



New bench scale plant for biosorption

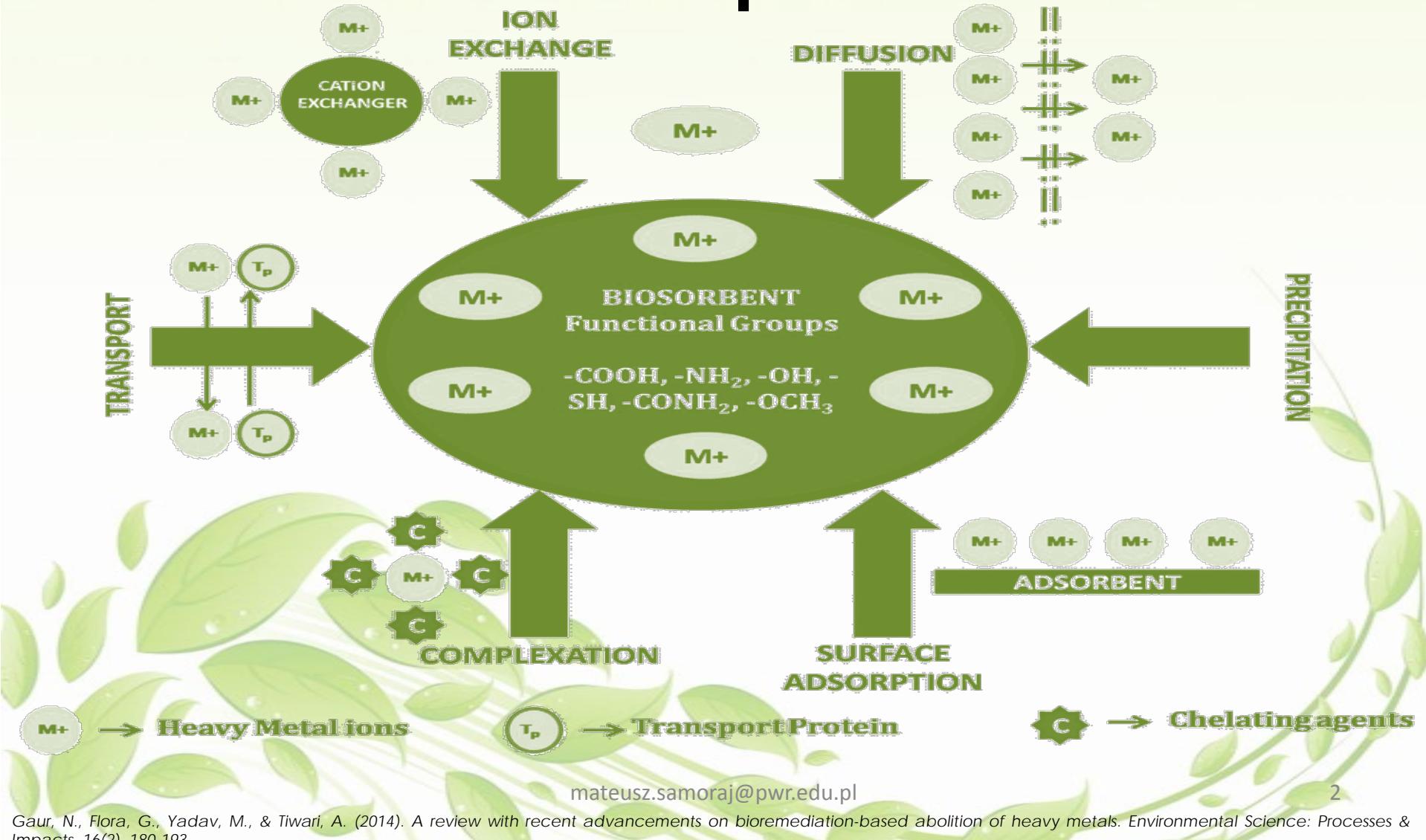
Mateusz Samoraj, Łukasz Tuhy, Katarzyna Chojnacka

Department of Advanced Material Technologies
Faculty of Chemistry, Wrocław University of Technology
Smoluchowskiego 25, 50-372 Wrocław, Poland
Tel. +48 71 320 6313

mateusz.samoraj@pwr.edu.pl



Biosorption





Biosorption – new trends

Biosorption process

Focus on:
sorbate removal

Wastewater
treatment
(11 993 papers)

