 2a). The same digestate was used during the whole study. Initial pH in the digestate was 7.92 and it increased to 8.48 at the end of period I. The same trend was observed for period II, reaching pH values up to 9.3. However, pH was maintained in 9 during period III (Fig. 2a).

Thirty seven percent of the initial TAN in the digestate was removed during the whole experimental time (Fig. 2b). TAN removal was greater during period II, with a pH of 8.9 ± 0.3, with a 19% out of 37%. Opposite to previous assays in batch mode (García-González and Vanotti, 2015; Dube et al., 2016), TAN removal increased with time, following a linear trend. As an average, 52% of the removed TAN was recovered in the trapping solution. The trapping solution reached a final TAN concentration of 18.8 g TAN L⁻¹, meaning that TAN was approx. 6 times concentrated than in the digestate. The TAN recovery rate was around 14 g m⁻² membrane day⁻¹ for the three pH-values tested.

Conclusions

Nitrogen recovery from digestate by gas-permeable membranes was greatly affected by pH in the anaerobic digestate. Although high pH values enhance NH₃ removal, a high proportion of this NH₃ is lost to the atmosphere by stripping. In order to avoid NH₃ stripping, it is recommended to increase the ratio of membrane surface per volume unit of digestate.

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