Focus on:
biosorbent enrichment

Biological feed
supplements with
micronutrients
(30 papers)

Micronutrients
fertilizer
components
(29 papers)



Potential biosorbents

BIOSORBENTS



Plants

- Leaves
- Seeds



Fungi

- Micromycetes
- Macromycetes



Algae

- Microalgae
- Macroalgae



Animal

- Bones
- Eggshells



Microbes

- Bacteria
- Archea



Waste

- Manure
- Agricultural waste



Micronutrient fertilizer additives

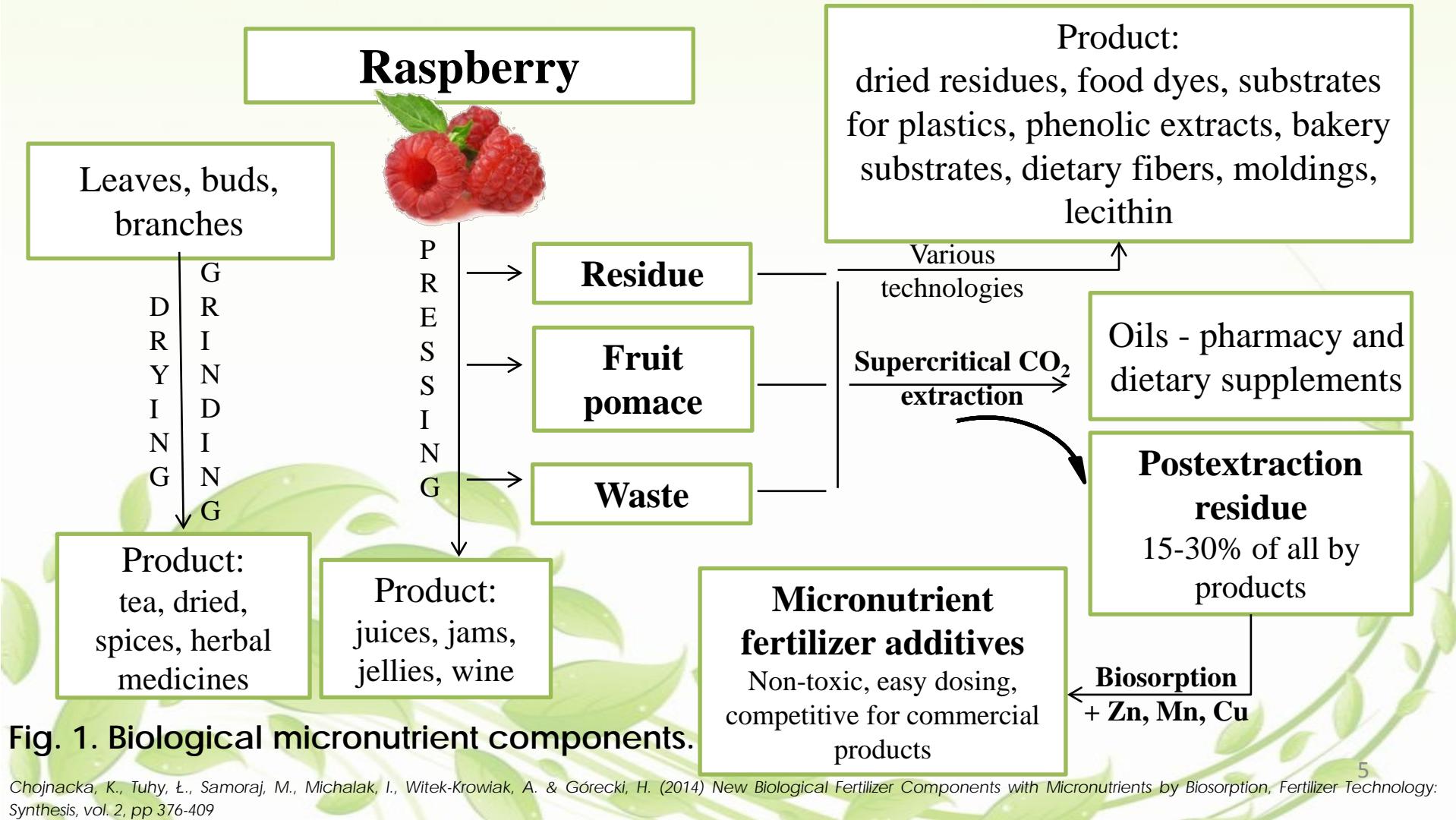
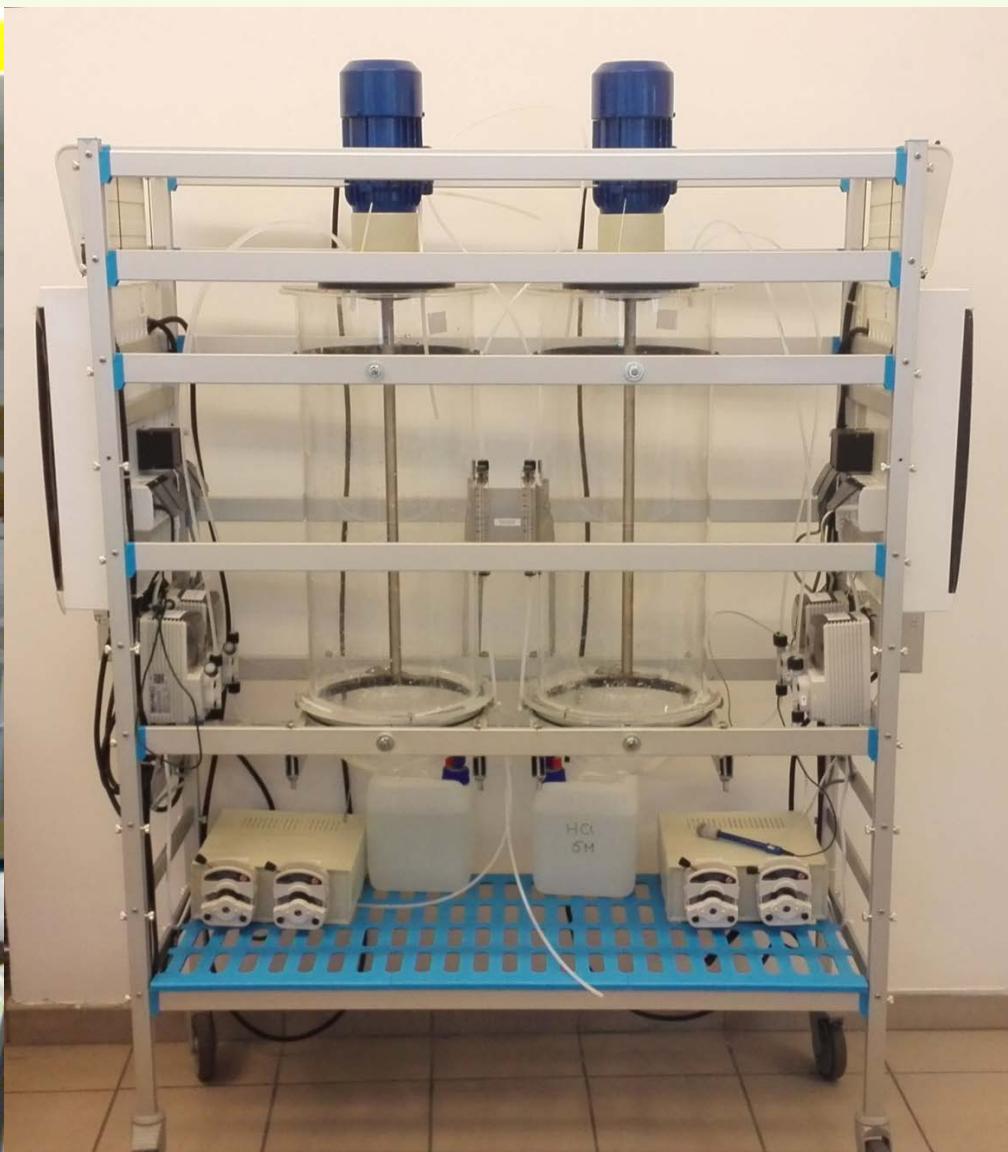


Fig. 1. Biological micronutrient components.



Installation





Installation – stirred tank mode

Fig. 3. Simplified scheme of the process in stirred tank mode – a system of two stirred tank reactors.

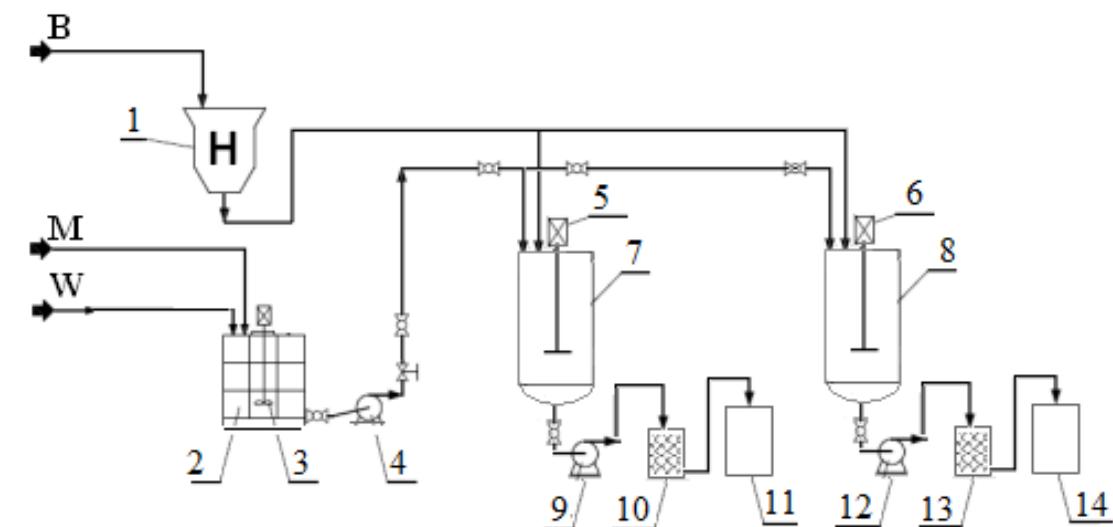
Streams:

B – Biomass,

M – Micronutrients,

W- Deionized water.

Equipment: 1 – Biomass homogenizer; 2 – Micronutrient solution tank; 3, 5, 6 – Stirrers; 4, 9, 12 – Peristaltic pumps; 7, 8 - Stirred tank reactors tank (equipped with pH regulator); 10, 13 – Biomass sedimentators; 11, 14 – Post-process tanks.





Installation – fixed bed mode

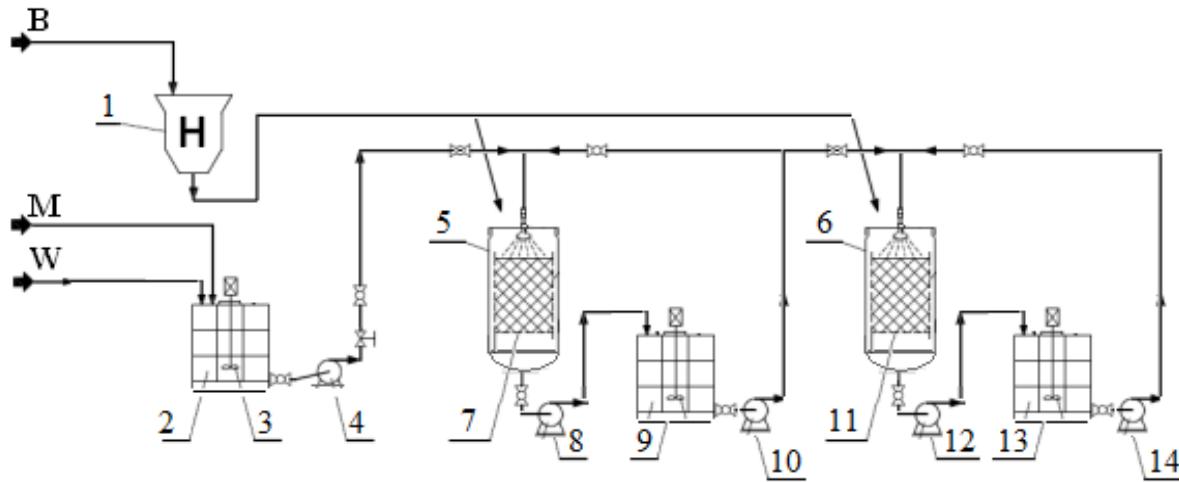


Fig. 4. Simplified scheme of the process in fixed bed mode – a system of two column reactors.

Streams: B – Biomass, M – Micronutrients, W- Deionized water.

Equipment: 1 – Biomass homogenizer; 2 – Micronutrient solution tank; 3 – Stirrer; 4, 8, 10, 12, 14 – Peristaltic pumps; 5, 6 - Reactor tanks; 7, 11 – Sieve; 9, 13 – Recirculated solution tanks (equipped with pH regulator).

mateusz.samoraj@pwr.edu.pl





Process parameters

Mode	Stirred tank	Fixed bed
Sorbate	Cu(II) ($\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$)	
Sorbate concentration		300 mg/L
Reactor volume		70L
Solution volume	50L	200L
pH		5
Temperature		25°C
Material losses		10%
Volumetric flow rate	-	1L/min
Process time	2.5h	6h
Drying		24 h, 50°C

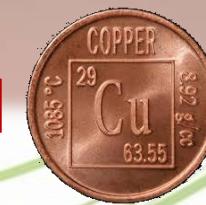


Final product

R



R + Cu(II)





Results



Daily productivity

- 300–400 g at stirred tank mode
- 10–16 kg at fixed bed mode



Material loses

- 10% in both cases



Estimated 100 kg production costs

- 419\$ at stirred tank mode
- 48.1\$ at fixed bed mode



Multielemental content of product

Element (mg/kg)	R - natural	R + Cu – stirred tank mode	R + Cu – fixed bed mode
Micronutrients	Cu	<u>8.96±1.79</u>	<u>12611±2522 EC=1407</u>
	Zn	34.6±6.9	171±34
	Mn	75.9±15.2	14±3
	Fe	122±24	184±37
	Mo	6.54±1.31	4.41±0.88
Macronutrients	P	1551±310	797±159
	K	2767±553	156±31 (-94.4%)
	S	1407±281	1370±274 (-2.63%)
	Ca	2502±500	585±117 (-76.6%)
	Mg	1802±360	170±34 (-90.6%)
	Na	472±94	630±126
Toxic elements	Cd (5)*	0.83±0.17	0.532±0.106
	Ni (60)*	<0.03	<0.03
	As (50)*	<0.3	<0.3
	Pb (140)*	<0.5	19±4
	Cr (100)*	0.117±0.023	0.616±0.123



Stirred tank mode modelling

Model	Equation	Linear form	Parameters	R ²
Pseudo-1 st order	$\frac{dq_t}{dt} = k_1(q_{eq1} - q_t)$	$\ln(q_{eq1} - q_t) = \ln(q_{eq1}) - k_1 \cdot t$	$k_1 = 0.0235 \text{ (1/min)}$ $q_{eq1} = 12.7 \text{ (mg/g)}$ $q_{ICP-OES} = 12.6 \text{ (mg/g)}$	0.997
Pseudo-2 nd order	$\frac{dq_t}{dt} = k_2(q_{eq2} - q_t)^2$	$\frac{t}{q_t} = \frac{1}{k_2 \cdot q_{eq2}^2} + \frac{1}{q_{eq2}} \cdot t$	$k_2 = 0.00426 \text{ (g/mg}\cdot\text{min)}$ $q_{eq2} = 15.3 \text{ (mg/g)}$	0.913

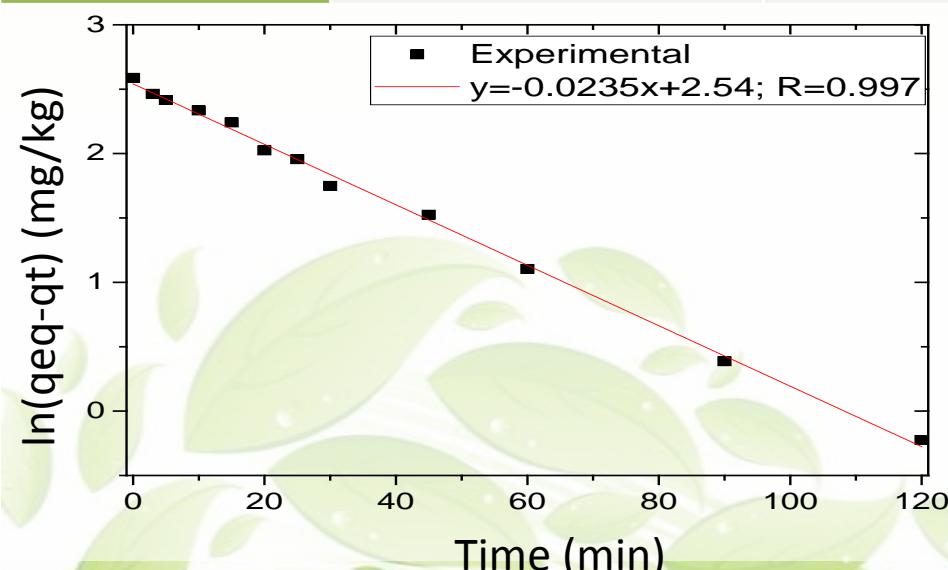


Fig.5. Pseudo-1st order model linear plot.

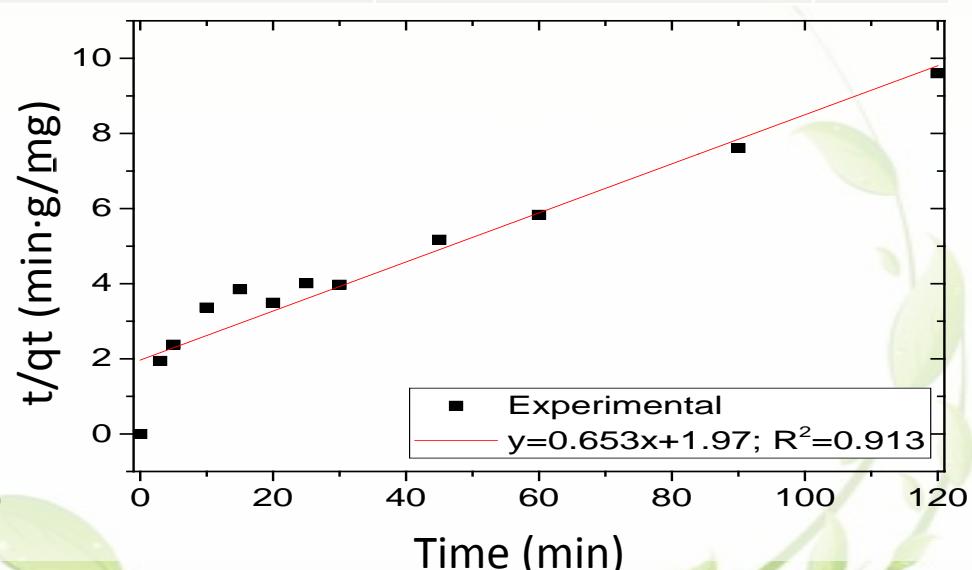


Fig. 6. Pseudo-2nd order model linear plot.

mateusz.samoraj@pwr.edu.pl

13

Fixed bed mode modelling

Model	Equation	Linear form	Parameters	R ²
Yoon-Nelson	$\frac{C_t}{C_0-C_t} = \exp(k_{YN}t - \tau k_{YN})$	$\ln \frac{C_t}{C_0-C_t} = k_{YN}t - \tau k_{YN}$	$\tau=105$ (min) $k_{YN}= 0.0078$ (1/min)	0.996
Thomas	$\frac{C_t}{C_0} = \frac{1}{1+\exp\left[\left(\frac{k_{Th}q_{eq}}{Q}\right) - k_{Th}C_0 t\right]}$	$\ln\left(\frac{C_0}{C_t} - 1\right) = \frac{k_{Th}q_{eq}x}{Q} - k_{Th}C_0 t$	$k_{Th}= 0.0260$ (mL/min mg) $q_{eq}= 13.5$ (mg/g) $q_{ICP-OES}=12.8$ (mg/g)	0.996
Walborska	$\partial \frac{\partial c_b}{\partial t} + v \frac{\partial c_b}{\partial H} + \frac{\partial q}{\partial t} = D_{ax} \frac{\partial^2 c_b}{\partial H^2}$	$\ln\left(\frac{c}{c_0}\right) = \frac{\beta_a c_0 t}{q} - \frac{\beta_a H}{v}$	$\beta_a= 0.181$ (1/min) $q= 17.5$ (mg/g)	0.935
Adams-Bohart	$\frac{C_t}{C_0} = \exp\left(k_{AB}C_0 t - k_{AB}N_0 \frac{z}{U_0}\right)$	$\ln\left(\frac{C_t}{C_0}\right) = k_{AB}C_0 - k_{AB}N_0 \frac{z}{U_0}$	$N_0 = 17500$ (mg/L) $k_{AB} = 0.0103$ (mL/min mg)	0.935

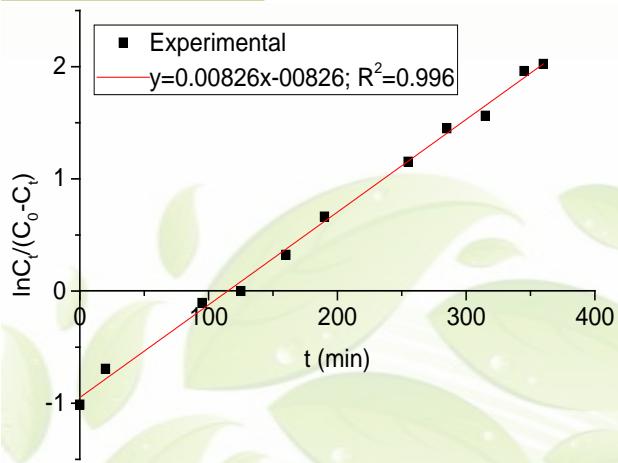


Fig. 7. Yoon-Nelson model linear plot vs. experimental data.

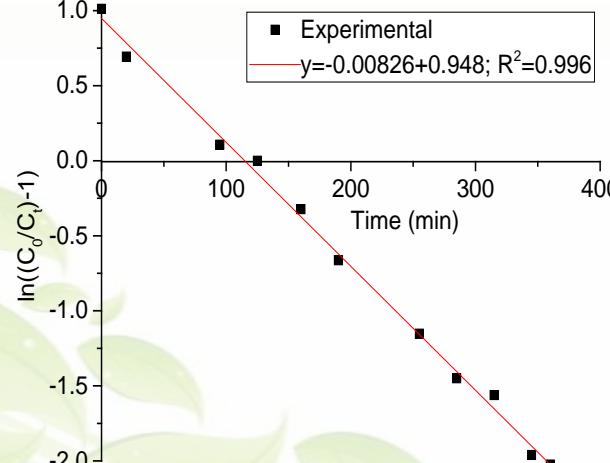


Fig. 8. Thomas model linear plot vs. experimental data.

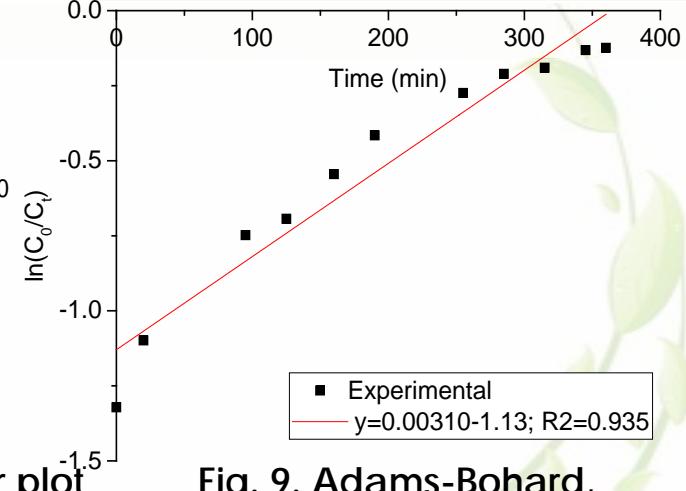


Fig. 9. Adams-Bohard, Walborska model linear plot.



Conclusions



New bench scale installation for biosorption and comparison of two possible process mode was presented with kinetics description



New eco-friendly micronutrient fertilizer additives with copper based on raspberry seeds were produced and multielemental content was determined after sample mineralization



The material losses in the process were about 10% for fixed bed and stirred tank reactors



The cost of the utilization of berries seeds to micronutrient fertilizer components in fixed bed mode was much lower (48.1 \$/100 kg) than in stirred tank mode (419 \$/100 kg)



New installation is an efficient tool in biosorption studies and enables investigation of parameter influence in a wide range



Perspectives



Selection of the process parameters
(and minor modifications of installation)



Vegetation tests of new product
(TF determination and comparison with commercial products)



Scale up to pilot plant scale
(transfer proven solutions)



The work was supported by Polish National Science Centre,
project no. 2012/05/E/ST8/03055
entitled: *Biosorption of metal ions to the biomass of seeds of berries*



NATIONAL SCIENCE CENTRE
POLAND



Thank you for your attention!

MSc. Mateusz Samoraj

Department of Advanced Material Technologies,
Wrocław University of Technology
Smoluchowskiego 25,
50-372 Wrocław, Poland

mail: mateusz.samoraj@pwr.edu.pl

Tel. +48 71 320 6313